IJPSR (2012), Vol. 3, Issue 10





INTERNATIONAL JOURNAL OF PHARMACEUTICAL SCIENCES AND RESEARCH

Received on 20 May, 2012; received in revised form 24 July, 2012; accepted 20 September, 2012

DEVELOPMENT AND VALIDATION OF HPLC METHOD FOR SOME AZOLES IN PHARMACEUTICAL PREPARATION

Wael Abu Dayyih^{*1}, Noor Al saadi¹, Mohmad Hamad², Eyad Mallah¹, Khalid Matalka² and Twafiq Arafat¹

Department of Pharmaceutical Medicinal Chemistry and Pharmacognosy ¹, Department of Pharmacology .and Biomedical Sciences ², Faculty of Pharmacy and Medical Sciences, Petra University, Amman, Jordan

ABSTRACT

Keywords: Azoles, Ketoconazole, Isoconazole, Miconazole, HPLC, validation, development

Correspondence to Author:

Wael Abu Dayyih

Department of Pharmaceutical Medicinal Chemistry and Pharmacognosy, Faculty of Pharmacy and Medical Sciences, Petra University, Amman, Jordan

E-mail: wabudayyih@uop.edu.jo

A new, simple and rapid reversed-phase high-performance liquid chromatography (HPLC) method was developed and validated for the determination of some azoles drug (Ketoconazole, Isoconazole and Miconazole) in pharmaceutical dosage forms in Jordan market. The HPLC separation was achieved on a C₁₈ BDS column (100 mm × 4.6 mm) using a mobile phase of acetonitrile and 0.05 M Ammonium Acetate buffer (70:30, v/v) that adjusted to pH 6 using phosphoric acid at a flow rate of 1 mL min⁻¹ and using UV absorbance detection at 254 nm. The method was validated for specificity, linearity, precision, accuracy, robustness. The separation was completed within 11.77 minutes and the assay was linear over the concentration range of 50% to 150% (r^2 = 0.9999), the percent recovery for test and reference formulation was (98.14%-101.12%).

INTRODUCTION: Azoles are synthetic compounds of which imidazoles and triazoles are two clinically useful families employed in the treatment of systemic fungal infections ^{1, 2, 3, 4, 5}, the basic structural unit of all of the antifungal azoles is a five-membered azoles ring which is attached by a carbon nitrogen bond to other aromatic rings, imidazole drugs contain two nitrogen atoms in the azoles ring. In contrast, the triazole class of agents contains a third nitrogen atom in the azoles ring ^{6, 7}.

Azoles antifungal agents inhibit the cytochrome P450 sterol 14-demethylase (CYP51) by a mechanism in which the heterocyclic nitrogen atom (N-3 of imidazole and N-4 of triazole) binds to the hemi iron atom in the binding site of the enzyme.

Ketoconazole (**fig. 1 (a**)) used as broad spectrum antifungal agent that is administered topically or orally. It is an imidazole derivative ^{8, 9}. Isoconazole (**fig. 1 (c**)) has been developed and marketed primarily as a oncea-day, topical anti-Candida agent for the treatment of vaginal candidiasis ¹⁰. Studies evaluating isoconazole have demonstrated that 80 to 90% of patients with vaginal candidiasis who were treated once a day with the drug remained clinically and mycologically cured ¹¹, application of the free base of isoconazole in combination with ethanol/propylene glycol, volatile/nonvolatile vehicle can increase drug bioavailability in the skin ¹².

Miconazole (**fig. 1 (b**)) another imidazole derivative, is applied widely as the nitrate salt (MIC), to treat fungal infections ^{8, 13, 14}, with wide activity spectrum, it is endowed with a powerful activity against dermatophytes and Candida albicans, as well as against



several Gram-positive germens ¹⁵, the generally accepted mode of action of azole antifungals is the inhibition of 14-alpha-lanosterol demethylase, a key enzyme in ergosterol biosynthesis, resulting in depletion of ergosterol and accumulation of toxic 14-alpha-methylated sterols in the membrane of susceptible yeast species ¹⁶.

