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ENHANCEMENT OF SOLUBILITY AND DISSOLUTION RATE OF BCS CLASS II DRUG RITONAVIR USING LIQUISOLID TECHNIQUE

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ABSTRACT: BCS class II drugs usually suffer inadequate bioavailability as a dissolution step is the absorption rate-limiting step. In this work, the effect of solubility increase at the main absorption site for these drugs was investigated using ritonavir as a drug model. The liquidsolid technique was applied to prepare ritonavir per-oral tablets of high dissolution rate using versatile non-ionic surfactants of high solubilizing ability [Transcutol HP, labrasol and labrasol/labrafil (1:1) mixture] as liquid vehicles at different drug concentrations (10–30%) and fixed (R). The prepared liquidsolid tablets were fully evaluated, and the dissolution rate at pH 1.2 was investigated. The formula of higher dissolution rate was subjected to solid state characterization using Differential Scanning Calorimetric (DSC), Infrared Spectroscopy (IR). Results showed that liquidsolid tablet prepared using labrasol/labrafil (1:1) mixture as a liquid vehicle containing 10% ritonavir is a compatible formula with a higher dissolution rate (100% in 35 min). This led to the conclusion that liquidsolid technique efficiently improved the drug solubility and dissolution rate of BCS class II drugs.

INTRODUCTION: To get a significant plasma drug concentration a desired pharmacological action is required. The plasma drug concentration is related to solubility in GI fluids ¹. The presence of the drug in solution form is important for drug absorption mechanisms from GIT ². According to BCS classification system, poorly water-soluble drugs are categorized as Class-II. In these, dissolution is the rate-limiting step in the absorption process.

For such drugs improvement of drug solubility and dissolution is the main challenge after oral administration ³. Till date, different techniques had been designed for improving the solubility characteristics of poorly water-soluble drugs.

Some of these include solvent change co-precipitation ⁴, solid dispersion ⁵, inclusion complexes with β -cyclodextrins ⁶, nanosuspensions ⁷, microencapsulation ⁸, soluble salts formation ⁹, lyophilization ¹⁰ and liquidsolid technique ¹¹. The liquidsolid technique is a recent one for the improvement of solubility of insoluble drugs. In this technique for the preparation of liquid medication, the poorly soluble drug is dissolved or suspended in a water-miscible non-volatile solvent. As per mathematical model described by Spireas ¹²

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to get dry, flowable, good compressibility powder mixture characteristics the liquid medication is mixed with selective powder additives of significant absorbing/adsorbing carrier and coat characteristics¹³.

This mathematical model depends on the flowable liquid retention potential (T-Value) of both carrier and coat materials at the selected carrier/coat ratio (*R*) to calculate the formula liquid load factor (Lf) which is directly used to calculate the amounts of carrier and coat materials that are usually enough to maintain good flow and compression properties to the prepared liquisolid compact^{14, 15}. The liquisolid technique is highly suggested for larger pharmaceutical applications because of its simple, economic and easily applied on industrial scale. This method also has improved photostability characteristics and photo protective action of the selected excipients (carrier and coat) can be an alternative for the conventional tablet coat^{11, 15}. For the application of this technique ritonavir, a BCS class-II drug was selected for improved solubility and dissolution rate at the main absorption site.

MATERIALS AND METHODS:

Materials Used: Ritonavir, avicel PH 102, nanometer-sized amorphous silicon dioxide (SiO₂), transcitol HP, labrasol and labrafil, pharmaburst 500 SPI, poly ethylene glycol 400, tween 20, all other solvents are HPLC grade.

Methodology:

Preparation of Ritonavir Liquisolids: For the preparation of ritonavir liquisolids carrier/coat mixtures were prepared by using avicel PH 102 and nanometer-sized amorphous silicon dioxide with *R* equal to 20.¹⁶ The ϕ -values of the carrier and the coating materials were calculated for transcitol HP, labrasol, PEG 400, tween 20 and labrasol/labrafil 1:1 mixture. For determination of the amount of carrier required was calculated from Lf-value and while the coating material amount was calculated from *R*-value¹².

