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SIMULTANEOUS ESTIMATION OF DICLOFENAC SODIUM AND TOLPERISONE HYDROCHLORIDE IN COMBINED PHARMACEUTICAL FORMULATION

Bhavesh Gevriya* and R.C. Mashru

Quality Assurance Laboratory, Centre of Relevance and Excellence in Novel Drug Delivery System, Pharmacy Department, G. H. Patel Building, Donor's Plaza, The Maharaja Sayajirao University of Baroda, Fatehgunj, Vadodara – 390 002, Gujarat, India

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Correspondence to Author:

Bhavesh Gevriya

Quality Assurance Laboratory, Centre of Relevance and Excellence in Novel Drug Delivery System, Pharmacy Department, G. H. Patel Building, Donor's Plaza, The Maharaja Sayajirao University of Baroda, Fatehgunj, Vadodara – 390 002, Gujarat, India

E-mail: gevriya.bhavesh@gmail.com

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ABSTRACT

Three simple, rapid, precise and accurate spectrophotometric methods have been developed for simultaneous analysis of Tolperisone Hydrochloride (TOL) and Diclofenac Sodium (DIC) in their combined dosage form. Method A, Simultaneous equation method (Vierodt's method) applies measurement of absorptivities at two wavelengths, 261.00 nm (λ_{\max} of Tolperisone Hydrochloride) and 279.00 nm, (λ_{\max} of Diclofenac Sodium) in zero order spectra. The concentrations can be calculated from the derived equations. Method B, Q-Absorbance equation method. It involves formation of Q-absorbance equation at 233.50 nm (isoabsorptive point) and 261.00 nm (λ_{\max} of Tolperisone Hydrochloride) in zero order spectra. Method C, Zero crossing first derivative spectrophotometry involves measurement of absorbance at 249.20 nm (for Tolperisone Hydrochloride) and 227.40 nm (for Diclofenac Sodium) in first derivative spectra. Developed methods were validated according to ICH guidelines. The calibration graph follows Beer's law in the range of 6.0 to 18.0 $\mu\text{g/ml}$ for Tolperisone Hydrochloride and 2.0 to 6.0 $\mu\text{g/ml}$ for Diclofenac Sodium with R square value greater than 0.999. Accuracy of all methods was determined by recovery studies and showed % recovery between 98 to 102%. Intraday and interday precision was checked for all methods and mean %RSD was found to be less than 2 for all the methods. The methods were successfully applied for estimation of Tolperisone Hydrochloride and Diclofenac Sodium in marketed formulation.

INTRODUCTION: Tolperisone Hydrochloride (TOL) chemically is 1-piperidino-2-methyl-3-(p-tolyl)-3 propanone hydrochloride (**Fig. 1**), is a centrally acting Muscle Relaxant for the Symptomatic treatment of Spasticity and Muscle Spasm.

Tolperisone Hydrochloride is official in Japanese pharmacopoeia JP15. JP15 describes Potentiometric Titration for its estimation¹.

Diclofenac Sodium (DIC) Chemically is sodium 2-[(2,6-dichlorophenyl)-amino]phenylacetate (**Fig. 2**), is a broadly used non-steroidal anti-inflammatory drug for the treatment of inflammatory conditions such as rheumatoid arthritis, osteoarthritis and ankylosing spondylitis. Diclofenac Sodium is official in Indian Pharmacopoeia (IP), British Pharmacopoeia (BP) and United States Pharmacopoeia (USP). IP, BP, USP describes potentiometric titration for its estimation.^{[2], 3,4}

