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## SYNTHESIS AND ANTITUBERCULAR ACTIVITY OF SOME SUBSTITUTED 2, 3-SUBSTITUTED QUINAZOLINONE ANALOGS

S. Rajasekaran<sup>\*1</sup>, Gopalkrishna Rao<sup>2</sup>, H.S. Niranjana<sup>1</sup>

Department of Pharmaceutical Chemistry, Al-Ameen College of Pharmacy<sup>1</sup>, Near Lalbagh Main Gate, Hosur Road, Bangalore-560 027, Karnataka, India  
Goa College of Pharmacy<sup>2</sup>, 18<sup>th</sup> June Road, Panaji-403 001, Goa, India

### ABSTRACT

#### Keywords:

Quinazolinone,  
Thiophene-2-carboxaldehyde,  
2-acetyl furan,  
Antitubercular activity

#### Correspondence to Author:

S. Rajasekaran

Assistant Professor, Department of  
Pharmaceutical Chemistry, Al-Ameen  
College of Pharmacy, Near Lalbagh Main  
Gate, Hosur Road, Bangalore-560 027,  
Karnataka, India

E-mail: rajasekaranpharm@gmail.com

In recent years, there is a tremendous increase of drug resistant pathogens, leading to the design and development of newer antitubercular agents. A series of 2-(substituted-phenyl)-1,3-benzoxazin-4-ones and 3-amino-2-(substituted-phenyl)quinazolin-4(3H)-ones were synthesized by the reaction of substituted benzoyl chloride with anthranilic acid and 2-(substituted phenyl)-1,3-benzoxazin-4-one with hydrazine hydrate in absolute alcohol respectively. The title compounds 2-substituted phenyl-3-(thiophen-2-ylmethyleneamino) quinazolin-4(3H)-ones and 2-(substituted phenyl)-3-(1-(furan-2-yl)ethylideneamino) quinazolin-4(3H)-ones were obtained by reacting 3-amino-2-(substituted phenyl) quinazolin-4(3H)-one with thiophene-2-carboxaldehyde and 2-acetyl furan respectively. The title compounds have been characterized by UV, IR, <sup>1</sup>H-NMR and Mass spectra. The synthesized compounds have been evaluated for their antitubercular activity. Few of the compounds exhibited significant antitubercular while other compounds showed moderate activity.

**INTRODUCTION:** Quinazolinones are an interesting class of organic compounds being studied over the years and reported to possess a wide spectrum of biological activities such as antibacterial<sup>1</sup>, antifungal<sup>2</sup>, Antitubercular<sup>3</sup>, anti-inflammatory<sup>4</sup>, antioxidant<sup>5</sup> and anticonvulsant<sup>6</sup>, antihypertensive<sup>7</sup>, bronchodilator<sup>8</sup>. The pharmacodynamic versatility of quinazolin-4-one moiety has been documented not only in many of its synthetic derivatives but also in several naturally occurring alkaloids isolated from animals, families of plant kingdoms and from microorganism<sup>9</sup>.

Quinazolinone is a versatile lead molecule for designing potential bioactive agents. Imines (Schiff bases) and their reaction products are an interesting class of organic compounds being studied over the years and reported to possess a wide spectrum of biological

activities such as antimicrobial<sup>10, 11</sup> antioxidant<sup>12</sup>, cytotoxic<sup>13</sup> and anticonvulsant<sup>14</sup>. They are formed by simple condensation of amine and aldehyde in alcoholic medium to form an azomethine linkage.

Tuberculosis (TB) is one of the oldest and most pervasive diseases in history. According to alarming data from the World Health Organization (WHO), TB has spread to every corner of the globe.



As much as one-third of the world's population is currently infected and more than 5000 people die from TB every day. It is estimated that between 2002 and 2020, approximately 1000 million people will be newly infected, over 150 million people will develop diseases and 36 million will die of TB if proper control measures are not established. The Directly Observed Treatment, short-course (DOTS) strategy, constitutes the cornerstone of the current protocol for control of TB. However, the three key drugs, isoniazid, pyrazinamide and rifampicin, used in the regimen are potentially hepatotoxic and may lead to drug associated hepatitis.

