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## SYNTHESIS OF NEW SERIES OF THIENYL ACRYLATE DERIVATIVES VIA BAYLIS-HILLMAN REACTION AND EVALUATION OF ANTIMICROBIAL ACTIVITY

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Thiophene,  
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### ABSTRACT

A series of new Thienyl acrylate compounds were synthesized via Baylis-Hillman reaction and tested to demonstrate *in vitro* antimicrobial activity. Some of these compounds exhibited a very good activity against Gram positive strains. The structure of these newly synthesized compounds were conformed by <sup>1</sup>H NMR, IR, Mass and Elemental analysis, some of them were also confirmed by <sup>13</sup>C NMR.

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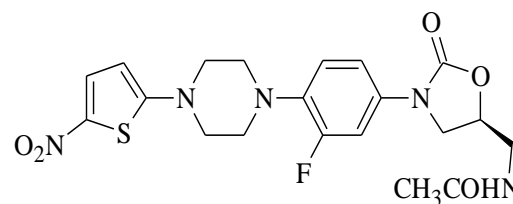
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**INTRODUCTION:** A dramatic increase in microorganisms resistant to multiple antimicrobial agents is a serious problem worldwide, especially the gram positive bacteria which triggered a clear need for the discovery of new antibacterials rather than analogs of the existing ones<sup>1-3</sup>. Traditionally, small molecules have been a reliable source for discovering novel biologically active compounds. Although a lot of work has been done on the heterocycles, they remained an active area of research<sup>4-6</sup>. In this context, thiophene nucleus represents a very important field in drug discovery, which is present in many natural and synthetic products with a wide range of pharmacological activities<sup>7-11</sup>. Studies on thiophene like compounds have served as a feasible field of research in the perusal of biologically active compounds (**Fig. 1**)<sup>12-15</sup>.

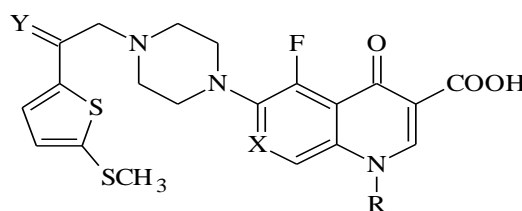
Consequently, there continues to be interest in developing new biologically active compounds. In our continuous endeavor to develop new and potent antibacterial molecules, we decided to study thiophene as our basic structure and investigate the activity by introducing substituent like quinoline, which is known to possess pharmacological activity<sup>16-23</sup>. Quinolines and their derivatives occur in numerous natural products. Many quinolines display interesting physiological activities and have found attractive applications as pharmaceuticals and agrochemicals as well as being general synthetic building blocks<sup>24-25</sup>.

In particular hydroxyquinoline at 8<sup>th</sup> position well known as oxyquinoline has wide variety of uses and its medicinal and agricultural significances were discovered before the start of current century. 8-hydroxy quinoline and its derivatives have been found to show diverse biological activities such as antibacterial, antimalarial, antipneumococcic, fungicidal, antiseptic, antituberculosic, antineoplastic, amebacidal, pesticidal, antihelmintic and ant diuretic activities like Chlorquinaldol,

Clioquinol and nitroxoline. Its derivatives benzoquine, 8-hydroxy quinoline citrate and 8-hydroxy quinoline sulfate are employed as antiseptic and disinfectants.



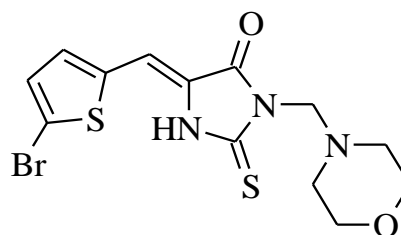
RBx 8700

**A**

X = CH, N

Y = O, NOH, NOME, NOBn

R = cyclopropyl, ethyl

**B****C****FIG. 1: STRUCTURE OF KNOWN COMPOUNDS CONTAINING THIOPHENE**

We have prepared a library of thienyl acrylate compounds via Baylis-Hillman reaction, which have been evaluated for their *in vitro* antibacterial and anti fungal activity. The Baylis-Hillman reaction<sup>26-32</sup> has attracted the attention of organic chemists for preparing synthetically useful multifunctional molecules which have been successfully employed in various syntheses.

**MATERIALS AND METHODS:** All the chemicals and reagents were procured from Sigma Aldrich lab grade source. All the solvents used were from commercial sources and redistilled before use. All melting points were determined on a Buchi apparatus and are uncorrected. The I.R spectra (in KBr pellets) were recorded on a JASCO spectrometer and frequencies are expressed in  $\text{cm}^{-1}$ . Mass spectra (CG/MS) were recorded on an Agilent MSD VL mass spectrometer.  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra were recorded on a Bruker Advance 400 spectrometer operating at 400 MHz. The chemical shifts are reported in ppm ( $\delta$ ) relative to TMS. Proton and carbon spectra were typically obtained at room temperature. The purity of the compounds was checked by TLC on silica gel plates using ethyl acetate: hexane or Methylene dichloride: methanol as eluent and spots were developed in ultraviolet.

#### General Synthetic procedure for synthesis of (4a-b):

**Methyl (2E)-2-(bromomethyl)-3-(2-thienyl) acrylate (4a):** Thiophene-2-carboxaldehyde **1a** (0.04 mole), methyl acrylate **2** (0.14 mole), 1, 4-diazabicyclo [2.2.2] octane (DABCO) (0.04 mole) were stirred at room temperature for 72 h and the reaction was monitored by TLC. Ethyl acetate was used to dilute the reaction mixture after the completion of the reaction and washed successively with 2N Hydrochloric acid, aqueous sodium bicarbonate solution and water. The organic layer was dried over anhydrous sodium sulphate and the solvent was evaporated to get crude hydroxy compound **3a**.

To the crude **3a**, 47 % Hydrobromic acid (0.285 mole) and concentrated sulphuric acid (0.13 mole) were added and stirred in methylene dichloride at 0-10°C for 3 h. The reaction mixture was extracted in methylene dichloride and washed with sodium bicarbonate solution and water. It was then dried, solvent evaporated to get the residue, to obtain a yellow solid which was recrystallized from

n-hexane. Yield- 9 g (78 %); m.pt. 52°C; I.R (KBr pellets  $\text{cm}^{-1}$ )  $\nu$  1710, 1602, 1415, 1214, 1201, 723;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz):  $\delta$  8.03 (s, 1H),  $\delta$  7.64 (d, J = 5 Hz, 1H),  $\delta$  7.47 (d, J = 3.6 Hz, 1H),  $\delta$  7.18 (dd, J = 3.96, 5.08 Hz, 1H),  $\delta$  4.59 (s, 2H),  $\delta$  3.87 (s, 3H);  $^{13}\text{C}$  NMR (300 MHz,  $\text{DMSO}-d_6$ ):  $\delta$  166.02, 136.11, 135.78, 135.28, 133.96, 128.27, 123.57, 52.43, 27.91.