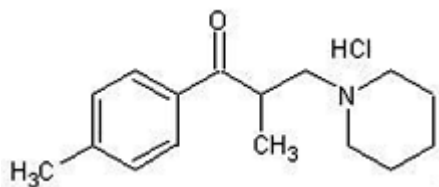


FIG. 1: TOLPERISONE HYDROCHLORIDE

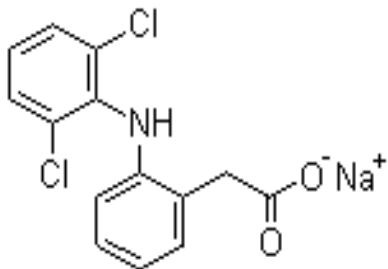


FIG. 2: DICLOFENAC SODIUM

Objective of Study: Survey of literature revealed that numbers of method have been reported in literature for the individual analysis of Tolperisone Hydrochloride and Diclofenac sodiumm by UV spectrophotometric and RP-HPLC method. UV spectrophotometric method available in literature for simultaneous determination of Diclofenac Sodium with other drugs like Peracetamol, Misoprostol and Thiocolchicoside Nimesulide ^{5, 6, 7, 8}. RP-HPLC method available in literature for simultaneous determination of Tolperisone Hydrochloride with Etodolac.^[9] However, to our knowledge, there is no reported uv-spectrophotometric method available for simultaneous estimation of Tolperisone Hydrochloride and Diclofenac Sodium.

The aim of the present work was to develop easy, economic, accurate, specific and precise spectrophotometric methods for simultaneous estimation of Tolperisone Hydrochloride and Diclofenac Sodium in bulk drugs and combined pharmaceutical formulations and validation of newly developed analytical methods.

MATERIALS AND METHODS:

Apparatus and Software: Shimadzu UV-1700 double beam spectrophotometer connected to a computer loaded with Shimadzu UV Probe 2.10 software was used for all the spectrophotometric measurements. The absorbance spectra of the reference and test solutions were carried out in 1cm quartz cells over the range of 200-400 nm. The samples were weighed on electronic analytical balance (A×120, shimadzu).

Reagents and Chemicals:

Solvent: Methanol: Water (30:70), Methanol analytical reagent grade (Spectrochem Pvt. Ltd, Mumbai, India). Water, single distilled water.

Diluent: Methanol: Water (30:70), Methanol analytical reagent grade (Spectrochem Pvt. Ltd, Mumbai, India). Water, single distilled water.

Year of Experiment– 2012.

Site- Quality Assurance Laboratory, Centre of Relevance and Excellence in Novel Drug Delivery System, G. H. Patel Building, Donor's Plaza, The Maharaja Sayajirao University of Baroda, Fatehgunj, Vadodara – 390 002, Gujarat, India.

Preparation of Stock Solution: Accurately weighed TOL and DIC (in quantities of 25.0 mg) were transferred to two separate 25 ml volumetric flasks, dissolved with the use of methanol : water (30:70) and volume was made up to the mark with methanol : water (30:70) to obtain stock solution of TOL (1000 µg/ml) and DIC (1000 µg/ml)

Preparation of Working Standard Solution: From the above solution, standard stocks solutions of TOL (50 µg/ml) and DIC (50 µg/ml) were prepared by transferring 2.5 ml aliquots to 50 ml volumetric flasks and making up the volume with methanol : water (30:70).

Preparation of Calibration Curve of Standard TOL and DIC: From working std. solution of TOL (50 µg/ml) 1.2, 1.8, 2.4, 3.0 and 3.6 ml were transferred to 10 ml volumetric flasks and volume were made up to the mark with methanol : water (30:70). This gives 6.0 to 18 µg/ml of TOL. From working std. solution of DIC (50 µg/ml) 0.4, 0.6, 0.8, 1.0 and 1.2 ml were transferred to 10 ml volumetric flasks and volume were made up to the mark with methanol : water (70:30). This gives 2.0 to 6.0 µg/ml of DIC.