Very few methods for development and analysis of azoles individually are reported in literature. Most methods include high performance liquid chromatography assay. Pascucci *et al* ¹⁷, A.Nguyen Minh Nguyet *et al* ¹⁸ and Y. Vander Heyden *et al* ¹⁹ have used ketoconazole with formaldehyde, the other method was optimization and validation with UV-detector for determination of ketoconazole in plasma and pharmaceutical preparations such as tablets, creams and shampoos ²⁰⁻²².

Literature survey reveals some methods for the determination of Miconazole in pharmaceutical preparations or in biological fluids including HPLC, (Tyler T.) did an experiment to analyze Miconazole, the sample is dissolved in ethanol, diluted in acetonitrilewater (1:1), and injected onto a C18 column. The mobile phase used consists of acetonitrile, a triethylammonium phosphate buffer; the active ingredient is determined using UV detection at 214 nm ²³⁻²⁷, a reversed-phase high performance liquid chromatography (RP-HPLC) method with UV detection is described for the simultaneous determination of miconazole in pharmaceutical dosage forms. Chromatography was carried out on a C18 reversedphase column, using a mixture of methanol-water (40:60, v/v) as a mobile phase, at a flow rate of 1.0 ml min^{-1 28}.

Gas chromatography 31 , also spectrophotometric method is based on high-performance liquid chromatography on a reversed-phase column using a mobile phase of methanol-water-phosphoric acid (30:70:0.20 v/v) (pH 2.8) with UV detection at 220.0 nm, as mentioned by Erk N $^{24, 30, 31, 32}$.

Many methods have been described in the literature for the determination of azoles were time consuming and expensive, as they require large amounts of reagents. However, the literature revealed that no method is reported for the simultaneous estimation of azoles combination products ³³⁻³⁸.

In this present paper we have developed and validated single HPLC method for determination combination of Ketoconazole (Keto), Isoconazole (Iso) and Miconazole (Mico) as active ingredients in some commercial products in Jordan market and in their raw materials. The developed method was accurate, robust and inexpensive.



FIGURE 1. A: CHEMICAL STRUCTURES OF KETOCONAZOLE (KETO), B: MICONAZOLE (MICO), C: ISOCONAZOLE (ISO)

MATERIALS AND METHOD:

1. Chemicals and reagents: Acetonitrile HPLC grade purchased from (RANKEM), Ammonium acetate, Phosphoric acid 85% were obtained from (Riedeldehaen). High pure water was prepared by using (milli-Q) distelled water system. Tefose, Batch Number: 120003, Labrafil, Batch Number: 120224, Ketoconazole, Batch Number: 0813k5, Miconazole, Batch Number: wsm 11/072, liquid paraffin and sodium benzoate were obtained from Dar Al Dawa, Jordan. Isoconazole, Batch Number: 11050063 were obtained from Al- Jazeera Pharmaceutical industry, KSA.

- 2. Equipment and Chromatographic Conditions: HPLC Thermo FINNIGAN, SUREVYOR UV-VIS plus SUREVYOR Auto detector, sampler plus, SUREVYOR LC pump plus. Analytical balance, (METTLER TOLEDO). рН meter, (METTLER TOLEDO). Magnetic stirrer, (VELP scientific). Ultrasonic, (Elmasonic), ELMAS 450H. Nylon syringe membrane filters 0.45 µm (TITAN2). PTFE syringe membrane filters 0.45 µm.UV-VIS: Therom scientific, EVOLUTION 300 UV-VIS. C18 column was used (100 mm x 4.6 mm, thermoquest) with the mobile phase containing acetonitrile and 0.05 M Ammonium Acetate buffer in the ratio of 70:30 (v/v pH 6.0) at ambient temperature. Flow rate was kept at 1.0 ml/min and the elution was monitored at 254 nm.
- 3. **Preparation of Standard Solutions:** A stock standard solution of each Keto, Iso and Mico was prepared by dissolving 5mg, 40mg and 40mg (highly pure material >99.80%) respectively in 20 mL mobile phase, then 5mL from each solution was taken and diluted in 10mL mobile phase, the concentration obtained for each form; Keto 0.125g/L, Iso 1g/Land for Mico 1g/L.
- 4. **Preparation of Placebo Solution:** The placebo solution was prepared based on the excepients present in the tablet and cream samples without any of the target substances as active- ingredients.