Hence, an attempt has been made in the present study to synthesize some novel compounds of quinazolinones containing five membered thiophene or furan substitutions fused through an imine linkage and evaluate the compounds for their antitubercular activity.

**MATERIALS AND METHODS:** Melting points were measured in open capillary tubes and are uncorrected. IR (KBR) spectra were recorded in film or in potassium bromide disks on a Perkin-Elmer 39 spectrophotometer ( $\nu_{\max}$  in  $\text{cm}^{-1}$ ) and  $^1\text{H}$  NMR spectra on a DPX 300 MHz Bruker FT-NMR spectrophotometer. The chemical shifts were reported as parts per million ( $\delta$  ppm) tetramethyl silane (TMS) as internal standard.

Mass spectra were obtained on a JEOL-SX-102 instrument using fast atom bombardment (FAB

positive). Elemental analysis was performed on a Perkin-Elmer 2400 C, H, N analyzer. The progress of the reaction was monitored on a readymade silica gel plates (Merck) using n-hexane: ethyl acetate as a solvent system. Spectral data (IR,  $^1\text{H}$ NMR and Mass spectra) confirmed the structure of the synthesized compounds and the purity of these compounds were ascertained by microanalysis.

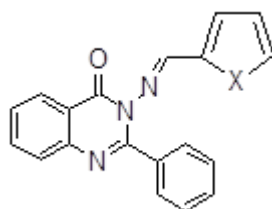
#### Procedure:

**Synthesis of 2-aryl-4H-3, 1-benzoxazin-4-one (1) and 3-amino-2-phenyl quinazolin-4(3H)-one (2):** These compounds were synthesized by following the reported procedure<sup>15</sup>.

**Synthesis of 2-aryl-3-(thiophen-2-ylmethyleneamino) quinazolin-4(3H)-one:** 3-Amino-2-substitutedphenyl quinazolin-4(3H)-one (0.008mole) was taken in a round bottomed flask containing absolute alcohol. Thiophene-2-carboxaldehyde (0.008mole) was added and the reaction mixture was refluxed with a catalytic amount of conc.  $\text{H}_2\text{SO}_4$  for 5h. The contents were poured into a beaker containing crushed ice; the solid obtained was filtered, washed with water and recrystallized from ethanol.

All the compounds of the series were synthesized by following the above procedure. The physical and the spectral data of the synthesized compounds are given in **Table 1 and 2** respectively.

**TABLE 1: PHYSICAL DATA OF THE SYNTHESIZED COMPOUNDS**



Comp. Code	R	X	Colour	Mol formula	Mol.wt	m.p °C	% yield	R <sub>f</sub> *
SRN 1	H	S	cream	C <sub>19</sub> H <sub>13</sub> N <sub>3</sub> OS	331	143-147	33.33	0.82
SRN 2	2-Cl	S	brown	C <sub>19</sub> H <sub>12</sub> ClN <sub>3</sub> OS	365	100-105	15.38	0.60
SRN 3	4-Cl	S	cream	C <sub>19</sub> H <sub>12</sub> ClN <sub>3</sub> OS	365	145-148	30.00	0.70
SRN 4	4-CH <sub>3</sub>	S	cream	C <sub>20</sub> H <sub>15</sub> N <sub>3</sub> OS	345	145-147	22.23	0.70
SRN 5	2-CH <sub>3</sub>	S	brown	C <sub>20</sub> H <sub>15</sub> N <sub>3</sub> OS	345	120-123	41.66	0.83
SRN 6	H	O	white	C <sub>20</sub> H <sub>15</sub> N <sub>3</sub> O <sub>2</sub>	329	180-185	38.50	0.41
SRN 7	2-Cl	O	brown	C <sub>20</sub> H <sub>14</sub> ClN <sub>3</sub> O <sub>2</sub>	363	195-200	18.72	0.57
SRN 8	4-Cl	O	cream	C <sub>20</sub> H <sub>14</sub> ClN <sub>3</sub> O <sub>2</sub>	363	190-195	52.63	0.63
SRN 9	4-CH <sub>3</sub>	O	brown	C <sub>21</sub> H <sub>17</sub> N <sub>3</sub> O <sub>2</sub>	343	155-160	30.76	0.60
SRN 10	2-CH <sub>3</sub>	O	brown	C <sub>21</sub> H <sub>17</sub> N <sub>3</sub> O <sub>2</sub>	343	150-153	34.24	0.60