Method A-

Simultaneous Equation Method (Vierodt's Method): If a sample containing two absorbing drug (X and Y) each of which absorbs at λ_{max} of other. It may possible to determine both drugs by the technique of

simultaneous equations (Vierodt's method) provided that certain criteria apply. The information required is the absorptivities of X at λ_1 and λ_2 a_{x1} and a_{x2} respectively (a) The absorptivities of Y at λ_1 and λ_2 a_{y1} and a_{y2} respectively (b) The absorbances of the diluted sample at λ_1 and λ_2 , A_1 and A_2 respectively. Let C_x and C_y be the concentrations of X and Y respectively in the diluted sample. Two equations are constructed based upon the fact that at λ_1 and λ_2 the absorbance of the mixture is the sum of the individual absorbance of X and Y.

From the stock solutions, working standard solutions of TOL (50 $\mu\text{g/ml}$) and DIC (50 $\mu\text{g/ml}$) were prepared. By appropriate dilutions, the solutions with concentrations 6.0-18 $\mu\text{g/ml}$ (for TOL) and 2.0-6.0 $\mu\text{g/ml}$ (for DIC) were prepared and scanned between 200 to 400 nm (Fig. 3). Calibration curve of absorbance

versus concentration were prepared. The calibration curves were found to be linear in the concentration range under study (Fig. 4.1 & 4.2). For TOL and DIC, analytical wavelengths of 261.00 nm and 279.00 nm were selected respectively. Absorptivity of TOL and DIC were calculated at both the wavelengths. The concentrations of TOL and DIC can be calculated from following equations^[10]:

$$C_x (\text{DIC}) = (A_2 a_{y1} - A_1 a_{y2}) / (a_{x2} a_{y1} - a_{x1} a_{y2})$$

$$C_y (\text{TOL}) = (A_1 a_{x2} - A_2 a_{x1}) / (a_{x2} a_{y1} - a_{x1} a_{y2})$$

Where; C_x & C_y are concentrations of DIC and TOL respectively in gm/100 ml in the sample solution. A_1 & A_2 are the absorbances of the mixture at 261.00 nm & 279.00 nm respectively; a_{x1} and a_{x2} = Absorptivity of DIC at 261.00 nm and 279.00 nm; a_{y1} and a_{y2} = Absorptivity of TOL at 261.00 nm and 279.00 nm.