In a porcelain mortar, the drug (4mg) was manually mixed with each of the non-volatile water-miscible vehicles to prepare homogeneous mixtures of different drug concentrations are ranging from 10 to 30%. To the prepared liquid solutions, avicel PH 102 was added under continuous mixing followed

by nanometer-sized amorphous silicon dioxide to absorb the excess fluid. This mixing order was suggested to favor optimal release rate¹⁷. Finally, the prepared liquisolid mixtures were compressed on a suitable punch after addition of 10% of pharmaburst 500 SPI (as disintegrant). The constituents for each formula are listed in **Table 1**. A batch of 50 tablets for each liquisolid formula was prepared.

Evaluation of the Prepared Liquisolid Compacts: These include pre-compression & post-compression evaluation.

Pre-compression includes the evaluation of an angle of repose, compressibility index and hausner ratio was calculated to understand the flow and compression properties of prepared liquisolid mixtures. Post-compression evaluation includes weight variation, hardness, friability, content uniformity and disintegration time and all these quality control tests were applied onto 10 liquisolid tablets sample from each formula according to standardized Pharmacopeial Indian Pharmacopeia conditions.

In-vitro Dissolution Studies: The drug release rate from the prepared liquisolid formulations was investigated in the USP XXIV dissolution testing apparatus II in 900 ml of 0.1N HCl as the dissolution medium maintained at 37 ± 0.5 °C at 50 rpm. At pre-determined time intervals (5, 10, 15, 25, 35, 45, 60 and 90 min) 5 ml sample was withdrawn with replacement. The absorbance of the drug was spectrophotometrically measured at λ_{max} 235 nm after filtration on 0.45 membrane filter. The cumulative percentage of drug release was calculated using an equation obtained from a previously constructed standard calibration curve. For comparison, the dissolution rate of the drug from conventional tablets prepared using the same excipients without adding the liquid vehicle by direct compression was also determined. The mean of six determinations was considered.

Optimization of the Prepared Liquisolid Formulation: Based on the pre,post-compression & dissolution studies a formulation was selected having better drug release rate & subjected to further evaluation, it includes.

Solid State Characterizations and Compatibility

Studies: For the selected formulation, pure drug & physical mixture (without the liquid vehicle) Differential Scanning Calorimetric (DSC), Infrared Spectroscopy (IR) were performed.

Differential Scanning Calorimetric (DSC)

Studies: Samples (2.5-4 mg) were separately weighed into an aluminum pan of differential scanning calorimeter (ToshvinDSC-60, Temp Check), calibrated with purified indium standard (99.9%), and continuously purged together with a blank with nitrogen gas over a temperature range of (10-400 °C) at heating rate of 10 °C/min. The differential Scanning Calorimetric (DSC) thermograms were recorded and analyzed.

Infrared Spectroscopy (IR):

Samples (2-4 mg) were separately mixed with about 400 mg of dry potassium bromide powder and compressed into transparent disc under pressure of 10000 to 15000 pounds/inch²; and scanned in the range of 4000-500 cm⁻¹ at ambient temperature using IR spectrophotometer (Bruker, Model-Alpha,

Germany) their IR spectra were recorded and analyzed.

RESULTS AND DISCUSSION: According to the compositions shown in **Table 1**, 15 different ritonavir containing liquisolid formulations were prepared as per the previously mentioned method. In this liquisolid mixture was varied from 10-30%. In addition to the commonly used solvents PEG 400 and tween 20, versatile non-ionic liquid surfactants (namely, transcitol HP, labrasol and labrasol/ labrafil 1:1 mixture) were applied as non-volatile water-miscible vehicles to prepare different liquisolid formulations of higher drug release rate.

As per Khanfar *et al.*, It was proven that the drug release rate from the liquisolid mixture depends mainly on the drug solubility in the applied water miscible liquid vehicle¹⁸. So, selection of the previously mentioned non-ionic surfactants as liquid vehicles was based on their high solubilization power as described by their high HLB values¹⁹.