**Methyl (2E)- 2- (bromomethyl)- 3- (5- bromo- 2-thienyl) acrylate (4b):** Yield- 6.5 g (73 %); m.pt. 80°C; I.R (KBr pellets  $\text{cm}^{-1}$ )  $\nu$  1708, 1606, 1411, 1253, 1207, 771;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz):  $\delta$  7.79 (s, 1H),  $\delta$  7.26 (s, 1H),  $\delta$  7.21 (d, J = 5.72 Hz, 1H),  $\delta$  4.50 (s, 2H),  $\delta$  3.86 (s, 3H);  $^{13}\text{C}$  NMR (300 MHz,  $\text{DMSO}-d_6$ ):  $\delta$  166.15, 135.88, 135.38, 134.26, 127.27, 122.63, 52.51, 27.86.

#### General Synthetic procedure for synthesis of 6-18:

**Methyl (2E) - 2- {[2-methylquinolin- 8- yl) oxy] methyl}- 3- (2- thienyl) acrylate(6):** Compound **4a** (0.006 mole) was treated with 8-hydroxy-2-methyl quinoline (0.006 mole) in the presence of  $\text{K}_2\text{CO}_3$  in dry dimethyl formamide for 1 h at room temperature. Water was added to the reaction mixture which precipitates out the solid was filtered and purified in methanol to get pure compound as yellow solid. Yield- 1.23g (82 %); m.pt. 165°C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1710, 1617, 1209, 1101, 983, 740 ppm.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ , 400 MHz):  $\delta$  8.23 (s, 2H),  $\delta$  8.21 (d, J = 8.4 Hz, 1H),  $\delta$  7.83 (d, J = 4.96 Hz, 1H),  $\delta$  7.70 (d, J = 3.44 Hz, 2H),  $\delta$  7.52-  $\delta$  7.45 (m, 1H),  $\delta$  7.42 -  $\delta$  7.34 (m, 1H),  $\delta$  7.19 (t, 1H),  $\delta$  5.10 (s, 2H),  $\delta$  3.80 (s, 3H),  $\delta$  2.59 (s, 3H).  $^{13}\text{C}$  NMR (300 MHz,  $\text{DMSO}-d_6 + \text{TFA}-d$ ):  $\delta$  167.31, 148.68, 146.25, 139.01, 136.81, 135.72, 133.47, 129.89, 128.51, 125.36, 121.55, 121.18, 117.36, 114.77, 113.55, 109.74, 65.18, 52.50, 21.03. Mass (m/z) = 340 ( $\text{M}^+$ ); Anal. Calcd for  $\text{C}_{19}\text{H}_{17}\text{NO}_3\text{S}$ : C, 67.24; H, 5.05; N, 4.13. Found: C, 67.20; H, 5.04; N, 4.11.

**Methyl (2E)- 3- (5- bromo- 2- thienyl)- 2- {[2-methylquinolin- 8- yl) oxy] methyl} acrylate (7):**

Yield- 1.2 g (82 %). m.pt. 176°C; I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1691, 1614, 1261, 1209, 1099, 831, 752 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.22 (d, J = 8.4 Hz, 1H),  $\delta$  8.15 (s, 1H),  $\delta$  7.57 (d, J = 3.88 Hz, 1H),  $\delta$  7.54 (d, J = 8.0 Hz, 1H),  $\delta$  7.49 (d, J = 7.52 Hz, 1H),  $\delta$  7.45 (t, 1H),  $\delta$  7.35 (t, 2H),  $\delta$  5.10 (s, 2H),  $\delta$  3.79 (s, 3H),  $\delta$  2.60 (s, 3H).  $^{13}\text{C}$  NMR (300 MHz, DMSO- $d_6$  + TFA- $d$ ):  $\delta$  166.69, 148.54, 145.34, 138.12, 137.70, 136.14, 131.63, 129.39, 128.11, 125.07, 121.76, 120.89, 118.93, 117.18, 114.22, 113.35, 64.87, 52.41, 21.03. Mass (m/z) = 420 ( $\text{M}^{2+}$ ). Anal. Calcd for  $\text{C}_{19}\text{H}_{16}\text{BrNO}_3\text{S}$ : C, 54.55; H, 3.86; N, 3.35. Found: C, 54.53; H, 3.85; N, 3.33.

**Methyl (2E) - 3- (5- bromo- 2- thienyl)- 2- [(quinolin- 8- yloxy) methyl] acrylate (8):**

Yield-1.05 g (80 %). m.p. 161 °C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1693, 1616, 1261, 1209, 1101, 757 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.80 (t, 1H),  $\delta$  8.36 (d, J = 8.2 Hz, 1H),  $\delta$  8.16 (s, 1H),  $\delta$  7.61-  $\delta$  7.55 (m, 3H),  $\delta$  7.42 (t, 2H),  $\delta$  7.31 (d, J = 3.88 Hz, 1H),  $\delta$  5.10 (s, 2H),  $\delta$  3.79 (s, 3H).  $^{13}\text{C}$  NMR (300 MHz, DMSO- $d_6$  + TFA- $d$ ):  $\delta$  166.70, 149.06, 146.01, 145.54, 138.15, 137.69, 136.16, 131.68, 130.18, 129.98, 129.84, 122.99, 121.83, 121.10, 120.89, 119.02, 117.27, 113.83, 113.44, 109.62, 64.49, 52.45. Mass (m/z) = 406 ( $\text{M}^{2+}$ ). Anal. Calcd for  $\text{C}_{18}\text{H}_{14}\text{BrNO}_3\text{S}$ : C, 53.48; H, 3.49; N, 3.46. Found: C, 53.45; H, 3.47; N, 3.45.

**Methyl (2E) - 2- [(quinolin- 8- yloxy) methyl] - 3- (2-thienyl) acrylate (9):**

Yield- 1.0 g (78%). m.pt. 160°C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1714, 1617, 1207, 1099, 981, 738 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.79 (d, J = 3.96 Hz, 1H),  $\delta$  8.33 (d, J = 8.28 Hz, 1H),  $\delta$  8.25 (s, 1H),  $\delta$  7.81 (d, J = 4.88 Hz, 1H),  $\delta$  7.66 (d, J = 3.4 Hz, 1H),  $\delta$  7.56-  $\delta$  7.52 (m, 3H),  $\delta$  7.41 (t, 1H),  $\delta$  7.18 (t, 1H),  $\delta$  5.15 (s, 2H),  $\delta$  3.79 (s, 3H).  $^{13}\text{C}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  167.0, 154.10, 148.90, 139.62, 137.72, 136.44, 135.69, 135.05, 133.20, 128.99, 127.93, 126.75, 122.30, 121.81, 119.92, 109.44, 63.09, 52.18. Mass (m/z) = 326 ( $\text{M}^+$ ). Anal.