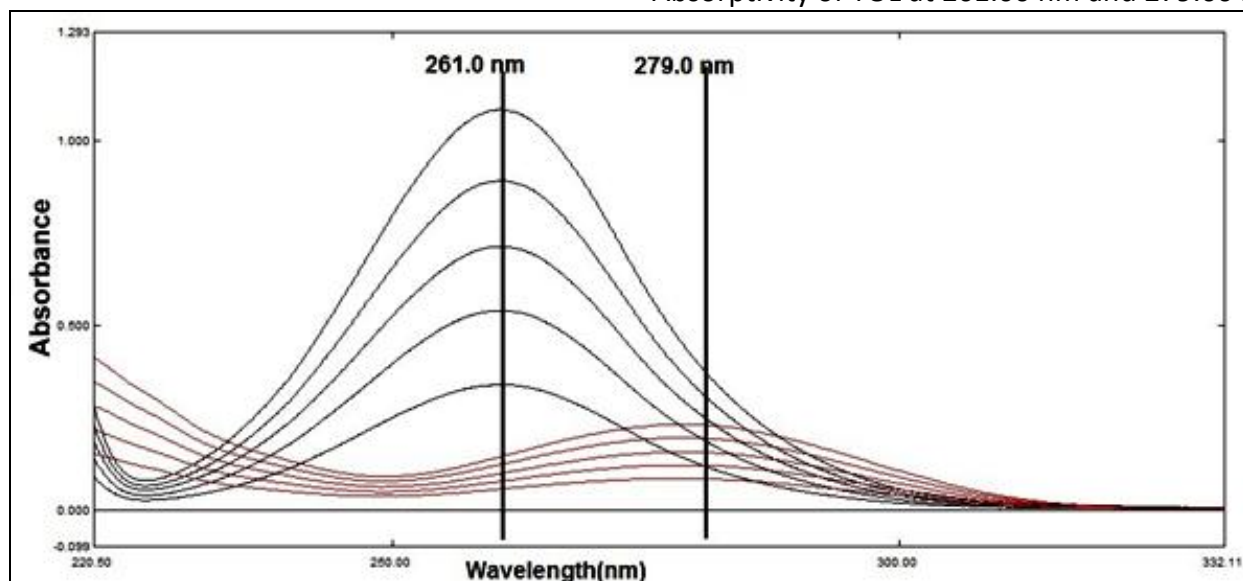


FIG. 3: ZERO ORDER OVERLAIN SPECTRA OF DIC (2, 3, 4, 5, 6 $\mu\text{g/ml}$, RED) AND TOL (6, 9, 12, 15, 18 $\mu\text{g/ml}$, RED)

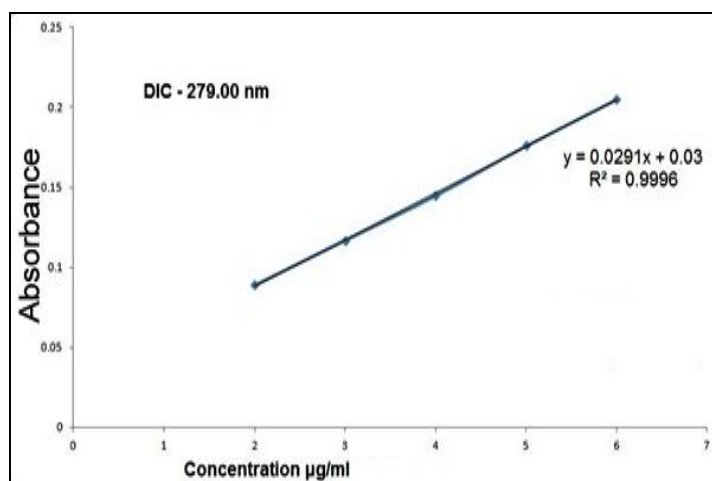


FIG. 4.1 CALIBRATION GRAPH OF DIC.

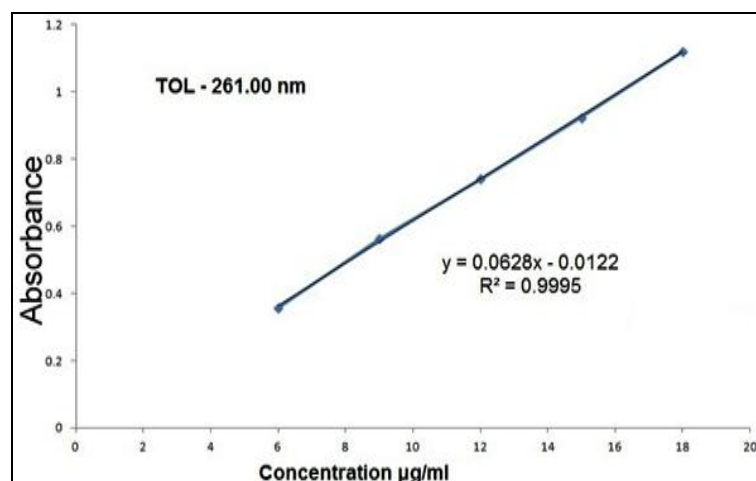


FIG. 4.2: CALIBRATION GRAPH OF TOL.

Method B-

Q-Absorbance Ratio Method: Q method uses the ratio of absorbances at two selected wavelengths, one at isoabsorptive point and other being the λ_{max} of one of the two compounds. From the stock solutions, working standard solutions of TOL (50 $\mu\text{g/ml}$) and DIC (50 $\mu\text{g/ml}$) were prepared. By appropriate dilutions, the solutions with concentrations 6.0-18 $\mu\text{g/ml}$ (for TOL) and 2.0-6.0 $\mu\text{g/ml}$ (for DIC) were prepared and scanned between 200 to 400 nm (**Fig. 5**). series of standard solutions ranging from 2.0-6.0 $\mu\text{g/ml}$ for DIC and 6.0-18.0 $\mu\text{g/ml}$ for TOL were prepared and the absorbance of solutions was recorded at 233.50 (isoabsorptive point) and 261.00 nm (λ_{max} of TOL) to plot a calibration curve of absorbance versus concentration (**Fig. 6.1 & 6.2**). Calibration curves were found to be linear in the

