(Research Article)

IJPSR (2013), Vol. 4, Issue 11







ENHANCEMENT OF DISSOLUTION RATE OF CIPROFLOXACIN BY USING VARIOUS SOLID DISPERSION TECHNIQUES

Brahmaiah Bonthagarala*, Leela Madhuri Pola and Sreekanth Nama

Department of Pharmaceutics, Priyadarshini Institute of Pharmaceutical Education & Research (PIPER), 5th Mile, Pulladigunta, Kornepadu (V), Vatticherukuru (M), Guntur-522017, Andhra Pradesh, India

Keywords:

Solid dispersions, Solubility, Dissolution rate, Ciprofloxacin, Crosscarmellose sodium

Correspondence to Author:

Brahmaiah Bonthagarala

Assistant Professor, Department of Pharmaceutics, Priyadarshini Institute of Pharmaceutical Education & Research, 5th Mile, Pulladigunta, Kornepadu (V), Guntur-522017, Andhra Pradesh, India

E-mail: brahmaiahmph@gmail.com

INTRODUCTION: The poor solubility and low dissolution rate of poorly water soluble drugs in the aqueous gastro-intestinal fluids often cause in sufficient bioavailability rather than the limited permeation through the epithelia and the formulation of poorly soluble drugs for oral delivery now presents one of the major challenges tom formulation scientists in the industries ¹. The term solubility is defined as maximum amount of solute that can be dissolved in a given amount of Solvent quantitatively it is defined as the concentration of the solute in saturated solution at a certain temperature in qualitative terms, solubility may be defined as spontaneous interaction of two or more substances to form a homogenous molecular dispersion ².



ABSTRACT: The aim of this research wok is to formulate and evaluate Ciprofloxacin solid dispersions system by using the different techniques. This will increase the solubility of the drug or Ciprofloxacin and give the immediate release of the drug from the formulations. The main objective is to formulate a drug product as immediate release oral solid dosage form of Ciprofloxacin solid dispersion system which is considered to be stable, robust quality and enhanced dissolution rate. To optimize the method of manufacture by formulate the Ciprofloxacin solid dispersion system by various techniques like Physical mixing, Co-grinding, Kneading and solvent evaporation techniques. The disintegrating agent used in the present study is Crosscarmellose sodium. Among the four different techniques used for preparation of solid dispersions solvent evaporation technique has shown the increase in dissolution rate that is the Trail-5 was found to have a faster solubility and dissolution property which was prepared by using Crosscarmellose sodium as a disintegrant in the ratio of 1:1.

> **Mechanism of action:** Ciprofloxacin is a semisynthetic (partially man-made), oral antibiotic in the cephalosporin family of antibiotics that stops bacteria from multiplying by preventing bacteria from forming the walls that surround them and it inhibit the cell wall synthesis. The walls are necessary to protect bacteria from their environment and to keep the contents of the bacterial cell together; bacteria cannot survive without a cell wall.

> **Solid Dispersions:** The term solid dispersion refers to a group of solid products consisting of at least two different components, generally a hydrophilic matrix and a hydrophobic drug³. The matrix can be either crystalline or amorphous. The drug can be dispersed molecularly, in amorphous particles (clusters) or in crystalline particles ⁴. Therefore, based on their molecular arrangement, six different types of solid dispersions can be distinguished. They are described in **Table 2**. Moreover, certain combinations can be encountered, i.e. in the same

International Journal of Pharmaceutical Sciences and Research

sample; some molecules are present in clusters while some are molecularly dispersed ⁵. Moreover, not the preparation method but the molecular arrangement governs the properties of solid dispersions 6 .

Advantages of Solid Dispersion: Particles with reduced particle size, Particles with improved wettability, particles with higher porosity, drugs in amorphous state ⁷.