TABLE 1: COMPOSITION OF RITONAVIR LIQUISOLID FORMULATIONS

Formulation	Vehicle Type	LDC	Lf	Q	q	DA (mg)	UDW (mg)
F1	Transcutol HP	10	0.243	165	8.25	26.75	200.24
F2		20		83	4.15	13.375	100.768
F3		30		55	2.75	6.687	64.68
F4	Labrasol	10	0.202	198	9.9	26.75	234.85
F5		20		99	4.95	13.375	117.527
F6		30		66	3.3	6.687	76.189
F7	PEG 400	10	0.104	385	19.25	26.75	431.1
F8		20		193	9.65	13.375	216.12
F9		30		128	6.4	6.687	141.19
F10	Tween 20	10	0.142	282	14.1	26.75	322.99
F11		20		141	7.05	13.375	161.57
F12		30		94	4.7	6.687	105.52
F13	Labrasol/Labrofil (1:1)	10	0.151	265	13.25	26.75	305.15
F14		20		133	6.65	13.375	153.176
F15		10		89	4.45	6.687	100.28

LDC-Liquid Drug Concentration, Lf (Liquid Load Factor) = W/Q, W= Weight of the drug solution, Q=Weight of the Carrier, q=weight of the coating material, DA-Disintegrant amount (mg), UDW=Unit dose Weight.

Pre-Compression Characterization: To get a uniform product, the powder mixture must have uniform flow into the dies²⁰. The angle of repose, compressibility index, and Hausner ratio are the quantitative parametric indicators to describe the powder flow. In the present research, all prepared liquisolid mixtures can be considered of acceptable flow properties, where the angle of repose ranging from 29.47° to 31.98°, percentage of compressibility of maximum 18.98% and Hausner

ratio values close to unity **Table 2**. As the good flow is the large barricade and also a principal target in the liquisolid technique, these results indicate the accuracy of the basic calculated parameters including carrier and coat Ø-values for used non-volatile water-miscible vehicles and Lf-value at the selected carrier/coat ratio.

Post-Compression Characterization of the Prepared Ritonavir Liquisolid Tablets: The

above **Table 3** describes the different post-compression parameters of prepared ritonavir liquisolid tablets. In this, the content uniformity & hardness of all 15 formulations were found to be 96.98–100.32% & 3.87-5.32 kg/cm² respectively and these values are within the Pharmacopeial limits with small acceptable weight variation. pharmaburst 500 SPI (Co-processed blends of mannitol, starch, crospovidone, croscarmellose

sodium, colloidal silica, and silica) was applied as an efficient superdisintegrant of high drug compatibility, superior organoleptic properties, high solubility and very rapid disintegration due to combined wicking, swelling and elastic characteristics of croscarmellose sodium and crospovidone content. This could clearly explain the rapid disintegration of all prepared liquisolid tablets (72–120 s)²¹.

TABLE 2: PRE-COMPRESSSION CHARACTERIZATION OF RITONAVIR LIQUISOLID FORMULATIONS

Formulation	The angle of repose in degrees (°)	Carr's Index (%)	Hausner Ratio
F1	30.22±0.89	16.12±0.92	1.21±0.42
F2	29.56±0.76	16.23±0.87	1.31±0.51
F3	30.78±0.12	17.39±0.62	1.56±0.73
F4	31.76±0.81	18.91±0.51	1.78±0.92
F5	30.12±0.08	17.56±0.24	1.65±0.42
F6	32.56±0.32	16.21±0.91	1.32±0.76
F7	31.98±0.64	18.98±0.74	1.10±0.11
F8	30.34±0.11	17.71±0.29	1.72±0.46
F9	30.21±0.22	16.32±0.43	1.91±0.43
F10	29.91±0.37	16.29±0.61	1.96±0.65
F11	30.87±0.89	18.11±0.72	1.89±0.71
F12	31.91±0.72	17.83±0.32	1.42±0.12
F13	31.72±0.63	16.82±0.11	1.22±0.27
F14	30.23±0.67	16.53±0.79	1.54±0.32
F15	29.47±0.92	16.43±0.92	1.76±0.11

All the values are represented in mean ± S.D & n=3.