Calcd for  $\text{C}_{18}\text{H}_{15}\text{NO}_3\text{S}$ : C, 66.44; H, 4.65; N, 4.30. Found: C, 66.45; H, 4.65; N, 4.29.

**Ethyl (2E) - 2- [(quinolin- 8- yl) oxy] methyl] - 3- (2-thienyl) acrylate (10):**

Yield- 1.0 g (81 %). m.pt. 148.5°C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1683, 1613, 1280, 1212, 1098, 795, 713 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.93 (t, 1H),  $\delta$  8.21 (s, 1H),  $\delta$  8.15 (d, J = 7.92 Hz, 1H),  $\delta$  7.52-  $\delta$  7.39 (m, 5H),  $\delta$  7.26 (t, 1H),  $\delta$  7.04 (t, 1H),  $\delta$  5.26 (s, 2H),  $\delta$  4.31 (t, 2H),  $\delta$  1.33 (t, 3H). Mass (m/z) = 340 ( $\text{M}^+$ ). Anal. Calcd for  $\text{C}_{19}\text{H}_{17}\text{NO}_3\text{S}$ : C, 67.24; H, 5.05; N, 4.13. Found: C, 67.22; H, 5.03; N, 4.12.

**Propyl (2E) - 2- [(quinolin- 8- yloxy) methyl] - 3- (2-thienyl) acrylate (11):**

Yield- 1.1 g (78 %). m.pt. 121°C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1682, 1615, 1465, 1282, 1215, 1100, 986, 943, 821, 711 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.93 (t, 1H),  $\delta$  8.21 (s, 1H),  $\delta$  8.14 (d, J = 8.24 Hz, 1H),  $\delta$  7.52-  $\delta$  7.38 (m, 5H),  $\delta$  7.26 (t, 1H),  $\delta$  7.04 (t, 1H),  $\delta$  5.26 (s, 2H),  $\delta$  4.19 (t, 2H),  $\delta$  1.73 (s, 3H),  $\delta$  0.97 (t, 3H). Mass (m/z) = 354.0 ( $\text{M}^+$ ). Anal. Calcd for  $\text{C}_{20}\text{H}_{19}\text{NO}_3\text{S}$ : C, 67.97; H, 5.42; N, 3.96. Found: C, 67.96; H, 5.41; N, 3.95.

**Cyclopentyl (2E) - 2- [(quinolin- 8- yl) oxy] methyl] - 3- (2- thienyl) acrylate (12):**

Yield- 1.2 g (76 %). m.pt. 116°C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1682, 1616, 1465, 1283, 1215, 1100, 986, 944, 821, 711ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.93 (t, 1H),  $\delta$  8.14 (s, 1H),  $\delta$  8.12 (s, J = 1.56 Hz, 1H),  $\delta$  7.51-  $\delta$  7.38 (m, 5H),  $\delta$  7.26 (t, 1H),  $\delta$  7.04 (t, 1H),  $\delta$  5.30 (q, 1H),  $\delta$  5.25 (s, 2H),  $\delta$  1.89 (t, 2H),  $\delta$  1.76 (m, 2H),  $\delta$  1.66 (s, 2H),  $\delta$  1.60 (t, 2H). Mass (m/z) = 380 ( $\text{M}^+$ ). Anal. Calcd for  $\text{C}_{22}\text{H}_{21}\text{NO}_3\text{S}$ : C, 69.63; H, 5.58; N, 3.69. Found: C, 69.60; H, 5.56; N, 3.68.

**Benzyl (2E) - 2- [(quinolin- 8- yloxy) methyl] - 3- (2-thienyl) acrylate (13):**

Yield- 1.25 g (84 %). m.pt. 99.5°C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1694, 1617, 1206, 1099, 982, 741, 693 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.93 (t, 1H),  $\delta$  8.24 (s, 1H),  $\delta$  8.15 (d, J = 8.2 Hz, 1H),  $\delta$  7.49-  $\delta$  7.21 (m, 11H),  $\delta$  7.04 (t, 1H),  $\delta$

5.29 (s, 4H). Mass (m/z) = 402 ( $M^+$ ). Anal. Calcd for  $C_{24}H_{19}NO_3S$ : C, 71.80; H, 4.77; N, 3.49. Found: C, 71.78; H, 4.76; N, 3.49.

**4-Trifluoromethylbenzyl (2E) - 2- [(quinolin- 8-yl) oxy] methyl} - 3-(2-thienyl) acrylate (14):** Yield- 1.05 g (70 %). m.pt. 117°C. I.R (KBr pellets  $cm^{-1}$ ):  $\nu$  1689, 1620, 1329, 1212, 1102, 1067, 984, 694 ppm.  $^1H$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.93 (t, 1H),  $\delta$  8.25 (s, 1H),  $\delta$  8.15 (d, J = 1.68 Hz, 1H),  $\delta$  7.61-  $\delta$  7.40 (m, 9H),  $\delta$  7.26 (t, 1H),  $\delta$  7.06 (t, 1H),  $\delta$  5.30 (s, 4H). Mass (m/z) = 470 ( $M^+$ ). Anal. Calcd for  $C_{25}H_{18}F_3NO_3S$ : C, 63.96; H, 3.86; N, 2.98. Found: C, 63.95; H, 3.85; N, 2.97.

**Methyl (2E) - 2- {[5, 7- dichloroquinolin- 8-yl) oxy] methyl} - 3- (2- thienyl) acrylate (15):** Yield- 1.1 g (78 %). m.pt. 137°C. I.R (KBr pellets  $cm^{-1}$ ):  $\nu$  3430, 2944, 1715, 1619, 1502, 1308, 1206, 1094, 989, 778, 738 ppm.  $^1H$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  9.04 (d, J = 0.8 Hz, 1H),  $\delta$  9.02 (t, 1H),  $\delta$  8.62 (d, J = 8.8 Hz, 1H),  $\delta$  8.32 (s, 1H),  $\delta$  7.85-  $\delta$  7.81 (m, 2H),  $\delta$  7.68 (d, J = 3.4 Hz, 1H),  $\delta$  7.62 (d, J = 8.88 Hz, 1H),  $\delta$  7.20 (t, 1H),  $\delta$  5.30 (s, 2H),  $\delta$  3.81 (s, 3H).  $^{13}C$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  166.91, 159.63, 150.28, 138.68, 138.63, 137.58, 136.28, 135.77, 133.72, 131.78, 128.14, 127.71, 124.86, 122.10, 121.11, 107.52, 64.36, 52.40. Mass (m/z) = 394 (M). Anal. Calcd for  $C_{18}H_{13}Cl_2NO_3S$ : C, 54.83; H, 3.32; N, 3.55. Found: C, 54.84; H, 3.30; N, 3.55.