Preparation methods of Solid Dispersions⁸:

- (a) Solid dispersion techniques.
- (b) Solvent evaporation method.
- (c) Fusion method/melting method.
- (d) Hot melt extrusion.
- (e) Supercritical fluid technology (SCF).
- (f) Dropping method.
- (g) Electrostatic Spinning Method.
- (h) Co-precipitation method.

MATERIALS AND METHODS:

Materials used: Ciprofloxacin (Pellets Pharma Ltd), Crosscarmellose Sodium (Diocon Pharma Ltd), Distilled Water, Methanol, Di - Chloro methane, Potassium Di-hydrogen Phosphate, Sodium Hydroxide.

Instruments used: Digital Balance – Semsung AN ISO 9001-2004, Tray Dryer – ELITE scientific, Dissolution apparatus – VEEGO, U-V spectrophotometer – Spectro 2080 plus Analytical technologies limited, Distillation Tank – JSGW double distillation plant.

Experimental methods:

Estimation of Ciprofloxacin: A Simple Sensitive and accurate Spectrophotometric method was used for the measurement of Ciprofloxacin at a λ_{max} 288nm. The absorbance of standard dilutions was measured at 288nm.

Procedure: The standard solutions of Ciprofloxacin were subsequently diluted with pH 7.2 Phosphate buffer to obtain series of dilutions containing 5, 10, 15, 20 and 30 µg of Ciprofloxacin solution⁹. The absorbances of above dilutions were measured in Spectro 2080 plus Analytical technologies limited U-V Spectrophotometer at 288 nm using distilled water as blank. The concentration and the corresponding absorbance values were given in Table 5. The absorbance values were plotted against concentration of Ciprofloxacin as shown in Figure 2.

Drug release studies: Dissolution studies on each formulation were performed in a calibrated eight station dissolution test apparatus equipped with paddles employing pH 7.2 phosphate buffers as medium. The paddles were operated to rotate at 75 rpm and the temperature was maintained at $37\pm$ 1°C throughout the experiment. Samples were withdrawn at regular intervals up to 60 min and replaced with equal volume to maintain the constant volume of dissolution medium throughout the experiment ¹⁰.

Drug content of the samples was determined by Spectro 2080 plus Analytical technologies ltd UV Spectrophotometer at 288nm after suitable dilution of samples ¹¹. Necessary corrections were made for the loss of drug due to each sampling. The drug dissolved experiments were conducted in triplicate. The dissolution profile were depicted in **Table 12 to 13** and shown in **Fig. 13**. The zero order and first order plots and their corresponding release constant were given in **Table 6 to 10** and shown in **Figure 3 to 12**.



FIG 1: DISSOLUTION APPARATUS

Dissolution Test Acceptance Criteria: Acceptance criteria for dissolution tests are set on the basis of requirements for a percent quantity of drug to be released after a certain period of time in the dissolution apparatus. Since each test is usually conducted using six individual dosage units, acceptance criteria must be established on this basis ¹¹. The compendia recognize that reasonable tolerances are required to ensure that criteria are not prohibitive, but at the same time, they need to discriminate between acceptable and unsatisfactory batches of products.

The approach developed within various compendia is now harmonized for immediate release and extended release-though not for delayed releaseand allows for three levels of staged testing. For immediate release products, acceptance criteria are based on a single time point and a single value, expressed as a Q value. Then, at each of the three stages, the specification requires that mean values not be less than Q, but a set number of individual units are allowed to release a percent quantity of active which may be as low as Q - 25% for one unit at stage three ¹².

For extended release products, specifications are based on three or four time points. For the intermediate time points, the requirements are based on ranges; for the final time point, they are usually based on a single value. Therefore, the acceptance criteria at each stage are expressed in terms of variances around ranges for intermediate time points and minimum acceptable release at the final time point ¹³.

Determination of amount of drug content: The amount of drug substance present in the given dissolution samples were determined with the help of UV Spectrophotometer at a wavelength of 288nm. The obtained absorbance values were substituted in the given equation to get the amount of drug substance at different time intervals of dissolution.

