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SOLID DISPERSION- AN APPROACH TO ENHANCE THE DISSOLUTION RATE OF ACECLOFENAC BY USING 3² FACTORIAL DESIGN

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ABSTRACT

Aceclofenac is an analgesic and anti-inflammatory agent used in the management of osteoarthritis, rheumatoid arthritis and ankylosing spondylitis. The objective of the present work was to investigate the effect of different types of carriers such as polyvinyl pyrrolidone (PVP), polyethylene glycol (PEG) 6000 and sodium lauryl sulphate (SLS) as solubilizer on in vitro dissolution of aceclofenac. Aceclofenac solid dispersions were prepared using 32 factorial design by fusion and solvent evaporation method with PEG 6000, PVP & SLS. Prepared aceclofenac solid dispersions were evaluated for physical appearance, drug content uniformity, and in vitro dissolution studies. The dissolution was determined by USP XXIII apparatus using phosphate buffer pH 7.4. The highest aceclofenac dissolution rate, 99.87% in 60 minutes, was obtained from solid dispersion containing SLS (ASS₇) prepared by solvent evaporation method. The general trend indicated that there was an increase in dissolution rate for solid dispersions prepared in following order SLS>PVP>PEG 6000. IR and DSC studies showed no chemical change between drug and polymer and aceclofenac is homogeneously distributed in an amorphous state within the carrier and no aceclofenac crystallized out of the dispersions. The formulations studied were found to be stable. Finally it may be concluded that, dissolution rate of aceclofenac can be increased by solid dispersion technique, which may be due to increased hydrophilic nature of carrier and also possibly due to reduction in drug crystallinity.

INTRODUCTION: Poorly water-soluble drugs are increasingly becoming a problem in terms of obtaining the satisfactory dissolution within the gastrointestinal tract that is necessary for good bioavailability. It is not only existing drugs that cause problems but it is the challenge of medicinal chemists to ensure that new drugs are not only active pharmacologically but have enough solubility to ensure fast enough dissolution at the site of administration, often gastrointestinal tract ¹.

Solubilization is the process by which the apparent solubility of a poorly water soluble substance is increased. Solubilization techniques include addition of a cosolvent, salt formation, prodrug design, complexation, particle size reduction, and the use of surface active agents (Micellization) ². Use of solvate and hydrates ³, polymorphs ⁴, hydrotrophy ⁵, use of absorbents ⁶, pH adjustment, solubilizing vehicles, etc. are the some other physico-chemical approaches to enhancing oral absorption of poorly water soluble drugs.

Solid dispersion technology is the science of dispersing one or more active ingredients in an inert matrix in the solid stage in order to achieve increased dissolution rate, sustained release of altered solid state properties, enhanced release of drugs from ointment and suppository bases, and improved solubility and stability 7. Solid dispersions are prepared by various methods like Fusion process, Solvent process, Fusion Solvent process and Supercritical fluid process 8. Various methods, which can contribute information regarding the physical nature of the solid dispersions, are thermo analytical methods (Thermal Analysis ⁹, DSC ¹⁰, Xray Diffraction Methods ¹¹, Spectroscopic Methods and Microscopic Methods ⁹). Aceclofenac is aceclofenacum (O- (2, 6-dichloroaniline) phenyl] acetate glycolic acid ester, 2-(2, 6-dichloraniline)

phenyl acetoxyacetic acid. Aceclofenac is a NSAID. It is used in the management of osteoarthritis, rheumatoid arthritis and ankylozing spondylitis. Aceclofenac when taken orally shows gastrointestinal disturbances such as GΙ discomfort, nausea, and diarrhea. In some peptic ulceration patients and severe gastrointestinal bleeding may also occur 12.

Solid dispersion technology can be used to improve the *in vitro* and *in vivo* dissolution properties of dissolution dependent poorly water soluble drugs. PEG's, PVP ¹³ and surfactant like SLS ¹⁴ have been reported to be used for increasing the solubility of poorly soluble drugs. The usual dose of aceclofenac is 100 mg given twice daily by mouth. The initial dose should be reduced to 100 mg daily in patients with hepatic impairment. Its low solubility makes it a suitable candidate for solid dispersion systems.

MATERIALS AND METHODS:

Materials: Aceclofenac was obtained as a gift sample from Suyash Laboratories Ltd., and Polyvinyl pyrrolidone, Polyethylene Glycol 6000, Sodium lauryl sulphate, Microcrystalline cellulose, Lactose, Potassium dihydrogen orthophosphate, Sodium Hydroxide and methanol all other ingredients used were SD Fine Grade.

Methods of Preparation of Aceclofenac Solid Dispersion System: Solid dispersions of aceclofenac in PEG 6000, PVP & SLS were prepared using 3² factorial design with PEG 6000, PVP & SLS and aceclofenac as variables and maintaining the amount of lactose and MCC (4:1) as constant (**table 1**). The methods used for the preparation of these solid dispersions were physical mixtures, solvent evaporation method and fusion method.

