PURIFICATION AND SEPERATION OF INDIVIDUAL CURCUMINOIDS FROM SPENT TURMERIC OLEORESIN, A BY-PRODUCT FROM CURCUMIN PRODUCTION INDUSTRY.

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ABSTRACT: Turmeric, curcuma longa L of Zingiberaceae family is a widely cultivated spice in India and other Asian countries. Turmeric is rich in curcuminoids, and recognized for their broad spectrum of biological activities, curcuminoids vary in chemical structures, physico-chemical characteristics as well as the functional properties. This study focused on separation of individual curcuminoi ds from spent turmeric oleoresin considered as an industrial waste and its purification by column chromatography followed by purity analysis by HPLC. Chromatographic purification of spent turmeric oleoresin was done with Diaion HP 20 resin, fractions were collected. Curcuminoi d enriched oleoresin fractions combined & then subjected to hexane precipitation, purified curcuminoid enriched fraction obtained. Purified curcuminoid enriched fraction subjected to silica gel column chromatography to separate the individual curcuminoi ds with chloroform: methanol at increasing polarity. 10ml each 40 fractions were collected and TLC profile done. Same Rf fractions collected & evaporated. Total curcuminoi ds of individual fractions were determined by HPLC. Recrystallization of each compound was done using methanol: chloroform (5:1) at 3°C. Curcumin, Demethoxy curcumin, Bisdemethoxy curcumin having purities 98.4%, 97.1%, 97.3% are separated from the spent turmeric oleoresins. The above study reveals that the spent turmeric oleoresin being wasted at present can be used to isolate and separate the remaining curcuminoi ds.

INTRODUCTION: Phytochemicals are non – nutritive plant chemicals that have protective or disease preventive properties. It is well known that plants produce these chemicals to protect themselves but recent research demonstrates that they can also protect humans against diseases.

There are more than thousand known phytochemicals available in the scenario. Some of the well known phytochemicals are curcumin from turmeric, lycopene in tomatoes , isoflavones in soy and flavanoids in fruits.

There has been considerable public and scientific interest in the use of phytochemicals derived from dietary components to combat human diseases, especially the two commonest killers in the developed world, cardiovascular disease and cancer. Curcumin is one such medicine. Its history goes back over 5000 years , to the heyday of Ayurveda ( which means the science of long life) ¹.
Turmeric (Curcuma longa) a rhizomatous herbaceous perennial plant of family Zingiberacea, has been used for centuries in Indian and Chinese traditional medicine for the ailment of various diseases, including biliary disorders, anorexia, Cough, Diabetic wounds, hepatic disorders and sinusitis. It is a spice of golden color that is used in cooking in the Indian subcontinent. Because of its color and taste, turmeric was named Indian saffron in Europe.

Today, India is the primary exporter of turmeric (known as haldi in India). Although its ability to preserve food through its antioxidant mechanism, to give color to food, and to add taste to the food is well known, its health promoting effects are less well recognized or appreciated. Turmeric being anti microbial, is used extensively for cosmetic applications. Turmeric is the only spice which finds application in all the three segments of life-food, cosmetics and health. In ayurveda medicine, turmeric has a long history of use for its anti-Inflammatory and anti-arthritic effects.

Curcumin (C), main coloring substance in *Curcuma longa* and two related compounds, demethoxycurcumin (DMC) and bisdemethoxycurcumin (BDMC), are altogether known as curcuminoid. The chemical structures of three curcuminoids and its molecular mass are shown in Fig. 1 and Table 1.

The total of curcuminoids which is about 7-10%, turmeric also contains 2-4% essential oil, including turmerone, atlantone, and zingiberone and 2-3% of fixed oil. Other constituents include sugars, proteins and resins. Curcumin was first isolated from turmeric in 1815, and the structure was delineated in 1910 as diferuloylmethane. Most currently available preparations of curcumin contain approximately 70-77% diferuloylmethane, 14-18% demethoxycurcumin and 3-5% bisdemethoxycurcumin. Curcumin is hydrophobic in nature and frequently soluble in dimethylsulfoxide, acetone, ethanol, and oils. It has absorption maxima around 420nm. When exposed to alkaline conditions, the color of curcumin turns from yellow to deep red, and the form in which it is used routinely for various religious ceremonies. Curcumin is unstable at basic pH, and degrades within 30 min to Trans-6-(40-hydroxy-30-methoxyphenyl)-2, 4-dioxo-5-hexanal, ferulic acid, feruloylmethane and vanillin. Under acidic conditions, the degradation of curcumin is much slower, with less than 20% of total curcumin decomposed at 1 hr. The dried roots of turmeric, which have been used for centuries as a spice (Curry), food preservative and a colouring agent have been found to be a rich sources of phenolic compounds (curcuminoids) with versatile biological mechanisms. The content of total curcuminoids in turmeric powder plays an important role in its antioxidant activity and effectiveness of the product. The content of curcuminoids may vary in turmeric rhizome grown in different agro-climatic zones. Curcumin acts as a free radical scavenger and antioxidant, inhibiting lipid peroxidation and is a well known hepatoprotective protects the liver from damages caused due to alcohol consumption.