**Methyl (2E) - 2- {[5-chloro- 7- iodoquinolin- 8-yl) oxy] methyl} - 3- (2- thienyl) acrylate (16):** Yield- 0.9 g (72%), m.pt. 124.5°C. I.R (KBr pellets  $cm^{-1}$ ):  $\nu$  3381, 2944, 1703, 1617, 1570, 1330, 1219, 1084, 938, 780, 710 ppm.  $^1H$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  9.05 (d, J = 2.76 Hz, 1H),  $\delta$  8.56 (d, J = 8.56 Hz, 1H),  $\delta$  8.09 (s, 1H),  $\delta$  8.0 (s, 1H),  $\delta$  7.86 (d, J = 4.76 Hz, 1H),  $\delta$  7.81-  $\delta$  7.76 (m, 2H),  $\delta$  7.22 (t, 1H),  $\delta$  5.59 (s, 2H),  $\delta$  3.66 (s, 3H).  $^{13}C$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  167.09, 155.26, 150.47, 141.22, 136.42, 136.17, 134.86, 134.59, 133.06, 132.15, 128.07, 127.79, 124.85, 124.19, 123.01, 91.13, 69.28, 52.07 Mass (m/z) =

486 ( $M^+$ ). Anal. Calcd for  $C_{18}H_{13}ClINO_3S$ : C, 44.51, H, 2.70; N, 2.88. Found: C, 44.49, H, 2.68; N, 2.86.

**Methyl (2E) - 2- {[5, 7- dichloro- 2- methylquinolin- 8- yl) oxy] methyl} - 3- (2-thienyl) acrylate (17):** Yield- 1.0 g (80 %). m.pt. 147°C. I.R (KBr pellets  $cm^{-1}$ ):  $\nu$  3387, 2949, 1705, 1616, 1493, 1314, 1251, 1089, 945, 792, 709 ppm.  $^1H$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.42 (s, 1H),  $\delta$  7.97-  $\delta$  7.83 (m, 4H),  $\delta$  7.64 (s, 1H),  $\delta$  7.22 (s, 1H),  $\delta$  5.55 (s, 2H),  $\delta$  3.63 (s, 3H),  $\delta$  2.74 (s, 3H).  $^{13}C$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  167.14, 159.95, 149.91, 142.48, 136.44, 136.28, 134.56, 134.17, 133.05, 132.23, 128.29, 126.37, 125.60, 123.76, 119.33, 97.97, 68.91, 52.29, 25.03. Mass (m/z) = 408 (M). Anal. Calcd for  $C_{19}H_{15}Cl_2NO_3S$ : C, 55.89; H, 3.70; N, 3.43. Found: C, 55.86; H, 3.69; N, 3.42.

**Methyl (2E) - 2 - {[5- nitroquinolin - 8- yl) oxy] methyl} - 3 - (2- thienyl) acrylate (18):** Yield- 1.1 g (70 %), m.pt. 154°C. I.R.: (KBr pellets  $cm^{-1}$ ):  $\nu$  3435, 1705, 1611, 1433, 1283, 1117, 942, 710 ppm.  $^1H$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  9.09 (t, 1H),  $\delta$  8.58 (t, 1H),  $\delta$  7.99 (s, 1H),  $\delta$  7.91 (s, 1H),  $\delta$  7.85-  $\delta$  7.82 (m, 2H),  $\delta$  7.79-  $\delta$  7.75 (m, 2H),  $\delta$  7.21-  $\delta$  7.18 (m, 1H),  $\delta$  5.57 (s, 2H),  $\delta$  3.60 (s, 3H).  $^{13}C$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  167.12, 151.09, 136.43, 136.38, 134.72, 133.87, 133.04, 132.26, 131.78, 130.19, 128.46, 127.98, 127.34, 126.56, 123.93, 123.02, 69.13, 51.89. Mass (m/z) = 371 ( $M^+$ ). Anal. Calcd for  $C_{18}H_{14}N_2O_5S$ : C, 58.37; H, 3.81; N, 7.56. Found: C, 58.36; H, 3.81; N, 7.55.

#### General Synthetic procedure for synthesis of 19-20:

**(2E) - 2- {[2-methylquinolin-8-yl) oxy] methyl} - 3- (2- thienyl) acrylic acid (19):** Compound 6 (0.006 mole), 4N NaOH solution (0.006 mole) and methanol were stirred at room temperature for 15 h and the reaction was monitored by TLC. Solvent was completely distilled out and solid precipitates out when pH of the reaction was adjusted to 6-7 to obtain yellow solid Yield- 1.3g (86 %). m.p. 196 °C.

I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  3370, 1702, 1625, 1428, 1265, 838, 698 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.22 (s, 1H),  $\delta$  8.21 (d,  $J = 8.5$  Hz, 2H),  $\delta$  7.80 (d,  $J = 4.84$  Hz, 1H),  $\delta$  7.65 (d,  $J = 3.16$  Hz, 1H),  $\delta$  7.58-  $\delta$  7.45 (m, 4H),  $\delta$  7.19 (t, 1H),  $\delta$  5.18 (s, 2H),  $\delta$  2.73 (s, 3H).  $^{13}\text{C}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  167.18, 156.93, 148.63, 145.15, 143.61, 137.12, 136.15, 126.85, 125.79 (2C), 125.44, 124.72, 123.79, 122.46, 116.75, 24.54. Mass ( $m/z$ ) = 326 ( $M^+$ ). Anal. Calcd for  $\text{C}_{18}\text{H}_{15}\text{NO}_3\text{S}$ : C, 66.44; H, 4.65; N, 4.30. Found: C, 66.42; H, 4.64; N, 4.29.

**(2E) - 2- [(quinolin- 8- yloxy) methyl] - 3- (2- thienyl) acrylic acid (20):** Yield- 1.25 g (85 %). m.pt. 198°C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  3330, 1681, 1504, 1201, 1105, 823, 719 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.79 (t, 1H),  $\delta$  8.33 (d,  $J = 8.24$  Hz, 1H),  $\delta$  8.20 (s, 1H),  $\delta$  7.76 (d,  $J = 4.88$  Hz, 1H),  $\delta$  7.60 (d,  $J = 3.0$  Hz, 1H),  $\delta$  7.56-  $\delta$  7.51 (m, 3H),  $\delta$  7.40 (dd,  $J = 3.12, 5.48$  Hz, 1H),  $\delta$  7.16 (t, 1H),  $\delta$  5.12 (s, 2H).  $^{13}\text{C}$  NMR (300 MHz, DMSO- $d_6$  + TFA- $d$ ):  $\delta$  168.17, 148.86, 146.96, 145.28, 138.34, 136.84, 135.20, 133.02, 130.53, 129.90, 128.86, 123.08, 122.19, 120.71, 117.19, 114.22, 64.85. Mass ( $m/z$ ) = 312 ( $M^+$ ). Anal. Calcd for  $\text{C}_{17}\text{H}_{13}\text{NO}_3\text{S}$ : C, 65.58; H, 4.21; N, 4.50. Found: C, 65.55; H, 4.21; N, 4.50.