TABLE 3: POST-COMPRESSSION CHARACTERIZATION OF RITONAVIR LIQUISOLID FORMULATION TABLETS

Formulation	Avg. weight (mg) ± SD	Thickness (cm)	Diameter (cm)	Friability (%)	Hardness (kg/cm ²)	Content Uniformity (%) ± SD	Disintegration time (Sec)
F1	202.24±0.34	0.23	0.3	0.62	4.12	97.53±0.14	105±0.32
F2	104.56±0.43	0.13	0.2	0.61	4.64	98.64±0.23	97±0.98
F3	63.68±1.56	0.22	0.1	0.72	4.87	97.98±0.35	110±0.21
F4	230.75±1.32	0.26	0.4	0.73	4.32	97.23±0.56	120±0.41
F5	119.52±0.54	0.17	0.2	0.57	4.89	98.31±0.67	100±0.87
F6	75.19±0.78	0.16	0.1	0.59	3.98	99.34±0.38	98±0.23
F7	435.16±0.92	0.23	0.8	0.55	4.23	98.89±0.76	95±0.92
F8	210.12±0.62	0.28	0.3	0.65	5.17	97.12±0.87	92±0.76
F9	145.39±1.74	0.11	0.2	0.71	5.32	96.98±1.27	112±0.11
F10	324.89±1.89	0.19	0.7	0.74	4.64	97.98±0.92	116±0.71
F11	160.67±1.53	0.21	0.3	0.67	4.79	97.86±1.34	118±0.39
F12	109.32±0.83	0.12	0.2	0.69	4.59	97.94±0.95	114±0.73
F13	302.25±0.23	0.18	0.7	0.54	3.98	99.56±0.32	76±0.53
F14	154.18±0.51	0.25	0.2	0.68	3.89	100.32±0.98	72±0.74
F15	105.22±0.34	0.24	0.2	0.66	3.87	99.98±0.87	74±0.96

All the values are represented in mean ± S.D & n=3.

Dissolution Data of the Prepared Ritonavir Liquisolid Tablets: All prepared ritonavir liquisolid tablet formulations showed higher drug release rates in comparison to ritonavir powder and the conventional tablets up to 90 min, where percentage drug dissolved reached only 34.6% and 49.8% for ritonavir powder and conventional

tablets, respectively while this percentage exceeded 80% and up to 100% from the prepared liquisolid tablets within the same time. Further, investigation of the dissolution data showed that the drug release rate from liquisolid tablets varied with the applied liquid vehicle type and concentration. The percentage of drug released after 90 min from

liquisolid formulae containing tween 20 (F10–F12) as liquid vehicle was 84.6, 82.4.7 and 80.2% while the percentage was 92.4, 88.6 and 86.2% from formulae containing PEG 400 (F7–F9) at drug concentration of 10, 20 and 30%, respectively. Liquisolid formulations containing transcuto HP,

labrasol and labrasol/labrafil (1:1) mixture as liquid vehicles showed higher drug release rate, where the percentage of drug released reached 90-95% within 60 min at all drug concentrations, and this could be related to the high solubilizing power of these vehicles.