A placebo solution was prepared by dissolving 1.00 g of a placebo mixture of excepients in 20.0 ml mobile phase.

RESULTS AND DISCUSSION: Optimization of chromatographic conditions: are included parameters such as: composition of mobile phase, pH, and retention time and in additionally it provides information about robust method in variation of some parameters like pH and temperature. With the optimized chromatographic conditions, a steady baseline was recorded.

The retention time was found to be 6.48, 10.54 and 11.77 min for KETO, ISO and MICO respectively. A typical chromatogram of sample solution is given in figures 2, 3 and 4. Detection was done at 254 nm and separation was performed using BDS C18 column with 3μ m particle size (10cm x4.6mm). The assay procedure was repeated for ten times and mean peak was calculated. All results are compared with validation of analytical procedure and methodology, according to ICH guidelines³⁹.

Method Validation:

Precision: The purpose of system precision is to find the degree of agreement between individual test results when the procedure is applied repeatedly to multiple injections (10 injections) of the same homogenous sample (**Table 1**). All RSD values were below 2% which indicates good system suitability.

TABLE 1: RSD VALUES FOR 10 SAMPLES FOR EACH KETO, ISO AND
MICO

Parameters	KETO (AUC)	ISO (AUC)	MICO (AUC)
Average area of 10 injections	2717352	506364	655162
RSD%	0.17	0.14	0.14
Asymmetry	1.21	1.51	1.43
Theoretical plates	51475	43890	47591
Resolution	0	26.3	6.2
Retention time	6.385	10.503	11.740

Accuracy:

Dilution for Accuracy and Recovery Studies: Recovery studies were carried out by addition of standard drug solution to sample at 3 different levels In order to estimate the accuracy, three samples at three different concentrations levels 50,100 and 150% were analyzed. Analytical concentration at level 100% is analyzed in triplicate preparation at each level against standard as shown in **Table 2**.

TABLE 2: RESULTS OF ACCURACY FOR MICO

	50%	100%	150%
Area (AUC)	308174	610745	915203
Assay %	99.45	100.37	100.47
RSD%	0.12	0.25	0.14

Figures 2, 3 and 4 show a correlation relationship in which an increase in concentration leads to increase in the area of the peaks.



FIGURE 2: ACCURACY TEST OF 50% CONCENTRATION.



FIGURE 3: ACCURACY TEST OF 100% CONCENTRATION TABLE 3: RESULTS OF STABILITY TEST FOR AZOLES (ASSAY %)

Time and Temp.	Ketoconazole(KETO)	Miconazole(MICO)	Isoconazle(ISO)
Standard fresh sample	99.36%	97.1%	97.64%
24 hours at 5C°.	98.62%	98.2%	99.06%
24 hours at 25C°.	97.7%	98%	98.51%

It is important to know at which level the analyte solution are stable. This could be evaluated by determining the concentrations of the analytical solutions (standard and sample) which were stored for 24 Hours under fridge and at room temperature compared to freshly prepare standard solutions. The accepted limits of stability were within 98.0% -102.0%.

Robustness regarding Temperature Variation: The method used shows a robustness regarding to the temperature slight changes. These results are obtained by comparing the standard solution tested under room temperature 25celsius versus ±5celsius samples solutions. The results are shown in Table 4.

TABLE 4: ROBUSTNESS TEST RESULTS REGARDING TO CHANGE IN TEMPERATURE BY +5 CELSIUS (30°C).

	KETO.	ISO.	MICO.
Area	2625850	499571	646245
RSD%	0.12	0.12	0.14
Theoretical plates	53942	50006	52823
Asymmetry	1.17	1.36	1.34
Resolution	0	25.5	6.23



FIGURE 4: ACCURACY TEST OF 150% CONCENTRATION

Stability of Analytical Solution: Analytical concentration at level 100% is analyzed by preparations at each level (duplicate readings) against a standard solution. Concentration for each are about 0.125 g/L for KETO, 1 g/L for ISO and 1g/L for MICO.

The stability results of sample and standard solutions under fridge and room temperature are shown in **Tables 3**.



FIGURE 5: THE 30[°]C TEMPERATURE CHROMATOGRAM

Electivity: It is important to study the selectivity of the method to determine the capability of the analytical procedure to measure accurately and specifically in the presence of the active ingredients, placebo and another ingredient.