#### General Synthetic procedure for synthesis of 21-22:

**(2E)- N- benzyl- 2 - [(quinolin- 8- yl) oxy) methyl] – 3 - (2- thienyl) acrylamide (21):** Compound 20 (0.006 mole), benzyl amine (0.006 mole), Hydroxybenzotriazole (HOBt) (0.001 mole), 1-Ethyl-3- (3- dimethylaminopropyl) carbodiimide hydrochloride (EDC.HCl) (0.01 mole), Triethyl amine (0.05 mole) in dry dimethyl formamide were stirred for 10 h at room temperature. Water was added to the reaction mixture which precipitates out the solid was filtered and purified in methanol to obtain white solid. Yield- 1.1 g (68 %). m.pt. 135°C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1657, 1603, 1499, 1378, 1102, 979, 820, 706, 699 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  9.88 (s, 1H),  $\delta$  8.20 (s, 1H),  $\delta$  8.16 (t, 2H),  $\delta$  7.60-  $\delta$

7.44 (m, 3H),  $\delta$  7.52 (t, 3H),  $\delta$  7.41 (d,  $J = 6.88$  Hz, 3H),  $\delta$  7.32 (d,  $J = 4.8$  Hz, 1H),  $\delta$  7.29 (t, 1H),  $\delta$  7.15 (t, 1H),  $\delta$  5.3 (s, 2H),  $\delta$  4.7 (d,  $J = 5.6$  Hz, 2H). Mass ( $m/z$ ) = 401 ( $M^+$ ). Anal. Calcd for  $\text{C}_{24}\text{H}_{20}\text{N}_2\text{O}_2\text{S}$ : C, 71.98; H, 5.03; N, 6.99. Found: 71.96; H, 5.01; N, 6.98.

**(2E) - N- butyl - 2- [(quinolin-8-yloxy) methyl] - 3- (2- thienyl) acrylamide (22):** Yield-1.05 g (64 %). m.pt. 110°C. I.R (KBr pellets  $\text{cm}^{-1}$ ):  $\nu$  1655, 1600, 1500, 1375, 1242, 1102, 978, 820, 705 ppm.  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  8.91 (t, 1H),  $\delta$  8.24 (t, 1H),  $\delta$  8.11 (s, 1H),  $\delta$  7.60-  $\delta$  7.44 (m, 4H),  $\delta$  7.37 (d,  $J = 3.52$  Hz, 1H),  $\delta$  7.26 (s, 1H),  $\delta$  7.13 (dd,  $J = 3.72, 5.0$  Hz, 1H),  $\delta$  5.32 (s, 2H),  $\delta$  3.52 (q, 2H),  $\delta$  1.66 (q, 2H),  $\delta$  1.42 (q, 2H),  $\delta$  0.92 (t, 3H). Mass ( $m/z$ ) 367 ( $M^+$ ). Anal. Calcd for  $\text{C}_{21}\text{H}_{22}\text{N}_2\text{O}_2\text{S}$ : C, 68.82; H, 6.05; N, 7.64. Found: C, 68.80; H, 6.03; N, 7.63.

#### RESULTS AND DISCUSSION:

**Chemistry:** As per Scheme-1, Baylis-Hillman reaction was carried out using 1, 4-diazabicyclo [2.2.2] octane (DABCO) as catalyst without any solvent using thiophene-2-carboxaldehyde **1a** and methyl acrylate **2** as the starting compound for the synthesis of thienyl acrylates<sup>33</sup>. The structure of this intermediate was confirmed by I.R and NMR spectral analysis. The sharp absorption at 1716  $\text{cm}^{-1}$ , 1631  $\text{cm}^{-1}$ , and 1438  $\text{cm}^{-1}$  in the I.R spectrum showed that compound **3a** to be  $\alpha$ ,  $\beta$ -unsaturated ester.

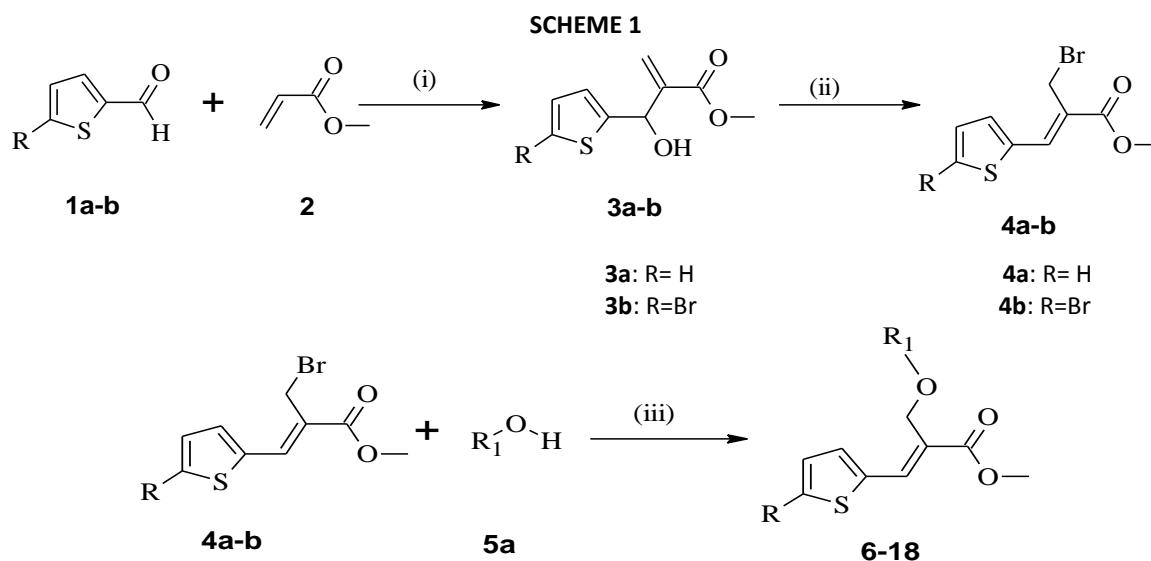
The broad absorption at 3448  $\text{cm}^{-1}$  in the I.R spectrum showed the presence of hydroxyl group. Three singlets at  $\delta$  5.7,  $\delta$  5.9 and  $\delta$  6.3 each integrating for one proton in the  $^1\text{H}$  NMR spectrum correspond to two vinylic protons and the single hydroxy methylene proton. The appearance of singlet at  $\delta$  5.7 is due to the presence of a hydroxy methylene group and the deshielding is due to its presence adjacent to vinylic group. Signals corresponding to thiophenyl protons appear around  $\delta$  6.9 and  $\delta$  7.7 and the 3-carbomethoxy

protons as singlet at  $\delta$  3.7. The compound **3a** is thus confirmed by the above spectral data. The hydroxy compound **3a** was converted to thienyl bromo ester **4a** by treatment with 47% Hydrobromic acid in the presence of concentrated sulphuric acid in Methylene dichloride at room temperature according to the reported procedure<sup>34</sup>. It was purified in hexane solvent to afford a low melting solid, which was the key intermediate in synthesizing the title compounds. The conversion of alcohol to bromide is evident from the appearance of absorption at  $723\text{ cm}^{-1}$  and disappearance of broad absorption around  $3448\text{ cm}^{-1}$  in the I.R spectrum.