<table>
<thead>
<tr>
<th>Name of the constituent</th>
<th>Molecular formula</th>
<th>Molecular mass</th>
</tr>
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<tbody>
<tr>
<td>Curcumin (Diferuloyl methane)</td>
<td>C_{21}H_{20}O_{6}</td>
<td>368</td>
</tr>
<tr>
<td>Demethoxy curcumin (p-Hydroxy-cinnamoyl-feruloyl-methane)</td>
<td>C_{20}H_{18}O_{5}</td>
<td>338</td>
</tr>
<tr>
<td>Bisdemethoxy curcumin (pp’-Dihydroxy-dicinnamoyl-methane)</td>
<td>C_{19}H_{16}O_{4}</td>
<td>308</td>
</tr>
</tbody>
</table>
Curcuminoids are recognized for their broad spectrum of biological activities, the potential use of curcumin in the prevention of cancer is the subject of intensive laboratory and clinical research. Recently it was reported that the effect of curcuminoids was examined on the proliferation of MCF-7 human breast tumour cells that demethoxycurcumin was the best inhibition of MCF-7 cells followed by curcumin and bisdemethoxycurcumin. Many of these properties could be enhanced through improving bioavailability of curcuminoids by different approaches and subject of intensive laboratory and clinical research.

Eventhough, curcumin has more pharmacological properties, the amount of total curcuminoids absorbed by the animal systems far less. Small doses of turmeric (curcumin) are taken daily as a spice by the population in many Asian countries. In one epidemiologic survey, in terms of its dietary use in Nepal, turmeric consumption was found to be up to 1,500 mg per person per day, equivalent to approx. 50 mg/day of curcumin. In India, where the average intake of turmeric can be as high as 2,000–2,500 mg per day (corresponding to approx. up to 100 mg of curcumin), no toxicities or adverse effects have been reported or studied at the population level.

However the doses administered in clinical trials are expected to be rather higher than those normally consumed in the diet. This fact underlines the need for systematic safety and toxicity studies. Based on repeated studies, turmeric is Generally Recognized As Safe (GRAS) by the US FDA, and curcumin has been granted an acceptable daily intake level of 0.1–3 mg/kg-BW by the Joint FAO/WHO Expert Committee on Food Additives, 1996.

In the systematic studies funded by the Prevention Division of the US National Cancer Institute (NCI) conducted in rats, dogs, or monkeys and at oral doses of curcumin up to 3,500 mg/kg-BW for up to 90 days, no adverse effects were observed. In a preclinical study involving the administration of 2% dietary curcumin (approx. 1,200 mg/kg-BW) to rats for 14 days or in a study of 0.2% dietary curcumin (approx. 300 mg/kg-BW) administered to mice for 14 weeks, no toxicity was observed. Furthermore, a reproductive toxicity study with the oral curcumin administration of up to 1,000 mg/kg-BW daily, no toxicity was observed in two successive generations in rats. Curcumin is an approved food colouring agent in addition to a dietary item. In spite of reported minor adverse effects, large doses of up to 12,000 mg per day of curcumin were found to be well tolerated in humans. Therefore, based on the safety and toxicity profile, in several clinical trials the targeted doses for curcumin can be recommended in between 4,000–8,000 mg to obtain the maximum therapeutic effects.

Literature studies have shown that the extraction of curcumin from turmeric could be done in several different ways. The curcuminoids are not water-soluble and therefore extractions have to be made in non-polar solvents. In previous studies hexane, acetone, ethylene dichloride and different alcohols have been used to extract the curcumin analogues. It has been shown that the best yield from dried turmeric was has been obtained from the attempts where the extraction is made by acetone.

Turmeric oleoresin (TO) contains 25-35% curcuminoids and a major portion of the TO prepared in the world has been used to isolate curcuminoids. The essential oil extracted from TO has been reported to have antibacterial, antioxidant and antifungal activities. After the partial separation of curcuminoids, the remaining oleoresin (mother liquor) known as spent oleoresin (STO) is considered as an industrial waste and used as fuel in boilers. Previous studies showed that still some more amount curcumin present in the STO. Spent turmeric oleoresin has anti diabetic and anti oxidant activity.