A standard, sample, solvent and a placebo solution were injected into the column according to the parameters stated under the developed method. It was found that there is no interference between the analyte and both the solvent or placebo. The selectivity test also includes the analysis of drugs in the pharmaceutical formulation, comparing the results of analysis between local Jordanian manufactured drugs formulations with some international foreign formulations.

A local finished product such as Kenazole^R tablets (Pharma International- Jordan) for Ketoconazole, Azonit^R cream (Pharma International- Jordan) for Isoconazole and Mycoheal^R Dar AL Dawa- Jordan, for Miconazole (test formulations). With the use of Nizoral^R tablets (Janssen-Cillag) for Ketoconazole, Travogen^R cream (Schering) for Isoconazle and Daktarin^R cream (Janssen-Cillag) for Miconazole, all as a foreign manufactured drug formulations (Reference formulations).

The results of recovery% are shown in **Table 5**.

TABLE 5: RECOVERY% FOR TEST AND REFERENCE FORMULATION.

	KETO	ISO	MICO
Test formulations	98.74	99.13	100.4
Reference formulations	98.33	101.12	98.14

Linearity and Range: A series of standard samples at different concentrations of the target compounds were prepared to evaluate the linearity. After analyzing each preparation in duplicate, a linear regression analysis was performed on the average peak areas versus the concentration of the levels studied. Results for KETO, ISO and MICO are shown in **Table 6.**

TABLE 6. RESULTS OF LINEARITY TEST FOR KETO

Concentration	Average area	RSD%
50% (0.0625 g/L)	1366416	0.24
70% (0.0875 g/L)	1877863	0.53
100% (0.125g/L)	2658483	0.05
120% (0.15 g/L)	3168156	0.08
150% (0.1875 g/L)	3955098	0.07

The calibration curve was constructed by plotting response factor against concentration of drugs. The slope and intercept value for calibration curve was:

 $R^2 = 0.999,$

slope =2587792 and intercept = 69166

The equation of calibration curve is:

Absorbance = 2587792x concentration +69166

The calibration curve of peak area versus concentration% for KETO is shown in figure 6. The results show that an excellent correlation exists between response factor and concentration of drugs within the concentration range indicated above.



FIGURE 6: CALIBRATION CURVE OF PEAK AREA VERSUS CONCENTRATION %FOR KETO.

The value of $R^2 < 1$ which indicating a good linearity test result.

The concentrations of ISO (50%-150%) are analyzed by using HPLC and give a good RSD% values.

TABLE 7: RESULTS OF LINEARITY TEST FOR ISO.

Concentration	Average area	RSD%
50% (0.5 g/L)	251201	0.41
70% (0.7 g/L)	348078	0.63
100% (1 g/L)	494350	0.1
120% (1.2 g/L)	592502	0.05
150% (1.5 g/L)	743569	0.07

Linearity equation of ISO samples gives the data:

Slope = 491807 .Intercept =3968,The equation of calibration curve is:

Absorbance =491807xconcentration + 3968



FIGURE 7: CALIBRATION CURVE OF PEAK AREA VERSUS CONCENTRATION % FOR ISO

The figure shows a good linearity for ISO within the stated limit to obtain a linear validation method, R^2 <1so the curve in **figure 5** is linear curve.



FIGURE 8: CALIBRATION CURVE OF PEAK AREA VERSUS CONCENTRATION % FOR MICO

CONCLUSION: In summary, a simple, rapid, accurate and practical HPLC method is described for determination of some azoles in pharmaceutical preparation. The validation data demonstrate good precision and accuracy, which prove the reliability of proposed method. Hence, this HPLC method could be used routinely for identification and estimation stability of dosage forms. The developed method is suitable for the quality control of the raw materials, formulations and dissolution studies.

ACKNOWLEDGMENT: The authors would like to convey appreciation to Dar Al Dawa pharmaceutical company –Amman –Jordan for technical support.