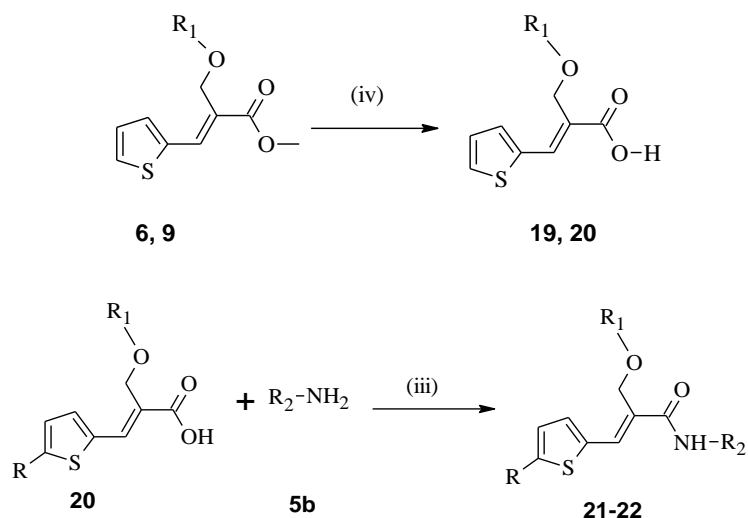
In  $^1\text{H}$  NMR spectrum, a singlet at  $\delta$  4.58 for two protons indicates the proton at the bromo methyl group and the vinylic protons appear much deshielded at  $\delta$  8.03. From the above data, the compound **4a** is structurally confirmed. The above synthesized bromo ester **4a** was treated with various hydroxyl quinoline compounds in presence of  $\text{K}_2\text{CO}_3$  in dry dimethyl formamide to afford a

series of esters (compounds **6** to **23**) at ambient temperature according to the reported procedure<sup>35-36</sup>. Completion of the reaction was judged by TLC and the isolation of products involves simple workup. Crude product obtained was further purified by simple recrystallization methods. Ether formation is evident from the absorption at  $1209\text{ cm}^{-1}$  and  $1101\text{ cm}^{-1}$  (compound **6**) and the oxymethylene protons appeared as singlet integrating for two protons at  $\delta$  5.10.

The vinyl proton of the  $\alpha, \beta$ -unsaturated system were deshielded significantly and appeared as a singlet at  $\delta$  8.23. In the  $^{13}\text{C}$  NMR spectrum, a signal at  $\delta$  167.31 correspond to the carbonyl, with the oxymethylene and methoxy carbons appearing at  $\delta$  52.50 and  $\delta$  21.03 respectively. The mass spectrum of **6** showed a molecular ion peak at  $m/z$  340.0 ( $\text{M}^+$ ), which further confirms the compound. The structures of the new compounds **6-22** are presented in **Table 1**.



R= H, Br  
 $\text{R}_1$ = 8-hydroxy quinoline  
 2-methyl-8-hydroxy quinoline  
 5, 7 dichloro-8-hydroxy quinoline  
 5, 7 dichloro-2-methyl-8-hydroxy quinoline  
 5-chloro-7-Iodo-8-hydroxy quinoline  
 5-nitro 8-hydroxy quinoline



R<sub>2</sub> = Benzyl amine  
Butyl amine  
Methyl amine

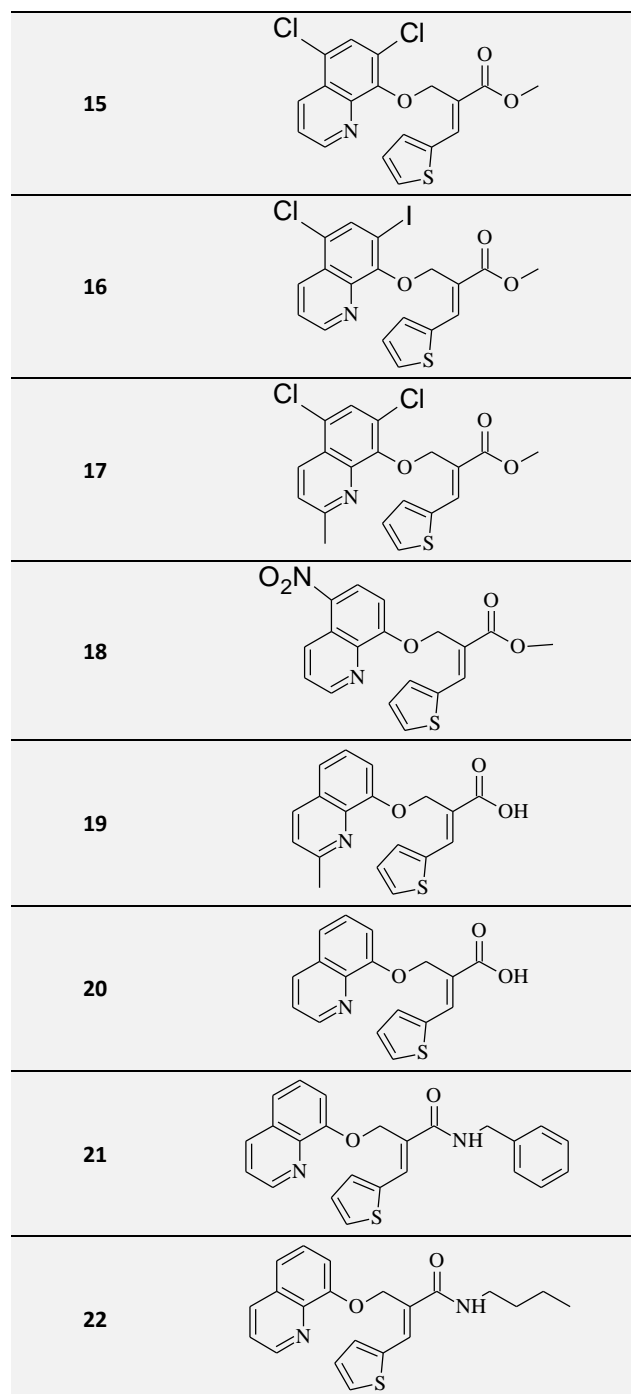
**Reagents and Condition:** (i) DABCO, room temperature, 80 h (ii) Hydrobromic acid, Concentrated sulphuric acid, Methylene dichloride, 5-10°C, 3 h (iii) K<sub>2</sub>CO<sub>3</sub>, Dimethyl formamide, room temperature, 1 h (iv) NaOH, Methanol, room temperature, 8 h (v) HOBT, EDC.HCl, Triethyl amine, Dimethyl formamide, room temperature, 10 h.