Amount=

Test o.d.

Standard o.d.

X Dilution factor ×900 /1000

Procedure for Assay: The amounts of drug present in the taken oven formulations were determined with the help of U-V Spectrophotometric method. For each batch 100 mg of sample was taken in to the volumetric flask and add methanol and the mixer was diluted with the pH 7.2 phosphate buffer. The dilutions are made as 5, 10, 15, 20 and 25 µg/ml. The absorbance of solution determined was at 288 nm by UV Spectrophotometer. The resultant drug content and practical yield was shown in Table 11.

Different formulation batches:

Formulation 1 - Pure Drug (100mg)

Formulation 2 – Drug: Polymer (1:1) Crosscarmellose sodium

TABLE 1: FORMULATION 2 (Preparation of Soliddispersion by physical mixing Method).

S. No.	Ingredients	Quantity
1	Ciprofloxacin	100mg
2	Crosscarmellose sodium	100mg

Physical Mixing: Accurately weighed required amount of Ciprofloxacin and Crosscarmellose Sodium (carrier) in 1:1 drug-to-carrier weight ratio were mixing thoroughly in a mortar until a homogeneous mixture was obtained for 3 min. The product was kept in desiccators at room temperature until for further study or investigation.

Formulation 3 - Drug: Polymer (1:1) Crosscarmellose sodium

TABLE 2: FORMULATION 3 (Preparation of Soliddispersion by Co-grinding technique)

S. No.	Ingredients	Quantity
1	Ciprofloxacin	100mg
2	Crosscarmellose sodium	100mg

Co-grinding Technique: Required quantity of drug was accurately weighed and transferred in to mortar to this adds requires quantity of Crosscarmellose sodium in the ration of 1:1 were mixing thoroughly until a homogenous mixer was obtained. Triturating was carried in 10-15 min to form a homogenous mixer. The product is packed and kept in a dedicator at room temperature until for further study or investigation.

Formulation 4 - Drug: Polymer (1:1) Crosscarmellose sodium

S. No.	Ingredients	Quantity
1	Ciprofloxacin	100mg
2	Crosscarmellose sodium	100mg
3	Distilled Water	Quantity sufficient

Kneading method: Drug and polymer was mixed with the small amount of the solvent i.e. water to form a thick paste by kneading and hence it was dried at 45°C in a tray dryer. The mass was passed through the sieve no. 30 and stored in the desiccators. The product is packed and kept in a dedicator at room temperature until for further study or investigation.

Formulation 5 – Drug: Polymer (1:1) Crosscarmellose sodium

TABLE 4: FORMULATION 5 (Preparation of Soliddispersion by solvent evaporation method)

S. No.	Ingredients	Quantity
1	Ciprofloxacin	100mg
2	Crosscarmellose sodium	100mg
3	Dichloromethane	Quantity sufficient

Solvent evaporation method: Required amount of drug is accurately weighed and transfer in to mortar. The drug and the polymer were dissolved in sufficient volume of dichloromethane with continuous stirring. The solvent was then completely evaporated at room temperature with continuous stirring to obtain dry granules.

The resulting solid dispersion was stored in airtight container till further use.

RESULTS AND DISCUSSION:

TABLE-5:	Calibration	Curve	for	Ciprofloxacin	in
methanol (N	[=6)				

S. No.	Concentration (µgm/ml)	Absorbance
1	0	0
2	5	0.106
3	10	0.209
4	15	0.136
5	20	0.415
6	25	0.52



ESTIMATION OF CIPROFLOXACIN

TABLE 6: KINETIC PROFILE OF CIPROFLOXACIN F-1(PURE DRUG)

S. No.	Time	OD	Amount of drug dissolved	%Drug dissolved	%Drug remained	Log% Drug remained
1	0	0	0	0	100	2
2	5	0.195	83.97	84.78	15.22	1.182
3	10	0.203	81.41	88.25	11.75	1.070
4	15	0.213	91.72	83.34	16.66	1.221
5	20	0.227	97.75	98.69	1.31	0.117