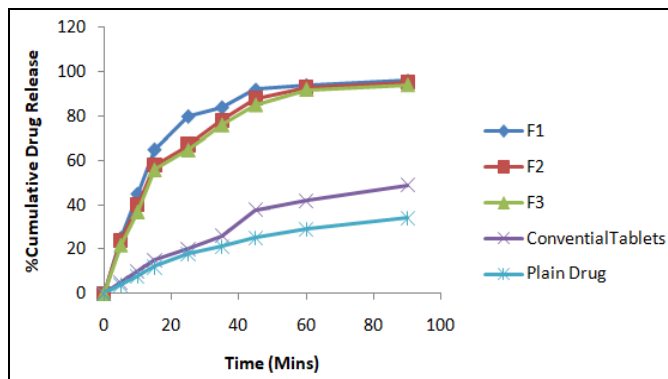


FIG. 1: DISSOLUTION PROFILES OF F1-F3 FORMULATIONS, CONVENTIONAL TABLETS AND PLAIN DRUG IN 0.1N HCl

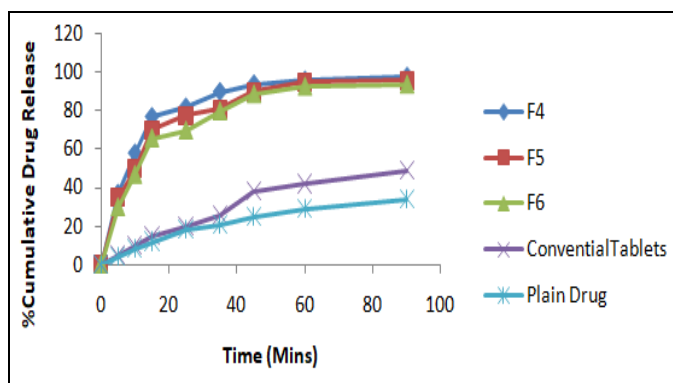


FIG. 2: DISSOLUTION PROFILES OF F4-F6 FORMULATIONS, CONVENTIONAL TABLETS AND PLAIN DRUG IN 0.1N HCl

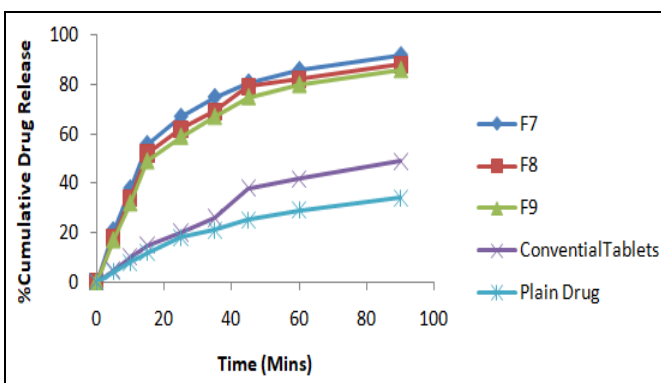


FIG. 3: DISSOLUTION PROFILES OF F7-F9 FORMULATIONS, CONVENTIONAL TABLETS AND PLAIN DRUG IN 0.1N HCl

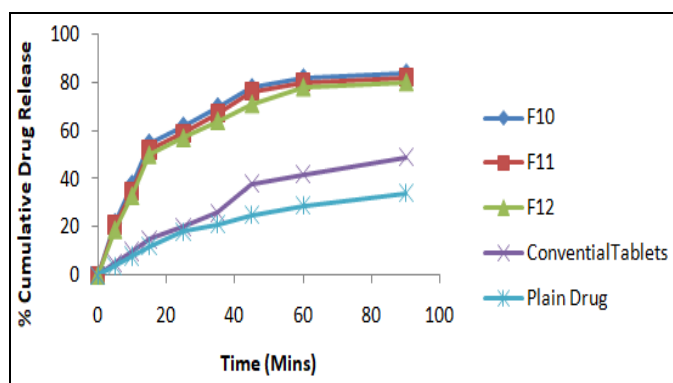


FIG. 4: DISSOLUTION PROFILES OF F10-F12 FORMULATIONS, CONVENTIONAL TABLETS AND PLAIN DRUG IN 0.1N HCl

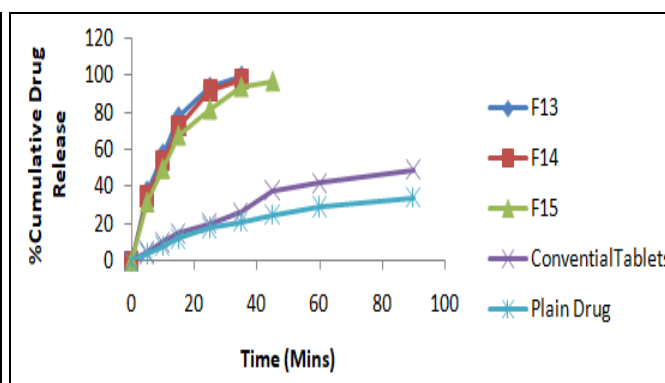


FIG. 5: DISSOLUTION PROFILES OF F13-F15 FORMULATIONS, CONVENTIONAL TABLETS AND PLAIN DRUG IN 0.1N HCl

This could clearly explain the high increase in drug dissolution rate from these formulae and also describes the effect of vehicle concentration, where the percentage of drug released from transcuto HP containing liquisolid formulation (F1–F3) was 85.8, 88.6 and 92.8% in comparison to 94.2, 90.1

and 94.2% from labrasol containing formulation (F4–F6) in only 45 min at drug concentration of 30, 20 and 10%, respectively. Labrasol based formulations usually suffer drug precipitation on dilution within aqueous media, this precipitation process decreases the amount of drug available for

absorption which is a principal cause of decreasing the oral bioavailability, especially for poorly water-soluble drugs. Mixtures of labrasol with other

pharmaceutical additives including lipids, surfactants, and co-solvents significantly inhibit and retard this precipitation process on dilution²².