concentration range under study. Absorptivity values of DIC and TOL were determined at selected wavelengths and are presented in **Table 2**. The concentration of two drugs in mixture was calculated by using following equations^[10]:

$$CX = [(QM - QY) / (QX - QY)] \times A1/aX1$$

$$CY = [(QM - Qx) / (Qy - Qx)] \times A1/ay1$$

Where; $Q_m = A_2/A_1$, $Q_x = a_{x2}/a_{x1}$, $Q_y = a_{y2}/a_{y1}$; **1** designates isoabsorptive point and **2** designates λ_{max} of TOL; a_{x1} and a_{x2} is Absorptivity of DIC at 1 and 2 wavelength respectively; a_{y1} and a_{y2} is Absorptivity of TOL at 1 and 2 wavelength respectively; A_1 and A_2 are absorbances of the mixture at 1 and 2 wavelength respectively.

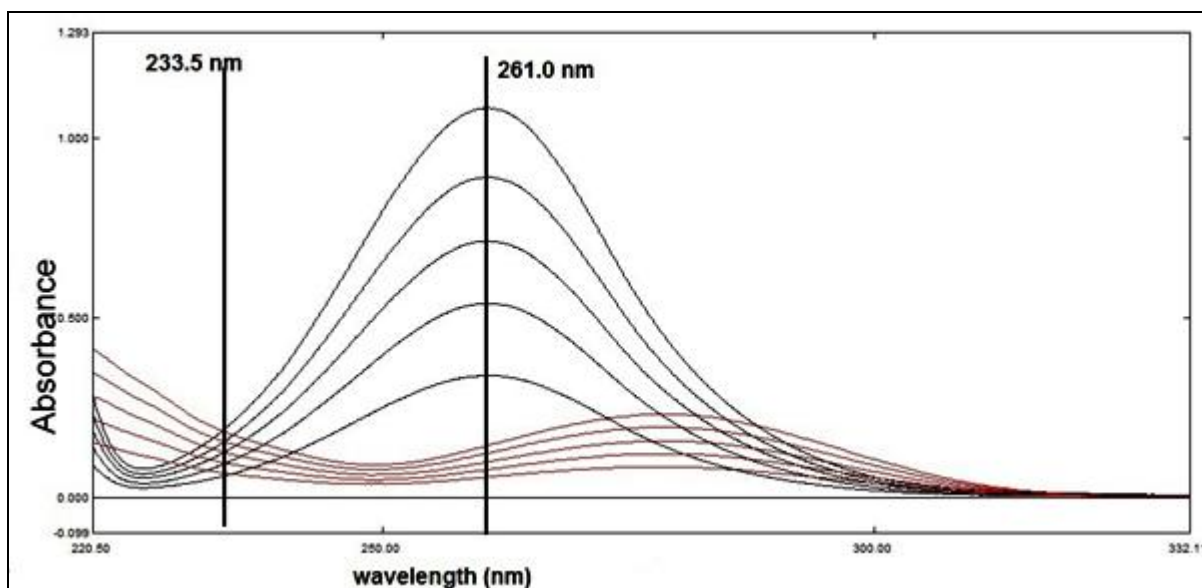


FIG. 5: ZERO ORDER OVERLAIN SPECTRA OF DIC (2, 3, 4, 5, 6 $\mu\text{g/ml}$, RED) AND TOL (6, 9, 12, 15, 18 $\mu\text{g/ml}$, RED)