REFERENCES:

- Zhu J., Lu J., Zhou Y., Li Y., Cheng J., Zheng C.: synthesis and antifungal activities in vitro of novel tetrahydroisoquinoline compounds based on the structure of lanosterol 14ademethylase (CYP51) of fungi. Bioorg Med. Chem Lett. 2006; 16:5285–5289.
- Roberts T.R., Hutson D.: Metabolic Pathways of Agrochemicals. Part 2: Insecticides and Fungicides. Cambridge: The Royal Society of Chemistry;p 1999; 1011–1104.
- 3. Sheehan D.J., Hitchcock C.A., Sibley C.M.:Current and emerging azole antifungal agents. Clin Microbiol Rev. 1999; 12:40–79.
- 4. Andrews F. A., L.R. Peterson, W.H. Beggs, D. Crankshaw, G. Sarosi, Antimicrob. Agents Chemother. 19 (1981) 110.
- Dodds Ashley ES, Lewis R, Lewis JS, Martin C, Andes D.: Pharmacology of systemic antifungal agents. Clin Infect Dis 2006;43:28-39.
- Michael S. Saag and William E. Dismukes Jan. 1988, 1-8, Vol. 32, Azole Antifungal Agents: Emphasis on New Triazoles.
- 7. Bastide M., S. Jouvert, and M. J. Bastide. 1982. A comparison of the effects of several antifungal imidazole derivatives and

polyenes on Candida albicans: an ultrastructural study by scanning electron microscopy. Can. J. Microbiol. 28:1119-1126.

- Reynolds J.E.F. (Ed.), 29th ed, Martindale, the Extra Phar- 20. macopoeia, The Pharmaceutical Press, London, 1989 Riley C.M., M.O. James, J. Chromatogr. 377 (1986) 287.
- 9. Clarke's isolation and identification of drugs in pharmaceuticals, body fluids, and post-mortem material, second ed., The Pharmaceutical Society of Great Britain, The Pharmaceutical Press, London, 1986.
- Hdffken, G., H. Lode, and H.-J. Kessler. 1983. Einfluss von Doxycyclin und Isoconazolnitrat auf die intestinale Pilzflora des Menschen. Arneim. Forsch. 33:273-276.
- 11. Walther, M. 1982. Feldstudie die Einmaltherapie von Vaginalmykosenmit Isoconazolnitrat. Praxis 71:152-156.
- 12. Tauber, U. 1987. Availability of isoconazole in human skin after dermal application as free base and as nitrate *in vitro*. Arneim. Forsch. 37:461-463.
- Hardman J.G. and L.E. Limbird (Eds.), Goodman & Gilman's, The Pharmacological Basis of Therapeutics, 9th ed., McGraw- Hill, New York, 1996.
- 14. Stevens DA, Levine HB, Devesinsky SC. Miconazole in coccidiomycosis: therapeutic and pharmacological studies in man. Am J Med 1976;60:191–202.
- Piel G., B. Evrard, M. Fillet, G. Llabres, L. Delattre, Int. J. Pharm. 169 (1998) 15.
- 16. Nafee N. A. , F.A. Ismail, N.A. Boraie, L.M. Mortada, Int. J. Pharm. 264 (2003) 1.
- 17. 17.Pascucci V. L., J. Bennett, P.K. Narang, D.C. Chatterji, Quantitation of ketoconazole in biological fluids using highperformance liquid chromatography, J. Pharm. Sci. 72 (1983) 1467–1469.
- Minh Nguyet A Nguyen, L Tallieu, J Plaizier-Vercammen, D L Massart, Y Vander Heyden.Validation of an HPLC method on short columns to assay ketoconazole and formaldehyde in shampoo.Journal of Pharmaceutical and Biomedical Analysis (2003)
- Y. Vander Heyden, A. Nguyen Minh Nguyet, M.R. Detaevernier, D.L. Massart J. Plaizier-Vercammen"Simultaneous determination of ketoconazole and formaldehyde in a shampoo: liquid chromatography method development and validation"Journal of Chromatography A, 958 (2002) 191–201.
- M.V. Vertzoni, C. Reppas, H.A. Archontaki. Optimization and validation of a high-performance liquid chromatographic method with UV detection for the determination of ketoconazole in canine plasma Journal of Chromatography 2006.03.010
- 21. Low A. S., J. Wangboonskul, An HPLC assay for the determination of ketoconazole in common pharmaceutical preparations, Analyst 124 (1999) 1589.
- Alton K. B., Determination of the antifungal agent, ketoconazole, in human plasma by high-performance liquid chromatography, J. Chromatogr. 221 (1980) 337–344.
- 23. Tyler T. A., J.A. Genzale, Liquid chromatographic determination of miconazole nitrate in cream and suppositories, J. Assoc. Off. Anal. Chem. 72 (1989) 442_/444.
- Cavrini V. , A.M. Di Pietra, R. Gatti, Analysis of miconazole and econazole in pharmaceutical formulations by derivative UV spectroscopy and liquid chromatography (HPLC), J. Pharm. Biomed. Anal. 7 (1989) 1535_/1543.
- Pietra A.M., V. Cavrini, V. Adrisano, R. Gatti, HPLC analysis of imidazole antimycotic drugs in pharmaceutical formulations, J.Pharm. Biomed. Anal. 10 (1992) 873_/879.
- 26. Kobylinska M., K. Kobylinska, B. Sobik, High performance liquid chromatographic analysis for the determination of miconazole