A number of studies are undertaken to separate curcuminoid pigments by thin layer chromatography (TLC), high-performance thin-layer chromatography (HPTLC), and column chromatography (CC). The stationary phase most commonly used is silica gel with different solvent systems including benzene, ethyl acetate, ethanol, chloroform, acetic acid, hexane, and methanol for chromatographic separations. HPLC method was sensitive, precise and accurate for detection and quantification of curcuminoids in the extract of rhizome Curcuma longa.
Separation by high-performance liquid chromatography (HPLC) was done mostly on reverse phase employing mixtures of water, acetonitrile, ethanol, and methanol. Since the curcuminoid pigments vary in chemical structures, it is possible that the physico-chemical characteristics as well as the functional properties would vary among them. As compounds DMC and BDMC are not commercially available, it could be important to obtain these pigments in high purity for detailed studies on their chemical and physiological attributes. Therefore it is important to obtain pure pigments and characterize them individually to study their biological properties.

The present study describes isolation, identification and purification of curcuminoids from spent turmeric oleoresin (industrial waste) by column chromatography followed by purity analysis by HPLC.

MATERIALS AND METHODS:

Year of experimentation: 2013

Site of experimentation: Research and development laboratory AVT Natural products, Vazhakulam, Kerala (India).

Chemicals, reagents and samples: The spent turmeric oleoresin (STO), after commercial isolation of curcuminoids from turmeric oleoresin, was procured from 3 different batches. Diaion HP 20 Resin purchased from Mitsubishi chemical, Japan and Silica gel 60-120 mesh from merck. All solvents / chemicals used were of AR/HPLC grade and obtained from Merck (Mumbai, India). The reference curcuminoids purchased from Chromadex company USA.

A. Spectrophotometric analysis:

a. Instrumentation: Spectroscopic evaluation was carried out by using Varian Cary 50 Bio UV/Visible Spectrophotometer.

b. Sample preparation: Solutions of individual curcuminoids containing 5 µg/ml were prepared in methanol. UV spectrum scan was done.

B. Separation of curcuminoids by TLC:
Samples were dissolved in methanol. Methanol extracts were tested on TLC pre-coated Silicagel (Merk-60 F254.0.25mm thick) plates. Plate were developed using TLC chamber, which was pre-saturated with mobile phase for 20 min and each plate was developed up to a height of about 9.5 cm. Chloroform: Ethanol: glacial acetic acid mobile phase was used with composition 94:5:1. After development, plates were removed and dried. Spots were analyzed and Rf value noted.

C. Estimation of curcuminoids by HPLC:

1) Preparation of standard: Weighed accurately 10 mg of standard curcuminoids and dissolved in 25 ml methanol. Pipetted out 1 ml and diluted to 5 ml with methanol. Filtered through 0.2 µm membrane filter before injection.

2) Preparation of Sample: Weighed accurately 25 mg sample and dissolved in 25 ml methanol. Pipetted out 1 ml and diluted to 5 ml with methanol. Filtered through 0.2 µm membrane filter before injection.

3) Chromatographic conditions: Samples were analysed by HPLC in a Shimadzu LC 30AD liquid chromatography system with SPD M-20A PDA detector in isocratic mode. The instrument is equipped with an auto sampler. 20 µl of sample was injected with a flow rate of 1 ml/ min at 25°C. Column used was Supelco Discovery HS C 18 250X4.6mm 5 µm particle size. Mobile phase used was Acetonitrile, Water, Orthophosphoric acid in proportion of (50:50:0.1) v/v.

4) Preparation of mobile phase: Mobile phase was prepared by mixing 500 ml acetonitrile (HPLC grade), 500 ml water (HPLC grade) and 0.5 ml orthophosphoric acid. Above mixture (50:50:0.1) was degassed in an ultrasonic water bath for 5 minutes and filtered through 0.45 µ filter under vacuum.
D. COLUMN CHROMATOGRAPHY

1. Diaion HP 20 Resin Activation: 100gm of Diaion HP 20 Resin pretreated overnight with 400ml of ethanol. It was then filled to the glass column and washed with 2BV of 80 % Methanol.

2. Column Chromatography for preparation of Curcuminoid Enriched Fraction using Diaion HP 20 resin: 25 gram of Spent Turmeric oleoresin composing of curcumin, demethoxy-curcumin and bis demethoxycurcumin was dissolved in 50ml of 80% Methanol. It was then loaded to the pre treated Diaion HP 20 Resin (Mitsubishi Chemical). Column eluted with 80% methanol and 10 fractions collected. The total curcuminoid content of the fractions were analyzed by HPLC at 420 nm. Curcuminoid enriched fractions combined & concentrated. Concentrated fractions, dissolved in 10ml methanol and precipitated in excess hexane. Precipitated product filtered & dried to yield curcuminoid enriched fraction in powder form with 50% purity by HPLC.