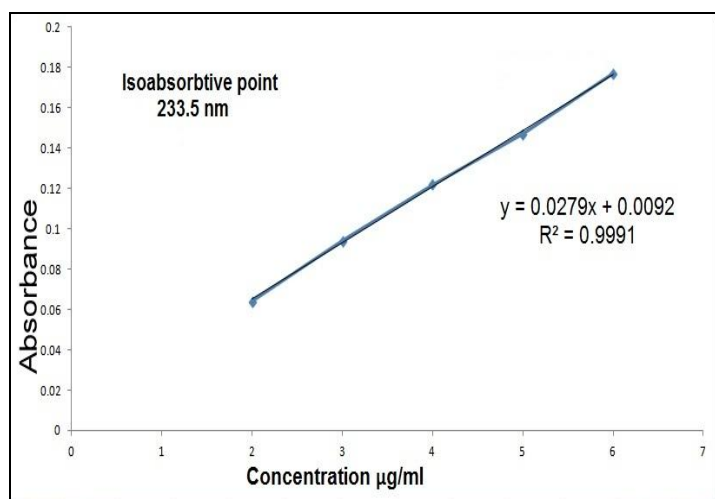


FIG. 6.1: CALIBRATION GRAPH AT ISOABSORPTIVE POINT

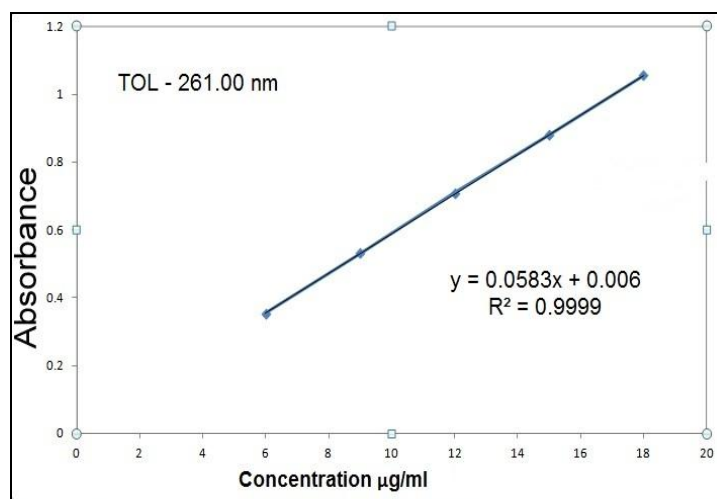


FIG. 6.2: CALIBRATION GRAPH OF TOL

Method C-**Zero Crossing First Derivative Spectrophotometry:**

The solutions of standard DIC and TOL were prepared in the range of 2.0 to 6.0 $\mu\text{g/ml}$ and 6.0 to 18.0 $\mu\text{g/ml}$ respectively. The absorption spectra of the solutions of DIC and TOL were recorded in the range of 200 nm to 400 nm and were stored in the memory of the instrument and transformed to first derivative with $\Delta\lambda = 8\text{nm}$ and scaling factor 50 (Fig. 7). At 227.40 nm, TOL

is having zero crossing point and DIC can be determined. At 249.20 nm, DIC is having zero crossing point and TOL can be determined. The amplitudes at 227.40 nm were plotted against respective concentrations of DIC and the amplitudes at 249.20 nm were plotted against the respective concentrations of TOL for the preparation of calibration graph¹⁰. Calibration graph for DIC and TOL are shown (Fig. 8.1 & 8.2).

