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# ANALYTICAL METHOD DEVELOPMENT AND ITS VALIDATION FOR THE ESTIMATIONOF THYMOL IN TACHYSPERMUM AMMI MILL FRUIT BY GAS LIQUID CHROMATOGRAPHY

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### **Keywords:**

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# **ABSTRACT**

Thymol is the main volatile compound extracted from dried fruits of Tachyspermum ammi Mill. The compound serves as a biological marker because of its interesting biological activities and is used in traditional medicine, as a bactericide, fungicide, Gastro-intestinal disorders, Bronchitis and asthma, and others. A GC method for the determination of thymol was developed. Chromatographic analysis was carried out on - stainless 30m×0.25m×0.25mm packed with polysiloxane column with flow rate 1.0 ml per minute of the carrier gas. Quantification was performed using a FID. The method was validated for specificity, precision, Linearity, accuracy and robustness. The method was found to be precise for different concentrations of thymol. Accuracy was checked by conducting recovery at different level of thymol and the average percentage and recovery was found to be in the range 90%-110%. The Linearity was established over a range of seven different concentrations of the analyte (70%, 80%, 90%, 100%, 110%, 120% and 130% of test concentration) with correlation coefficients of 0.9997 for thymol.

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INTRODUCTION: Herbal medicines or phytomedicines are medicinal products containing the active ingredients of exclusively plant origin have been used in medical practices since antiquity as the major remedy in traditional system of medicine <sup>1</sup>. The history of herbal medicines is as old as human civilization. The practices continue even today, because of there is much disillusionment of public with the conventional medicines due to its biomedical benefits as well as in culture beliefs that the herbal remedies are safer "more natural" and have made a great contribution towards maintaining human health 9.

Quality assurance is thrust area for the evaluation of traditional used medicinal plants and herbal formulations. Hence, it has become necessary tool to standardize the safety and quality assurance measures as to ensure the supply of medicinal plant materials of good quality. However, it may not be possible to apply the standardization parameters for herbal medicines as applicable to the modern synthetic drugs due to the complexity in their chemical composition. However, recent advances in analytical tool may be helpful to assure the safety, quality and efficacy of herbal drugs <sup>9</sup>.

There is also a primary need to increase the product quality by using the validated test methods. Analytical method development and validation play important role in discovery, development and manufacture of herbal medicinal products. The results from the validated test methods are used to ensure identity, purity, potency and performance of drug products. Healing power & curative properties<sup>6</sup>, Gastro-intestinal disorders <sup>8</sup>, the oil extracted from seeds is beneficial in the treatment of rheumatic and neuralgic pains. It should be applied on the affected parts <sup>7</sup>. The fruits are ovaloblong, greenish brown to yellowish brown in color. They have an aromatic characteristic odor and the taste is sweet aromatic. The fruit consists of two portions each called mericarp and connected by central stalk (carpophore). A single seed is seen in each mericarp. Fruit surface is glabrous and the five primary ridges of each mericarp are prominent, straight and pale straw coloured <sup>4</sup>.Ajwain essential oil showed the presence of 26 identified components which account for 96.3% of the total amount. Thymol (39.1%) was found as a major component along with p-cymene (30.8%),  $\gamma$ -terpinene (23.2%),  $\beta$ pinene (1.7%), terpinene-4-ol (0.8%) whereas acetone extract of ajwain showed the presence of 18 identified components which account for 68.8% of the total amount followed by oleic acid (10.4%), linoleic acid (9.6%), palmitic acid (1.6%), and xylene  $(0.1\%)^4$ .

CH<sub>3</sub>

$$CH_2$$

$$CH_3$$

$$CH_3$$

$$CH_3$$

$$CAMPHENE$$

$$H_2C CH_3$$

$$H_2C CH_3$$

$$CH_2$$

$$H_2C CH_3$$

$$CH_2$$

$$CH_2$$

$$CH_3$$

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$$CH_2$$

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$$CH_2$$

$$CH_3$$

$$CH_3$$

$$CH_2$$

$$CH_3$$

$$CH_$$

# **MATERIAL AND METHODS:**

# Instrumentation & Chromatographic conditions:

Experiments were performed on a GC system (Agilant) equipped with FID detector. The separations were performed on column a stainless steel 30m×0.25m×0.25mm packed with polysiloxane column. column Temperature of 70°C for 1minute, then increased to 100° at a

rate of 10° per minute, then increased to 200° at a rate of 20° per minute and maintained at this temperature for 2 minutes, inlet port temprature of 220°C and flow rate 1.0 ml per minute of the carrier gas (hydrogen, nitrogen and inert gas) and Sample volume of  $1\mu l^2$ .

Herbs, Chemicals and Reagents: Crude drugs were procured from local market and identification was confirmed by macroscopic and microscopic characters. All the herbs procured from the local market. All the chemicals and solvents used were of AR grade; Standard Thymol (99.5%) of Merck grade and acetone of HPLC grade was procured used for the sample preparation.