**TABLE 1: STRUCTURAL FORMULAE OF THE SYNTHESIZED COMPOUNDS, 6-23**

Comp. No	Structure
6	
7	
8	

9	
10	
11	
12	
13	
14	





### Pharmacology:

**In vitro antimicrobial activity:** Initial screening of antimicrobial susceptibility testing was carried out for all the seventeen synthesized compounds by agar well diffusion method as recommended by the CLSI<sup>37</sup>. Briefly, bacterial & yeast cultures were

grown to exponential phase concentration adjusted at recommended McFarland and plated by pour-plate method into the 150 mm Petri dishes and allowed to settle. Four representative bacterial isolates namely *S. aureus* ATCC 25923, *E. coli* ATCC 25922, *P. aeruginosa* ATCC 27853, *B. subtilis* ATCC 6633, and one fungal *Candida albicans* ATCC 90028 were used for screening. Wells were bored into the inoculated plate and the test compounds dissolved in DMSO were dispensed in the wells at three concentrations (10, 100 and 1000 µg/mL), allowed for complete diffusion and the plates were incubated at 37°C overnight.

Three antibacterial agents and one fungal (cefepime, amikacin, linezolid & fluconazole) were used as internal assay standards and 100% DMSO was used as a control. The zones of inhibition were measured using the digital Vernier's calipers. Further, the short-listed compounds were subjected to the determination of MIC by agar dilution method as per the CLSI guidelines<sup>37</sup>. Briefly, the selected bacterial strains were: Gram negative isolates of *E. coli* ATCC 25922, NCTC 13353, ATCC BAA200, *P. aeruginosa* ATCC 27853, *K. pneumoniae* ATCC 700603, ATCC 51503, *E. cloacae* 2160 P99+; and Gram positive strains *S. aureus* ATCC 25923, ATCC 43300, *B. subtilis* ATCC 6633 and fungal *Candida albicans* ATCC 90028.

The inoculum was prepared from 3- 5 well isolated bacterial colonies in 0.9 % NaCl such that the final inoculum spot contained  $1 \times 10^4$  CFU of bacteria when seeded onto agar plates containing different concentrations of the antimicrobial agents by a Multipoint Inoculator. The plates were incubated at 35°C for 18-20 h in an ambient atmosphere for bacterial strains and 48 h for fugal strains. MIC was recorded as the lowest concentration that inhibited visible growth of the inoculated culture.

The observations indicated that all the seventeen compounds screened for antibacterial activity exhibited a moderate to very good activity, but against the fungal strains all the compounds showed no activity (**Table 2**). The compounds which showed significant activity in the preliminary screening were further tested for antibacterial and antifungal activity by agar dilution method for determining the MIC as per CLSI guidelines and their results are tabulated (**Table 3**). Compounds in series which contain chloro/nitro displayed very good antibacterial activity against all the strains.

Compound 18 was highly active against *B. subtilis* ATCC 6633 (MIC = 1 µg/mL), *S. aureus* ATCC 43300 (MIC = 2 µg/mL) and *S. aureus* ATCC 25923 (MIC = 4 µg/mL), compound 15 and 16 were also active against *B. subtilis* ATCC 6633 (MIC = 2 µg/mL), *S. aureus* ATCC 43300 (MIC = 2 µg/mL) and *S. aureus* ATCC 25923 (MIC = 4 µg/mL). Compound 17 was highly active against *B. subtilis* ATCC 6633 (MIC = 2 µg/mL), and *S. aureus* ATCC 25923 (MIC = 4 µg/mL). Compounds 6, 9, 10 and 20 were also highly active against *S. aureus* ATCC 25923 (MIC = 4 µg/mL).

**TABLE 2: ANTIBACTERIAL ACTIVITIES OF THE NEWLY SYNTHESIZED COMPOUNDS (ZONE OF INHIBITION IN mm)**

Comp. No	<i>S. aureus</i> ATCC 25923			<i>B. subtilis</i> ATCC 6633			<i>E. coli</i> ATCC 25922			<i>P. aeruginosa</i> ATCC 27853		
	1000 µg	100 µg	10 µg	1000 µg	100 µg	10 µg	1000 µg	100 µg	10 µg	1000 µg	100 µg	10 µg
6	25.56	24.1	23.15	27.23	26.33	24.5	10.9	nz	nz	11.56	7.15	nz
7	19.35	18.36	8.54	18.2	14.32	nz <sup>#</sup>	nz	nz	nz	9.35	nz	nz
8	15.85	14.99	12.04	nz	13.15	nz	nz	nz	nz	4.99	nz	nz
9	27.31	25.28	24.85	28.54	27.63	26.98	8.91	nz	nz	9.28	nz	nz
10	26.84	25.11	23.48	27.59	26.55	25.28	6.32	nz	nz	10.01	4.06	nz
11	23.96	16.96	12.66	nz	8.66	nz	nz	nz	nz	nz	nz	nz
12	12.39	8.24	nz	nz	nz	nz	7.32	nz	nz	14.24	nz	nz
13	12.39	11.6	nz	nz	nz	nz	nz	nz	nz	11.6	nz	nz
14	20.05	21.16	12.13	nz	16.5	nz	nz	nz	nz	nz	nz	nz
15	29.24	29.15	28.54	28.93	27.78	27.10	10.3	7.62	nz	21.16	12.13	nz
16	29.54	28.95	28.24	27.68	26.54	26.32	18.3	11.2	nz	20.01	6.06	nz
17	28.64	27.32	27.10	27.10	26.28	25.32	9.82	nz	nz	14.99	8.04	nz
18	29.94	28.94	28.69	28.94	27.93	27.58	19.6	15.2	nz	23.96	12.66	nz
19	22.56	20.81	18.32	24.65	22.18	20.23	nz	nz	nz	9.56	nz	nz
20	27.31	25.28	24.85	nz	nz	nz	4.85	nz	nz	9.28	nz	nz
21	20.35	18.66	12.10	16.82	15.18	13.28	nz	nz	nz	6.12	nz	nz
22	14.96	12.47	10.78	16.86	13.15	11.31	nz	nz	nz	nz	nz	nz
Linezolid		30.16			30.53			nz			nz	
Amikacin		25.42			35.13			25.23			26.37	
Cefepime		28.21			37.6			31.47			26.69	