TABLE 7: KINETIC PROFILE OF CIPROPLOXACIN F-2 (PHYSICAL MIXING) SOLID DISPERSION

S. No.	Time	OD	Amount of drug dissolved	%Drug dissolved	%Drug remained	Log% Drug remained
1	0	0	0	0	100	2
2	5	0.206	88.70	89.17	10.83	1.034
3	10	0.219	94.30	94.80	5.2	0.716
4	15	0.222	100	-	-	-

TABLE 8: KINETIC PROFILE OF CIPROFLOXACIN F-3 (COGRINDING) SOLID DISPERSION

S. No.	Time	OD	Amount of drug dissolved	%Drug dissolved	%Drug remained	Log% Drug remained
1	0	0	0	0	100	2
2	5	0.213	88.70	74.89	25.11	1.4
3	10	0.232	97.38	91.72	8.28	0.918
4	15	0.230	99.04	100	0	-

 TABLE 9: KINETIC PROFILE OF CIPROFLOXACIN F-4 (KNEADING METHOD) SOLID DISPERSION

S. No.	Time	OD	Amount of drug dissolved	%Drug dissolved	%Drug remained	Log% Drug remained
1	0	0	0	0	100	2
2	5	0.199	85.69	85.77	14.23	1.153
3	10	0.210	90.43	90.52	9.48	0.976
4	15	0.226	97.32	100	2.59	0.413

TABLE 10: KINETIC PROFILE OF CIPROFLOXACIN F-5 (SOLVENT EVAPORATION METHOD) SOLID DISPERSION

S. No.	Time	OD	Amount of drug dissolved	%Drug dissolved	%Drug remained	Log% Drug remained
1	0	0	0	0	100	2
2	5	0.183	78.80	94.24	25.70	1.41
3	10	0.201	96.52	100	5.675	0.754

TABLE 11: ASSAY OF CIPROFLOXACIN

S. No.	Formulation	Drug content	Practical yield
1	F1	94 mg	93
2	F2	89.8 mg	91.6
3	F3	90.72	91.03
4	F4	90.28	91.33
5	F5	93.31	94.18



FIGURE 3: ZERO ORDER KINETIC PROFILE OF CIPROFLOXACIN F-1(PURE DRUG)



CIPROFLOXACIN F-2(PHYSICAL MIXING) SOLID DISPERSION



FIGURE 5: ZERO ORDER KINETIC PROFILE OF CIPROFLOXACIN F-4 (KNEADING METHOD) SOLID DISPERSION



FIGUER 6: ZERO ORDER KINETIC PROFILE OF CIPROFLOXACIN F-3 (COGRINDING) SOLID DISPERSION



FIGURE 7: ZERO ORDER KINETIC PROFILE OF CIPROFLOXACIN F-5 (SOLVENT EVAPORATION METHOD) SOLID DISPERSION



FIGURE 8: FIRST ORDER KINETIC PROFILE OF CIPROFLOXACIN F-1(PURE DRUG)