- i) Physical Mixture: The physical mixtures were prepared by weighing the calculated amount of aceclofenac and the carriers and then mixing them in a glass mortar by triturating. The resultant physical mixtures was passed through 44-mesh sieve and stored in dessicator until used for further studies.
- ii) Solvent Evaporation Method ¹⁵: The required amount of aceclofenac and the carrier were dissolved in sufficient volume of methanol with continuous stirring. The solvent was then completely evaporated at 45°C with continuous stirring to obtain dry mass. The dried mass was pulverized passed through 44 mesh sieve and stored in dessicator until used for further studies.
- iii) **Fusion Method** ¹⁶: Accurately weighed amount of carrier was melted in a porcelain dish at 80-85°C and to this calculated amount of aceclofenac was added with thorough mixing for 1-2 minutes followed by quick cooling. The dried mass was then pulverized passed through 44-mesh sieve and stored in a dessicator until used for further studies. PVP containing solid dispersions were not prepared by the melt method, because PVP melts above 250°C and degrades before its melting point.

TABLE 1: FACTOR & LEVELS IN THE DESIGN OF ACECLOFENAC SOLID DISPERSIONS WITH DIFFERENT POLYMERS

- OLIMENS				
Independent	Levels			
variables	(-1) Lower	(0) Middle	(+1) Upper	
PEG 6000 (X ₁) mg	250	300	350	
PVP (X ₁) mg	250	300	350	
SLS (X ₁) mg	25	50	75	
Aceclofenac (X ₂) mg	50	75	100	

(Amount of other additives - lactose and MCC (4:1) was maintained constant in all the preparations);

Batches were made with the aid of factorial design

- Physical mixture (containing PEG 6000 (APG), containing PVP (APV), containing SLS (APS).
- Solvent evaporation method (containing PEG 6000 (ASG), containing PVP (ASV), containing SLS (ASS)
- Fusion method (containing PEG 6000 (AFG), containing SLS (AFS)

Evaluation of Aceclofenac Solid Dispersion Systems:

Physical Appearance: All the batches of Aceclofenac solid dispersions were evaluated for color and appearance.

Determination of Aceclofenac Content: An accurately weighed amount of each preparation was dissolved in small volume of methanol and further diluted with methanol. The content of aceclofenac was determined spectrophotometrically at 275 nm using Shimadzu UV-visible spectrophotometer.

In Vitro Dissolution: The dissolution study was carried out using USP XXIII apparatus type-II (Electrolab TDT-OCT). The dissolution medium was 900 ml, 7.4 pH phosphate buffer kept at 37±1°C. The drug or physical mixture or solid dispersions was taken in a muslin cloth and tied to the rotating paddle kept in the basket of dissolution apparatus, the basket was rotated at 50 rpm. Samples of 5 ml were withdrawn at specified time intervals and analyzed spectrophotometrically at 275 nm using Shimadzu- 1700 UV- visible spectrophotometer; the samples withdrawn were replaced by fresh buffer solutions. Each preparation was tested in triplicate and then, mean values were calculated.

Statistical Comparison: The dissolution release kinetics and result of best fit model among the preparations were also compared.

Experimental Batches for Factorial Design:

Stability Studies: The stability studies should be conducted on the drug substance packaged in a container closure system is the same as or simulates the packing proposed for storage and distribution. Stability studies on various batches like ASG7, ASV7, ASS7, AFG7 and AFS7 were carried out by storing 1 gm of solid dispersions in an amber colored screw capped bottle at different temperatures for a period of 3 months. The solid dispersions were visually examined for any physical change and drug content was estimated at the end of 3 months ¹⁸.

Differential Scanning Calorimetry: Differential scanning calorimetries (DSC) of some selected preparations were compared with plain aceclofenac, PVP, PEG 6000 and SLS was carried out. The DSC (pyris-6) thermograms were recorded at a heating of 10 °C/ min from 100°C to 300°C.

Infrared Spectroscopy: The infrared spectra (IR) of aceclofenac, PVP, PEG 6000 and SLS and some selected preparations were obtained using FTIR (Perkin Elmer 1600 Series). The IR spectrum was carried by KBr pellet method.

RESULTS & DISCUSSIONS:

Physical appearance: The aceclofenac solid dispersions were prepared employing three techniques, physical mixture, solvent evaporation method and fusion method. All these types of solid dispersions employing PVP, PEG 6000 and SLS were white fine powders.

Determination of Aceclofenac Content Uniformity: The drug content uniformity of formulations prepared was found to be 97.44±2.31.