3. Silica gel Column chromatography for separation of individual curcuminoids: About 7.75gm of Curcuminoid enriched fraction was mixed with 15 gm of silica gel and loaded to the column containing 100gm silica gel 60-120 mesh of 46x2 cm and eluted with chloroform followed by chloroform:methanol with increasing polarity. All the column fractions were subjected to TLC silica gel (Merk-60 F254,0.25mm thick ) plate using Chloroform: Ethanol: glacial acetic acid (94:5:1) as the developing solvent system and detected as yellow spots. Fractions with the same Rf values were pooled and the organic solvent was removed by distillation under reduced pressure.

E. Purification of individual curcuminoids: The individual curcuminoids isolated by column chromatography was further purified by recrystallization using Methanol : Chloroform in the ratio 5:1 at 3°C. The crystals obtained were separated by filtration. The purity of Individual crystals were analysed by HPLC.

F. Melting point analysis: Melting point of individual curcuminoids was analysed by using Veego Digital melting point apparatus.

RESULTS AND DISCUSSIONS: Curcuminoids have immense biological properties in which curcumin (C) is reported for most of the medicinal properties. It has been used to cure various ailments including cancer, Parkinson, Alzheimer and HIV. Recently the analogs of curcumin were reported for its biological activities, Demethoxycurcumin (DMC) showed the best inhibition of MCF-7 cells. Bisdemethoxycurcumin (BDMC) was active for modulation of MDR-1 gene expression. These compounds are not commercially available due to the difficulty in fractionation. Therefore to study biological properties of individual curcuminoids we need to isolate compounds at high purity. Our present study throws light for obtaining individual curcuminoids with high purity from spent turmeric oleoresin which is considered as Industrial waste in curcumin production industry.

The HPLC profile of spent turmeric oleoresin (STO) taken for isolation had 18% total curcuminoids, of which curcumin (C), and its analogues DMC and BDMC are 4.4%, 4.9% and 8.7% respectively. Curcuminoid in the spent turmeric oleoresin enriched from 18% to 34.1% by purification using Diaion HP 20 as adsorbent and 80% methanol as eluent. The curcuminoid content of eluted fractions are shown in table 2. Based on the HPLC profile the eluted fractions from 3 to 8 was combined and concentrated to obtain curcuminoid enriched oleoresin. The HPLC profile of this curcuminoid enriched oleoresin contains the total curcuminoid of 34.1% with curcumin 9.6%, DMC 9.2% and 15.3% BDMC. This was further purified by dissolving the curcuminoid enriched oleoresin in methanol and precipitated in excess hexane.

Hexane precipitated product filtered and dried, to obtain curcuminoid enriched fraction with a yield of 31% with respect to spent turmeric oleoresin.
The HPLC profile of the curcuminoid enriched fraction shown in fig. 2 contains total curcuminoids of 50% with curcumin 11%, DMC 13%, BDMC 26%. This purified fraction has a composition different from curcuminoids in turmeric, oleoresin turmeric, and commercially available 95% curcumin powder. The concentration of DMC, BDMC is much higher compared to curcumin in curcuminoid enriched fraction isolated from spent turmeric oleoresin. The ratio of commercially obtained curcuminoids is C:D:B is 7:1.4:0.3 to 7.7 : 1.8 : 0.5 and the curcuminoids obtained from spent turmeric oleoresin is 2.2:2.6:5.2.

Seperation of individual curcuminoids was done using column chromatography over silica gel and chloroform : methanol as eluent. 40 fractions were collected. Fractions with same Rf values combined and concentrated. The HPLC profile of pooled fractions recorded. The silica gel column chromatography elution profile is shown in table 3.

These fractions on recrystallization gives Curcumin, DMC, BDMC crystals having purities 98.4%, 97.1%, 97.3% by HPLC with an yield of 2.8%, 3.6%, 7.2% with respect to the spent turmeric oleoresin. Melting point obtained for recrystallized material of curcumin 183°C, DMC 172°C, BDMC 222°C respectively. By TLC Curcumin, DMC, BDMC has 0.70, 0.40 & 0.27 Rf values shown in fig. 3 and UV absorption maximum has 425nm, 420nm, 416nm respectively was given in fig. 4. Properties of recrystallised curcuminoids summarized in table 4.
FIG. 2: HPLC PROFILE OF A, SPENT TURMERIC OLEORESIN B, CURCUMINOID ENRICHED FRACTION C, CURCUMIN CRYSTALLISED D, DEMETHOXY CURCUMIN CRYSTALLISED E, BISDEMETHOXY CURCUMIN CRYSTALLISED
CONCLUSION: Demethoxy curcumin and Bis demethoxy curcumin in pure form commercially not available. A method developed for separation of 3 curcuminoids from spent turmeric oleoresin - which is now considered as an industrial waste in curcumin production industry. The new process developed is simple and economical to adopt.

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REFERENCES:


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