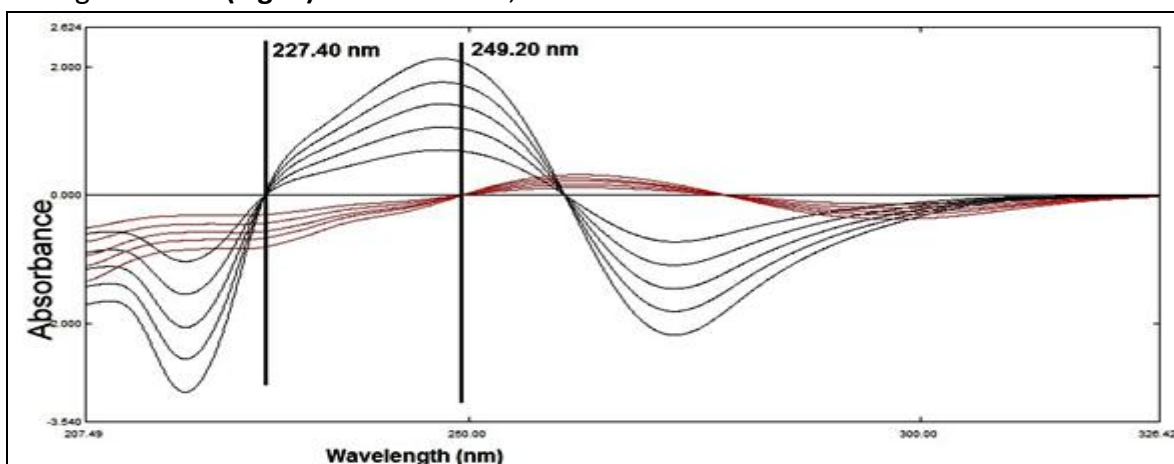


FIG. 7: FIRST ORDER OVERLAIN SPECTRA OF DIC (2, 3, 4, 5, 6 $\mu\text{g/ml}$, RED) AND TOL (6, 9, 12, 15, 18 $\mu\text{g/ml}$, RED)

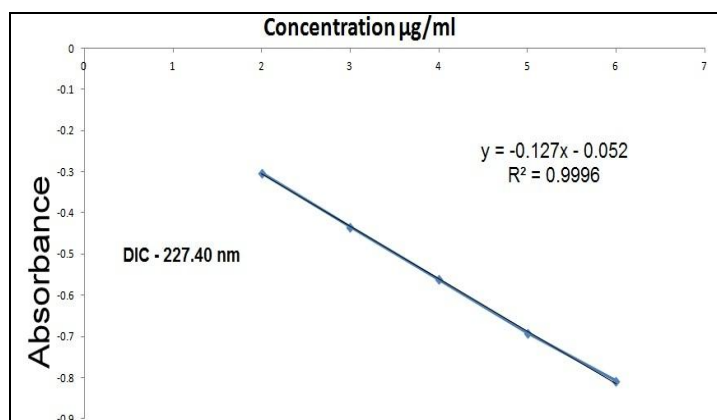


FIG. 8.1: CALIBRATION GRAPH OF DIC

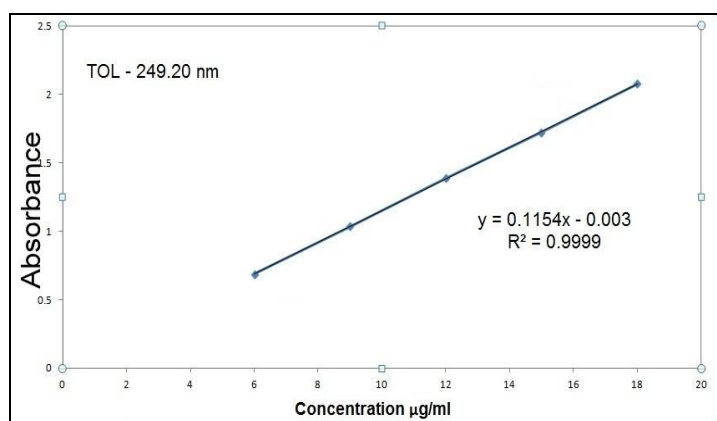


FIG. 8.2: CALIBRATION GRAPH OF TOL

Assay of Commercial Formulation by Method A, B, and C: 20 tablets were powdered and an amount equivalent to 25 mg DIC and 75 mg TOL was weighed and dissolved in 25 ml methanol: water (30:70). Solutions were filtered using whatmann filter paper grade 1. Appropriate dilutions were prepared in methanol: water (30:70) taking suitable aliquots of the clear filtrates and subjected to analysis using all the four methods described above. The result of analysis is reported (Table 1).