# Preparation of the formulations:

Preparation of reference solution of thymol: Accurately weighed thymol (30 mg) was transferred in 100 ml volumetric flask and dissolved in and diluted to 100 ml with acetone. The final solution contained 300 ② g of the Thymol per ml of the solution  $^2$ .

Sample preparation: Accurately weighed 2 gm of powdered *Tachyspermum ammi* Mill was taken and refluxed with 100 ml of Acetone for 1 hour. The extract was filtered and made up the volume up to 100 ml with acetone and sample was taken after filtered through Teflon filters of  $0.45\mu$  using 5ml syringe. Each of the solutions was subjected to GC and the area under the peak of thymol was recorded. The amount of thymol was calculated in the test material using the formula as

# %Thymol =

 $\frac{\text{Average area of sample x Conc. of standard}}{\text{Average area of std x Conc. of sample}} \times \text{Potency}$ 

#### **RESULTS AND DISCUSSION:**

Optimization of GC Condition: For giving the most chemical information and better separation in the chromatograms, the column, mobile phase, detection wavelength and conditions for gradient elution were investigated in this study. Two kinds columns as a stainless steel 30m×0.25m×0.25mm packed with polysiloxane column and a stainless steel 30m×0.53mm capillary column was investigated then stainless steel 30m×0.25m×0.25mm packed polysiloxane column was found to be more suitable and gave good peak separation and sharp peaks (Fig. 1).

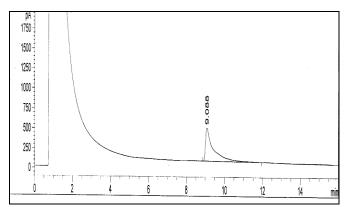


FIG. 1: GC CHROMATOGRAM OF STANDARD THYMOL

Method validation of quantitative analysis: The method was validated in terms of system precision, method precision linearity, repeatability and recovery test.

**Precision:** Precision of an analytical procedure as an expression of closeness of agreement (degree of scatter) between a series of measurement obtained from multiple sampling of the same homogeneous samples under the prescribed condition. Data shown in table 1 system precision data for thymol & table 2 shows method precision data for thymol and this indicate an acceptable level of precision for analytical system. (Acceptance criteria: %RSD should not be more than 2.0 %)

**TABLE 1: SYSTEM PRECISION DATA FOR THYMOL** 

Standard	Peak Area	Average	Standard	RSD
Injection	Peak Alea	Area	Deviation	(%)
1	11963.8			
2	11954.7			
3	12083.6	12093.8	121.26	1.00
4	12173.7			
5	11972.9			
6	12281.8			

**TABLE 2: METHOD PRECISION DATA FOR THYMOL** 

Sample Injection (µI)	Peak Area	Percentage Content	Average Content	Standard Deviation	RSD (%)
1	5044.90	0.76			
2	5147.90	0.76			
3	5080.55	0.76	0.75	0.1	1.97
4	4944.50	0.75			
5	4967.14	0.76			
6	4602.19	0.72			

**Calibration curve and Linearity:** Good linearity response for thymol was obtained with the developed method within the range of 93.67-

133.8 ppm as shown in fig. 2. Linear plot represented graphically indicated that the response is linear over the specified range and the value of correlation coefficient was found to be equal to 0.993. The quantification of thymol undertaken by the proposed method was proved to be simple, sensitive and reproducible.

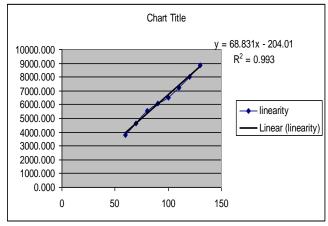


FIGURE 2: CALIBRATION CURVE OF STANDARD THYMOL

**Recovery Studies:** The results displayed in Table 3 shows that the used extraction method allowed the recovery of thymol b/w 90%-110%, indicating that both accuracy and recovery was satisfactory.

**TABLE 3: ACCURACY/RECOVERY OF THYMOL** 

Recovery level	Sample Weight (mg)	Standard Weight mg	Average Area	Theoretical value (in ppm)	Actual Value (in ppm)	Recovery%
80%	1024.6000	13.21	9739.61	257.47	231.89	90.07
100%	1041.8000	16.52	10685.83	241.22	254.42	105.47
120%	1018.2000	26.43	13455.90	339.82	320.38	94.28

**CONCLUSION:** The method was found to be precise for different concentrations of thymol. Accuracy was checked by conducting recovery at different level of thymol and the average percentage and recovery was found to be in the range 90%-110%. The Linearity was established over a range of seven different concentrations of

the analyte (60%,70%, 80%, 90%, 100%, 110%, 120% and 130% of test concentration) with correlation coefficients of 0.9993 for thymol. The proposed HPLC method was found to be simple, precise, specific, sensitive and accurate and can be used for routine quality control of rutin.

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