in human plasma using solid-phase extraction, J. Chromatogr. B: Biomed. Appl. 685 (1996) 191_/195.

- Musch G., D.L. Massart, Expert system for pharmaceutical analysis; II relative contribution of and rule validation for amperometric detection (oxidation mode), J. Chromatogr. 370 (1986) 1 /19.
- 28. Akay C. , S.A. Ozkan, Z. Senturk, S. Cevheroglu, II Farmaco 57 (2002) 953.
- 29. Neill G. P., N.W. Davies, S. Mc Lean, Automated screening procedure using gaschromatography-mass spectrometry for identification of drugs after their extraction from biological samples, J. Chromatogr. 565 (1991) 207_/224.
- Wrobel K., K. Wrobel, I.M. de la Garza Rodriguez, P.L. Lopezde-Alba, L. Lopez-Martinez, Determination of miconazole in pharmaceutical creams using internal standard and second derivative spectrophotometry, J. Pharm. Biomed. Anal. 20(1999) 99_/105.
- Khashaba P. Y., S.R. El-Shabouri, K.M. Emara, A.M. Mohamed, Analysis of some antifungal drugs by spectrophotometric and spectrofluorimetric methods in different pharmaceutical dosage forms, J. Pharm. Biomed. Anal. 22 (2000) 363_/376.
- 32. Erk N., M. Levent-Altun, J. Pharm. Biomed. Anal. 25 (2001) 115.

- Tendolkar N. M. , B.S. Desai, V.M. Shinde, Simultaneous determination of tinidazole clotrimazole from tablets by reversedphase HPLC, Indian Drugs 31 (1994) 551.
- Turner C. A., A. Turner, D.W. Warnock, High performance liquid chromatographic determination of ketoconazole in human serum, J. Antimicrob. Chemother. 18 (1986) 757–763.
- Tyler T. A., J.A. Genzale, Liquid chromatographic determination of miconazole nitrate in cream and suppositories, J. Assoc. Off. Anal. Chem. 72 (1989) 442_/444
- Wulff H., Miller M.J., Hansel W., Grissmer S., Cahalan M.D., Chandy K.G. (2000) Design of a potent and selective inhibitor of the intermediate-conductance Ca2+ -activated K+ channel, IKCa1: a potential immunosuppressant. Proc Nat Acad Sci USA; 97:8151–8156.
- Yuen K. H., K.K. Peh, Simple high-performance liquid chromatographic method for determination of Ketoconazole in human plasma, J. Chromatogr. (Biomed. Appl.) 715 (1998) 436.
- Zhao C. , L. Zhu, Z. Zhang, F. Jin, X. Wang, Quantitative determination of clotrimazole pH indicator absorbance ratio method, Shenyang Yaoxueyuan Xuebao 10 (1993); 27.
- 39. ICH guidelines–validation of analytical procedure: text and methodology, 2011

How to cite this article:

Abu Dayyih W, Al Saadi N, Hamad M, Mallah E, Matalka K and Arafat T: Development and Validation of HPLC method for some Azoles in Pharmaceutical Preparation. *Int J Pharm Sci Res.* 3(10); 3686-3692.