TABLE 1: RESULTS OF SIMULTANEOUS ESTIMATION OF MARKETED FORMULATION FOR METHOD A, B, AND C

Formulation:- Tolpidol – D		
Labelled Claim :- DIC : TOL (50mg : 150mg)		
Method	DIC* \pm SD	TOL* \pm SD
A	99.33 \pm 0.0308	99.67 \pm 0.0724
B	102.5 \pm 0.0353	100.83 \pm 0.0264
C	99.00 \pm 0.0316	101.00 \pm 0.0515

*Mean value of five determinations

RESULTS AND DISCUSSION: Developed spectrophotometric methods for the simultaneous estimation of DIC and TOL were validated according to ICH guidelines and data complying with the standards were obtained¹¹. The results of validation parameters for all the three developed methods are reported (Table 2 and 3).

TABLE 2: SUMMARY OF VALIDATION PARAMETERS BY DEVELOPED METHODS

Parameters	Method - A		Method - B		Method - C	
	DIC	TOL	DIC	TOL	DIC	TOL
Analytical wavelength (nm)	261.00 & 279.00		233.50 & 261.00		227.40	249.20
Beer's range ($\mu\text{g/ml}$)	2.0 -6.0	6.0 -18.0	2.0 - 6.0	6.0 - 18.0	2.0 - 6.0	6.0 -18.0
Slope	0.0291	0.0268	0.0279	0.0583	0.127	0.1154
Intercept	0.0300	0.0122	0.0092	0.006	0.052	0.003
Correlation coefficient	0.9996	0.9995	0.9991	0.9999	0.9996	0.9999
Intraday precision (%RSD)	0.6622	0.4484	1.2067	0.7898	1.1994	0.8171
Interday precision (%RSD)	1.3986	0.6990	1.5618	0.8453	1.3586	0.8801
LOD ($\mu\text{g/ml}$)	0.2884	0.3832	0.3248	0.3248	0.3092	0.4146
LOQ ($\mu\text{g/ml}$)	0.8652	1.1495	0.9744	1.3211	0.9276	1.2439

TABLE 3: RESULTS OF RECOVERY STUDY OF DIC AND TOL BY DEVELOPED METHODS

METHOD	% SPIKING	C_{ACTUAL} $\mu\text{g/ml}$		C_{ADDED} $\mu\text{g/ml}$		C_{FOUND}^* $\mu\text{g/ml}$		%RECOVERY \pm S.D.	
		DIC	TOL	DIC	TOL	DIC	TOL	DIC	TOL
		A	50	2	6	1	3	3.02	9.04
100	2		6	2	6	4.08	11.97	102.00 \pm 0.0406	99.75 \pm 0.1768
150	2		6	3	9	5.03	15.18	100.60 \pm 0.1267	101.20 \pm 0.0778
B	50	2	6	1	3	3.08	9.03	102.88 \pm 0.0768	100.37 \pm 0.0644
	100	2	6	2	6	4.07	11.84	101.75 \pm 0.1353	98.67 \pm 0.1058
	150	2	6	3	9	5.12	14.99	102.4 \pm 0.0557	99.93 \pm 0.1127
C	50	2	6	1	3	3.05	9.08	101.67 \pm 0.0158	100.88 \pm 0.1389
	100	2	6	2	6	4.09	12.09	102.25 \pm 0.0308	100.75 \pm 0.0818
	150	2	6	3	9	5.02	15.19	100.4 \pm 0.0872	101.27 \pm 0.0505

* Mean of three determinations

CONCLUSION: Three Spectrophotometric methods (Q-Absorbance equation method, Simultaneous equation method, Zero crossing first derivative spectrophotometry) were developed for simultaneous estimation of DIC and TOL in their combined formulation without prior separation. Methods were found to be precise and accurate as can be reflected from validation data. Developed methods were successfully applied for estimation of DIC and TOL in marketed formulation.

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