TABLE 4: COMPARISON OF DISSOLUTION PROFILES OF OPTIMIZED LIQUISOLID FORMULATION (F13), CONVENTIONAL TABLETS (DCTS) AND PLAIN DRUG

Time (Min)	% Cumulative drug release (%CDR)		
	F13 (Optimized liquisolid compact formulation)	Conventional tablet (DCTS)	Plain drug
0	0	0	0
5	38.9±0.54	5.9±0.83	4.8± 0.81
10	58.8±0.21	10.9±0.74	8.7± 0.65
15	78.8±0.96	15.7±0.97	12.9±0.43
25	94.3±0.67	20.6±0.54	18.4±0.76
35	100.1±0.76	26.8±0.34	21.7±0.83
45	-	38.5±0.39	25.6±0.34
60	-	42.7±0.11	29.5±0.77
90	-	49.8±0.34	34.6±0.76

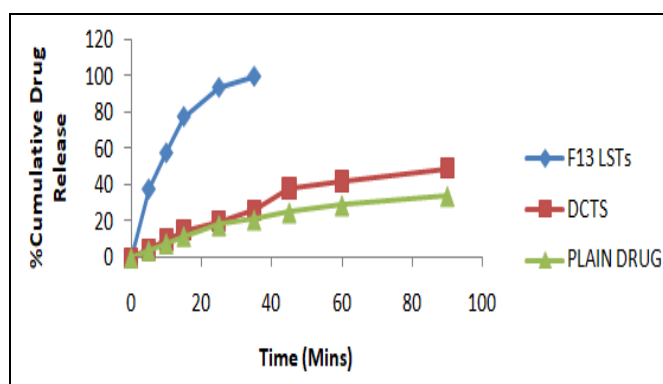


FIG. 6: COMPARISON OF DISSOLUTION PROFILES OF OPTIMIZED LIQUISOLID FORMULATION (F13), CONVENTIONAL TABLETS (DCTS) AND PLAIN DRUG

So in this work, labrasol/labrafil (1:1) mixture was applied as a liquid vehicle of high a powerful solubilizing power and higher drug tolerating ability. Results showed that the drug was completely dissolved in 45 min and the percentage of drug released was 94.5, 98.8 and 100.1% from labrasol/ labrafil (1:1) containing formulae (F15–F13) as liquid vehicle after only 35 min at a drug concentration of 30, 20 and 10%, respectively.

Solid-State Characterization: Results of solid-state characterization of the selected liquisolid tablet formula (F13) are summarized below.

Evaluation by FT-IR Spectra: Drug and excipient interactions in the liquisolid compacts were evaluated by FT-IR spectral study **Fig 7**. The FTIR spectra of ritonavir indicated characteristic peaks at 3,358 cm^{-1} (N–H stretching amide group), 2,963 cm^{-1} (C – H stretching hydrogen bonded acid within the molecule), 1,714 cm^{-1} (C =O), 1,640 cm^{-1} (1⁰ and 2⁰ Amines), 1,621 cm^{-1} , 1525 cm^{-1} (–C=C– stretching aromatic carbons).

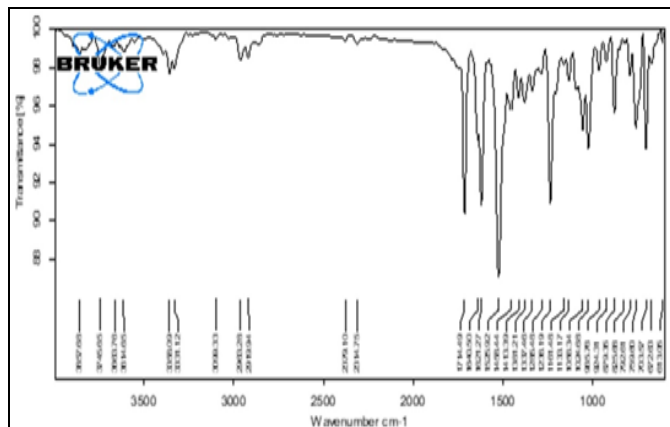
The spectra of ritonavir with all formulation excipients and liquisolid compacts tested also showed the above characteristic peaks. The FT-IR study revealed that the characteristic peaks of ritonavir were present in all the formulations and hence it indicates that there is no interaction between drug and excipients.