\* TLC Solvent - n Hexane: Ethyl acetate = 1:1

TABLE 2: SPECTRAL DATA OF THE SYNTHESIZED COMPOUNDS

Comp Code	Compound	UV ( $\lambda_{max}$ )	IR (KBr) $cm^{-1}$ / $^1H$ NMR ( $CDCl_3$ , $\delta$ )
SRN 1	2-Phenyl-3-(thiophen-2-ylmethyleneamino)quinazolin-4(3H)-one	298	3103 Ar CH str, 1679 CO str/ 9.25 s,1H, N=CH, 7.43-8.38 m,9H, ArH, 7.26-7.43 m,3H, ArH of thiophene
SRN 2	2-(2-Chlorophenyl)-3-(thiophen-2-ylmethyleneamino)quinazolin-4(3H)-one	330	3115 Ar CH str, 1654 CO str, Ar C=C str 1570
SRN 3	2-(4-Chlorophenyl)-3-(thiophen-2-ylmethyleneamino)quinazolin-4(3H)-one	237	3054 Ar CH str, 1673 CO str, 1590 Ar C=C str
SRN 4	3-(Thiophen-2-ylmethyleneamino)-2-4-tolylquinazolin-4(3H)-one	303	3220 Ar CH str, 1684 CO str, 1591 Ar C=C str
SRN 5	3-(Thiophen-2-ylmethyleneamino)-2-2-tolylquinazolin-4(3H)-one	331	3024 Ar CH str, 1672 CO str, 1596 Ar C=C str
SRN 6	3-(1-(Furan-2-yl)ethylideneamino)-2-phenylquinazolin-4(3H)-one	282	3214 Ar CH str, 1661 CO str/ 7.52-8.34 m,9H,ArH, 7.26-7.50 m,3H, ArH of furan, 1.63 s,3H,CH <sub>3</sub>
SRN 7	2-(2-Chlorophenyl)-3-(1-(furan-2-yl)ethylideneamino)quinazolin-4(3H)-one	284	3140 Ar CH str, 1678 CO str,
SRN 8	2-(4-Chlorophenyl)-3-(1-(furan-2-yl)ethylidene amino)quinazolin-4(3H)-one	232	3214 Ar CH str, 1654 CO str,
SRN 9	3-(1-(Furan-2-yl)ethylideneamino)-2-4-tolylquinazolin-4(3H)-one	271	3115 Ar CH str, 1682 CO str,
SRN 10	3-(1-(Furan-2-yl)ethylideneamino)-2-2-tolylquinazolin-4(3H)-one	284	3273 Ar CH str, 1659 CO str,