CIPROFLOXACIN F-2(PHYSICAL MIXING) SOLID DISPERSION





FIGURE 10: FIRST ORDER KINETIC PROFILE OF CIPROFLOXACIN F-3 (COGRINDING) SOLID DISPERSION



FIGURE 11: FIRST ORDER KINETIC PROFILE OF CIPROFLOXACIN F-4 (KNEADING METHOD) SOLID DISPERSION



FIGURE 12: FIRST ORDER KINETIC PROFILE OF CIPROFLOXACIN F-5 (SOLVENT EVAPORATION METHOD) SOLID DISPERSION



FIGURE-13- COMPARATIVE FIRST ORDER DISSOLUTION PROFILE ALL FORMULATIONS

TABLE 12: DISSOLUTION PARAMETERS



FIGURE 14: COMPARITIVE DISSOLUTION PROFILE OF ALL FORMULATIONS

S. No	Trails	Zero Order			First Order			
		PD10 (%)	T50 (min)	Regression (R)	Ko (mg/ml)	Slope (b)	R	K1 (min ⁻¹)
1	F-1	88.25	4	0.898	1.88	0.0917	0.9942	0.087
2	F-2	94.8	3.3	0.982	4.79	0.128	0.9608	0.022
3	F-3	100	3	0.992	2.73	0.108	0.998	0.077
4	F-4	90.52	3.5	0.899	0.762	0.0987	0.9702	0.126
5	F-5	100	3	0.991	10	0.1246	0.9995	0.55

TABLE 13: DSSOLUTION DATA:

S No	Time (min)	% Drug Dissolved						
5. INU.	Time (mm)	F1	F2	F3	F4	F5		
1	0	0	0	0	0	0		
2	5	83.97	88.70	74.89	85.77	94.24		
3	10	81.41	94.30	91.72	90.52	100		
4	15	91.72	100	100	100	-		
5	20	97.75	-	-	-	-		
6	30	100	-	-	-	-		

DISCUSSION:

Trail-1: This trail was done by using pure drug; it has shown poor dissolution property because of less solubility of drug in the dissolution medium.

Improvisation: As pure drug has shown poor dissolution property, in order to enhance the dissolution rate we used different methods for preparing solid dispersions.

Trail -2: In this trail solid dispersion of drug was prepared by using Crosscarmellose sodium as a disintegrant in the ratio of 1:1 and by using Physical mixing method. Maximum drug release was observed at time 15 min.

Improvisation: Further, this formula was optimized for enhancing the drug release by using other technique.

Trail-3: In this trail solid dispersion of drug was prepared by using Crosscarmellose sodium as a disintegrant in the ratio of 1:1 and by using Co-grinding method. Maximum drug release was observed at time 15 min.

Improvisation: Further this formula was further optimized for enhancing the drug release by using other technique.

Trail-4: In this trail solid dispersion of drug was prepared by using Crosscarmellose sodium as a disintegrant in the ratio of 1:1 and by using kneading method. Maximum drug release was observed at time 20 min.

Improvisation: Further this formula was further optimized for enhancing the drug release by using other technique.

Trail-5: In this trail solid dispersion of drug was prepared by using Crosscarmellose sodium as a disintegrant in the ratio of 1:1 and by using Solvent evaporation method. Maximum drug release was observed at time 10 min.

Hence, the **trial 5** which was prepared by using Crosscarmellose sodium and drug by the ratio of 1:1 by using solvent evaporation was found to be optimized formula.

SUMMARY AND CONCLUSION: This study was undertaken with an aim to formulate an Anti-Biotic drug in the form solid dispersion to overcome the poor solubility drawback of the drug. The selected Antibiotic agent was Ciprofloxacin. The drug Ciprofloxacin is having poor solubility in the water, under class 2 of BCS of classification of drug its solubility was tried to increase by formulating in the form of solid dispersion with polymer by using various techniques. Solid were prepared dispersions by using the Crosscarmellose sodium as a disintegrant in 1:1 ratio of different techniques.

Among the four different techniques used for preparation of solid dispersions solvent evaporation technique has shown the increase in dissolution rate that is the Trail-5 was found to have a faster solubility and dissolution property which was prepared by using Crosscarmellose sodium as a disintegrant in the ratio of 1:1. Hence finally it was concluded that **Trail-5** as an optimized formula with an increased rate of dissolution rate and solubility. Trail 5 which is prepared by using drug and disintegrant ratio of 1:1 ratio by using solvent evaporation techniques.