<sup>#</sup>nz: no zone of inhibition

**TABLE 3: MINIMUM INHIBITORY CONCENTRATION (MIC, µg/ml) OF SELECTED COMPOUNDS AGAINST GRAM POSITIVE BACTERIAL STRAINS<sup>a</sup>**

Compound No.	<i>S. aureus</i> ATCC 25923	<i>S. aureus</i> ATCC 43300	<i>B. subtilis</i> ATCC 6633
6	16	32	4
9	4	8	2
10	4	8	8
15	4	2	2
16	4	2	2
17	4	16	4
18	4	2	1
19	16	32	4
20	4	8	8
Linezolid	4	2	1
Amikacin	2	128	0.5
Cefepime	2	32	2

<sup>a</sup> Agar dilution method used to determine the MIC

**CONCLUSION:** In conclusion, from an SAR standpoint, we observe that, of all compounds tested (6-23 of table 1), the highest antimicrobial activity occurs in compound 15, 16 and 18 when the molecule possesses the choro substituents in position 5 and 7, the choro and iodo substituents in position 5 and 7 and nitro substituents in position 5. Hence it is concluded that there is ample scope for further developing this field.

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#### REFERENCES:

1. Richet H, Mohammed J, McDonald LC and Jarvis WR: Emerg Infec Dis 2001; 7: 319.
2. Payne DJ, Gwynn MN, Holmes DJ and Rosenberg M: Methods Mol Biol 2004; 266: 231.
3. Service RF: Science 1995; 270: 724.
4. Masunari A and Tavares LC: Biorganic and Medicinal Chemistry letters 2007; 15: 4229.
5. Foroumadi A, Mansouri S, Kiani Z and Rahmani A: European Journal of Medicinal Chemistry 2003; 38(9): 851.
6. Pires JR, Saito C, Gomes SL, Giesbrecht AM and Amaral AT: Journal of Medicinal Chemistry 2004; 44: 3673.
7. Kleemann A, Engel J, Kutscher and Reichert: Pharmaceutical substances; Synthesis, Patents, Applications; Thieme, 4<sup>th</sup>, ed.; vols. 1 and 2; 2001.
8. Van Rhee AM, Diddiqi SM, Melman N, Shi D, Padgett W, Daly JW and Jacobson KA: Journal of Medicinal Chemistry 1996; 39: 398.
9. Yasuma T, Oda T, Hazama M and Taketomi S: PCT Int. Appl. WO 98 09958, 1998 (Chem. Abstr. 1998; 128: 217366v).
10. McMahon G, Hirth KP and Tang PC: PCT Int. Appl. WO 96 40113, 1995 (Chem. Abstr. 1997; 126: 126918u).
11. Broughton H, Chambers MS, Hobbs SCh, Macleod AM and Reeve AJ: PCT Int. Appl. WO 98 18792, 1998 (Chem. Abstr. 1998; 129: 4573u).
12. AL-Obaid AM, El-Subbagh HJ, Khodair AI and El-Mazar MMA, Anticancer Drugs: 1996: 7: 873.
13. Alireza Foroumadi, Mehdi Oboudiat, Saeed Emami, Alireza Karimollah, Lotfollah Saghaee, Mohammad Hassan Moshafi and Abbas Shafiee: Biorganic and Medicinal Chemistry letters 2006; 14: 3421.
14. Sonali Rudra, Fnu Sangita, Arti Gujrati, Manisha Pandya, Pragya Bhateja, Tarun Mathur, Smita Singhal, Ashok Rattan, Mohammed Salman and Biswajit Das: Biorganic and Medicinal Chemistry letters 2007; 17: 4778.
15. Ryuichi Kimura, Takahiro Yabuuchi and Masakatsu hisaki: Chemical and pharmaceutical bulletin 1962; 10(12): 1226.
16. Yeong Woo Jo, Weon Bin Im, Jae Keol Rhee, Mi Ja Shim, Won Bae Kim and Eung Chil Choi: Biorganic and Medicinal Chemistry letters 2004; 12: 5909.
17. Chandrasekaran S and Nagarajan S: II Farmaco 2005; 60: 279.
18. Sam J and Advani GG: J Pharmaceutical Science 2006; 54: 753.
19. Daniel Dube, Marc Blouin, Christine Brideau, Chi-Chung Chan, Sylvie Desmarais, Diane Ethier, Jean-Pierre Falguyret, Richard Friesen W, Mario Girard, Yves Girard, Jocelyne Guay, Denis Riendeau, Philip Tagari and Robert Young N: Biorganic and Medicinal Chemistry letters 1998; 8: 1255.
20. Allen FH, Kennard O, Watson DG, Brammer L, Orpen AG and Taylor R: J Chem Soc Perkin Trans 1987; 2: S1.
21. Allen Jr GR, Baker BR, Dornbush AC, Joseph JP, Kissman H and Weiss MJ: J Med Pharm Chem 1960; 2: 391.
22. El-Agrody AM, Abd El-Latif MS, El-Hady NA, Fakery AH and Bedair: Molecules 2001; 6: 519.
23. Pratibha S and Shreeya P: Indian Journal of Chemistry 1999; 38B: 1139.
24. Michael JP: Nat Prod Rep 1997; 14: 605.
25. Balasubramanian M and Keay JG: In Comprehensive Heterocyclic Chemistry II. Katritzky AR, Rees CW and Scriven EFV, Eds Pergamon Press. Oxford 1996; 5: 245.
26. Baylis AD and Hillman: MED. Ger Pat 2,155,113, 1972.
27. Drewes SE and Roos GHP: Tetrahedron 1988; 44: 4653.
28. Basavaiah D, Rao PD and Hyma RS: Tetrahedron 1996; 52: 8001.
29. Kim JN and Lee KY: Curr Org chem 2002; 6: 627.
30. Basaviah D, Rao A.J and Satyanarayana T: Chem Rev 2003; 103: 811.

31. Ciganek E: Organic Reactions, Paquette, L.A.Ed. Wiley, New York 1997; 51: 201.
32. Langer P: Angew Chem Int Ed 2000; 39: 3049.
33. Yu C, Liu B and Hu L: Journal of organic chemistry 2001; 66: 5413.
34. Bucholz R and Hoffmann HMR: Helv Chim Acta 1991; 74: 213.
35. Rajan YC, Kanakam CC, Selvam SP and Murugaesan K: Tetrahedron Letters 2007; 48: 8562.
36. Rajan YC and Kanakam CC: Tetrahedron Letters 2008; 49: 3023.
37. Performance standards for antimicrobial susceptibility testing: Eighteenth information supplement Jan 2008; Vol 28 No. 1: M100-S18.

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