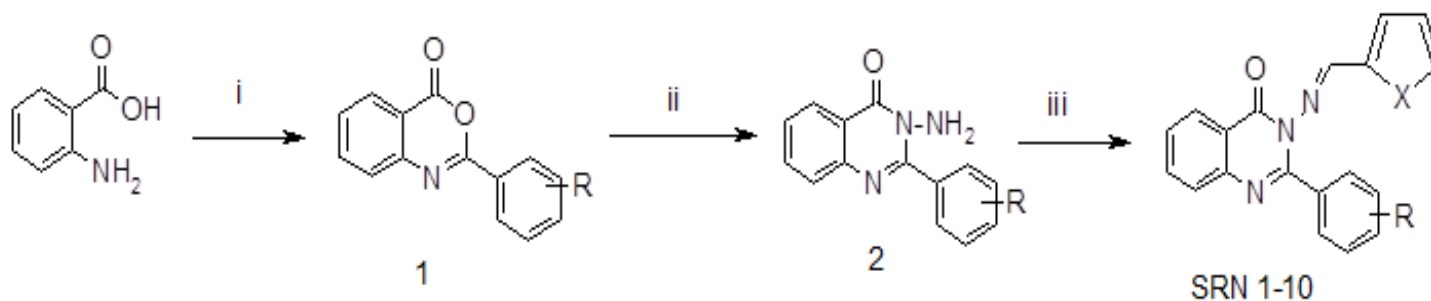


FIGURE 1: SCHEME OF SYNTHESIS, (I) VARIOUS SUBSTITUTED ARYL ACID CHLORIDES, (II) HYDRAZINE HYDRATE, (III) THIOPHENE 2-CARBOXALDEHYDE/FURAN 2-CARBOXALDEHYDE

**Antitubercular Activity:** The antitubercular activity<sup>16</sup> of compounds evaluated against *M. tuberculosis* using Microplate Alamar Blue Assay (MABA) method. This method is non-toxic, uses a thermally stable reagent and shows good correlation with proportional and BACTEC radiometric method. Briefly, 200 $\mu$ l of sterile deionized water was added to all outer perimeter wells of sterile 96 wells plate to minimize evaporation of medium in the test wells during incubation. The 96 wells plate received 100  $\mu$ l of the Middlebrook 7H9 broth and serial dilution of compounds were made directly on plate.

The final drug concentrations tested were 100 to 0.2  $\mu$ g/mL. Plates were covered and sealed with parafilm and incubated at 37°C for five days. After the specified period, 25 $\mu$ l of freshly prepared 1:1 mixture of Almar Blue reagent and 10% tween 80 was added to the plate and incubated for 24 hrs. A blue color in the well was interpreted as no bacterial growth, and pink color was scored as growth. The MIC was defined as lowest drug concentration which prevented the color change from blue to pink.

TABLE 3: ANTITUBERCULAR ACTIVITY OF SYNTHESIZED COMPOUNDS

Comp. Code	Concentration in $\mu\text{g/ml}$									
	100	50	25	12.5	6.25	3.125	1.6	0.8	0.4	0.2
SRN 1	S	S	R	R	R	R	R	R	R	R
SRN 2	S	S	S	S	S	S	R	R	R	R
SRN 3	S	S	S	S	R	R	R	R	R	R
SRN 4	S	S	S	S	R	R	R	R	R	R
SRN 5	S	S	S	R	R	R	R	R	R	R
SRN 6	S	S	R	R	R	R	R	R	R	R
SRN 7	S	R	R	R	R	R	R	R	R	R
SRN 8	S	S	S	R	R	R	R	R	R	R
SRN 9	S	S	S	S	R	R	R	R	R	R
SRN 10	S	S	R	R	R	R	R	R	R	R

\*S=Sensitive, R=Resistant

**RESULTS & DISCUSSION:** The compounds SRN 3, 4 and 9 possessing 4-chloro, 4-methyl substitution and thiophene side chain linked to quinazolinone moiety through imine group have shown activity at a concentration of 12.5  $\mu\text{g/ml}$ , while other derivatives have shown activity above 50  $\mu\text{g/ml}$ . the compound SRN2 which possess 2-chlorophenyl substitution at 2<sup>nd</sup> position of quinazolinone moiety and has thiophene ring system in the side chain has shown significant activity (3.125  $\mu\text{g/ml}$ ), highlighting the effect of substitutions on the quinazolinone moiety for their biological property. Hence these compounds shall be exploited further for their antitubercular activity to attain a potential pharmacophore.

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