In vitro drug release studies: Results of *in vitro* release of aceclofenac of selected formulations from various batches prepared by physical mixture, fusion method and solvent evaporation methods containing PEG 6000, PVP and SLS (table 2).

Stability Studies of Selected Batches of Solid Dispersions: Some selected solid dispersions were considered for stability studies. Formulations were stored at 40°C as per ICH guidelines. The drug content assayed after 3 months did not show any significant change (**table 3**).

TABLE 2: IN VITRO DRUG RELEASE STUDIES

Time	Percent drug release ±S.D.							
(min)	APG7	ASG7	AFG7	APV7	ASV7	APS7	ASS7	AFS7
0	0.000±0.0	0.000±0.0	0.000±0.0	0.000±0.0	0.000±0.0	0.000±0.0	0.000±0.0	0.000±0.0
10	19.477±0.45	25.642±0.41	21.943±0.55	21.121±0.42	27.902±0.35	25.642±0.17	35.300±0.29	30.779±0.0.33
20	31.299±0.21	49.211±0.67	39.326±0.39	38.089±0.37	41.620±0.44	42.635±0.24	57.689±0.18	48.623±0.53
30	39.075±0.61	60.374±0.33	55.573±0.24	51.657±0.18	53.769±0.29	56.228±0.08	73.421±0.44	66.153±0.38
40	47.716±0.15	69.749±0.12	63.277±0.24	59.340±0.42	66.806±0.51	62.292±0.32	84.512±0.17	75.560±0.31
50	55.581±0.09	75.064±0.27	66.502±0.15	63.570±0.38	74.571±0.22	69.210±0.25	93.607±0.14	81.318±0.2
60	63.900±0.24	78.146±0.19	69.948±0.07	66.795±0.5	81.761±0.13	74.932±0.2	99.873±0.09	83.818±0.12

Average of 3 determinations

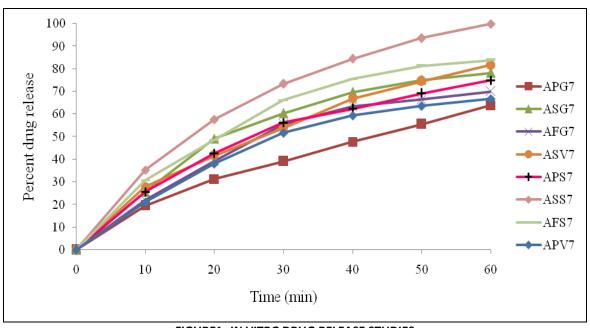


FIGURE1: IN VITRO DRUG RELEASE STUDIES

TABLE 3: STABILITY STUDIES OF SELECTED BATCHES OF SOLID DISPERSIONS

Formulation Code	Initial Drug Content ± S.D.	Drug Content After Storage (3 months) ± S.D.
ASG7	99.75±0.02	99.27±0.01
ASV7	99.51±0.04	98.78±0.03
ASS7	99.51±0.02	99.27±0.02
AFG7	99.51±0.01	98.29±0.02
AFS7	99.27±0.04	98.05±0.04

Average of 3 determinations

CONCLUSION: From the present study carried out on aceclofenac solid dispersion systems using PVP, PEG 6000 and SLS as solubilizer, using physical mixture, solvent evaporation method and fusion method resulted in fine white powder. Solid dispersions prepared by solvent evaporation method showed faster release of aceclofenac than those prepared by fusion method followed by physical mixture. The statistical study, indicated that amount of polymer affected the rate of release from aceclofenac solid dispersions prepared by physical mixture, solvent evaporation and fusion method. Maximum release was observed for ASS7 i.e., 99.87%, solid dispersions containing sodium lauryl sulphate (75 mg) with amount of aceclofenac (50 mg), lactose and MCC (4:1) used as additives prepared by solvent evaporation method. The general trend indicated

that there was an increase in dissolution rate for solid dispersions prepared in the following order SLS> PVP> PEG 6000. The mechanism of release of aceclofenac from highest releasing formulations in each batch designed by solvent evaporation method is as follows: ASG₇ (matrix r= 0.9911), ASV₇ (Peppas 'r'= 0.9990), and ASS7 (matrix 'r'= 0.9973). From the IR studies, it may be concluded that the drug, polymer and solubilizer used undergo physical interaction such as hydrogen bonding, there is no chemical change, thus the carrier (PVP, PEG 6000, SLS, MCC and lactose) are suitable for solid dispersion of aceclofenac. DSC studies showed that aceclofenac was homogeneously distributed within the carrier in an amorphous state and no drug crystallized out of dispersions. Stability studies indicated no

significant change in drug content. Thus, formulation studied was found to be stable.

Finally, it may be concluded that, dissolution rate of aceclofenac can be increased by solid dispersion technique which may be due to increased hydrophilic nature of the carrier and also possibly due to reduction in drug crystallinity.

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