ACKNOWLDGEMENT: The authors greatly acknowledge Director and Principal of Priyadarshini Institute of Pharmaceutical Education and Research (PIPER), Guntur, Andhra Pradesh and hearty thanks to my principal Sreekanth Nama, for providing the facilities to carry out the research work.

REFERENCES:

- Yogesh Thorat S, Indrajeet Gonjari D and Avinash Hosmani H: Solubility enhancement techniques: A Review on conventional and novel approaches. IJPSR 2011; 2(10): 2501-2513.
- Vijay Kumar Nagabandi, Ramarao, Jayaveera KN: Liquidsolid Compacts: A novel approach to enhance bioavailability of poorly soluble drugs. International Journal of Pharmacy and Biological Sciences 2011; 3: 89-102.
- 3. Lennernas H, Shah VP and Crison JR: Theoretical basis for a biopharmaceutical drug classification: The correlation of in vitro drug product dissolution and *in vivo* bioavailability. Pharm Research 1995; 12 (3): 413-420.
- 4. Breitenbach J: Melt extrusion from process to drug delivery system: Pharm Research 2002; 12(1): 1269-1292.
- Broman E, Khoo C, and Taylor LS: A Comparison of alternative polymer excipients and processing methods for making solid dispersion of poorly water soluble drugs: International Journal of Pharma Research 2011; 22(1): 139-151.
- 6. Buck ton G and Darcy P: The use of gravimetric studies to asserts the degree of crystalanity of predominantly crystalline powder: International of Journal of Pharma Research 2012; 165-171.
- Javadzadeh Y, Jafari-Navimipour B, Nokhodchi A: Liquisolid technique for dissolution rate enhancement of a high dose water-insoluble drug (carbamazepine). International Journal of Pharmaceutics 2007; 341:26-34.
- 8. Ford JL: The current status of solid dispersions. Pharm Acta 2006; 69-88.
- A.M Woodlinger: Cardiovascular in: D.B troy (Ed), Remington the science & practice of pharmacy, Lippincott Williams & Wilkins, Philadelphia 2005.
- Karavas E, Ktistis G, Xenakis A and Georgarakis E: Effect of hydrogen bonding interactions on the release mechanism of felodipine from nanodispersions with polyvinylpyrrolidone. European Journal of Biopharmaceutics 2006; 63(3): 103-114.
- Leuner C and Dress man J: Improving drug solubility for oral delivery using solid dispersions. European journal of Biopharmaceutics 2010, 50(3): 47-60.
- 12. Sahil Gavali M, Sharad Pacharane S, Kisan Jadhav R and Vilasrao Kadam J: Liquidsolid compact: A new technique for enhancement of drug dissolution. International Journal of Research in Pharmacy and Chemistry 2011; 1(3): 705-713.
- 13. Christina Hentzsachel Dissertation on Optimization of the Liquisolid technology identification of highly effective tableting excipients for liquid adsorption, 2011.
- S C Arora: Development, characterization and and solubility and solid dispersions of Ciprofloxacin by solvent method. International Journal of Drug Discovery and Research 2010, 424-430.
- Yogesh Pore V, Snehal Mulye P, Samina Jamadar A, Poonam Karekar S, Shashikant Dhawale C: Improvement in physicochemical properties of Ezetimibe using a crystal engineering technique. Powder Technology 2012; 222: 131– 138.

How to cite this article:

Bonthagarala B, Pola LM and Nama S: Enhancement of dissolution rate of Ciprofloxacin by using various Solid Dispersion techniques. *Int J Pharm Sci Res* 2013; 4(11): 4376-83. doi: 10.13040/IJPSR. 0975-8232.4(11).4376-83

All © 2013 are reserved by International Journal of Pharmaceutical Sciences and Research. This Journal licensed under a Creative Commons Attribution-NonCommercial-ShareAlike 3.0 Unported License.

This article can be downloaded to **ANDROID OS** based mobile. Scan QR Code using Code/Bar Scanner from your mobile. (Scanners are available on Google Playstore)