Evaluation by Differential Scanning Calorimetry: DSC was used to evaluate drug-excipient interactions in the liquisolid compacts. The DSC thermograms of ritonavir & optimized formulation are shown in **Fig. 8**.

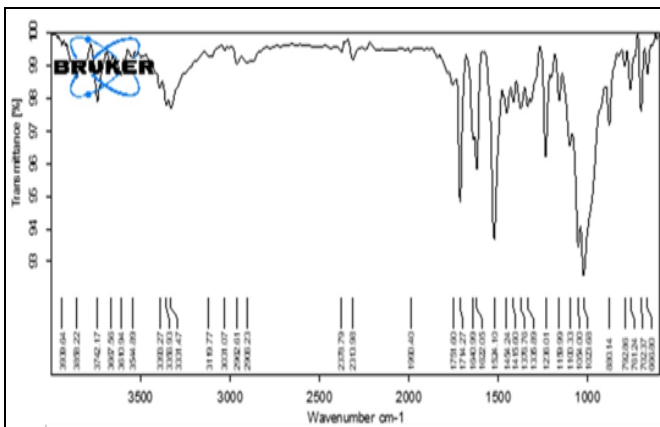
The DSC thermogram of ritonavir showed a sharp endothermic melting peak at 125.4 °C indicating the purity of the used drug sample. In physical mixture sample, this peak retained in the same position with lesser intensity (mainly due to drug dilution within the mixture) indicating compatibility of the mixture. The DSC thermogram of the selected liquisolid formula mixture (F13) showed a complete disappearance of the drug

melting peak and decrease in melting enthalpy to be an evidence of loss of drug crystallinity mainly due to dissolution within the prepared formula mixture in an amorphous form, the absence of any additional peaks within the thermogram insures absence of any interaction and chemical compatibility within the liquisolid formula mixture.

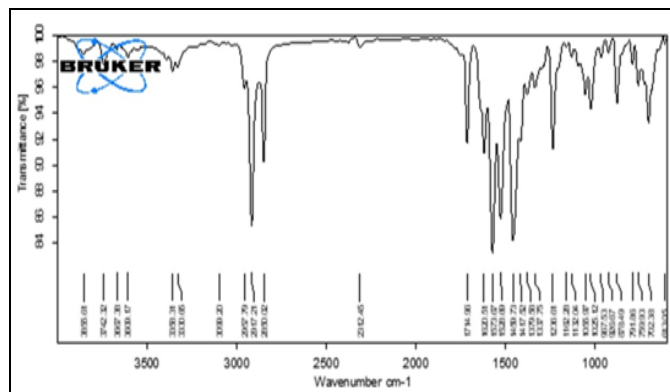
The melting peaks observed were in good agreement with the literature value of ritonavir. The melting peaks were observed at the same temperature for the pure drug as well as its liquisolid compacts. The FTIR and DSC studies, thus, indicated no interaction between ritonavir and excipients.



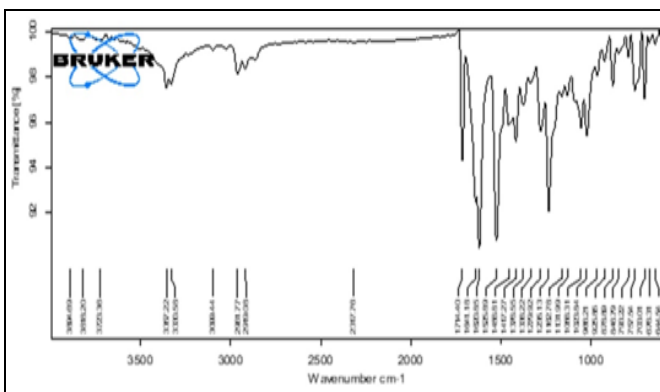
A. FT-IR SPECTRUM OF RITONAVIR PURE DRUG



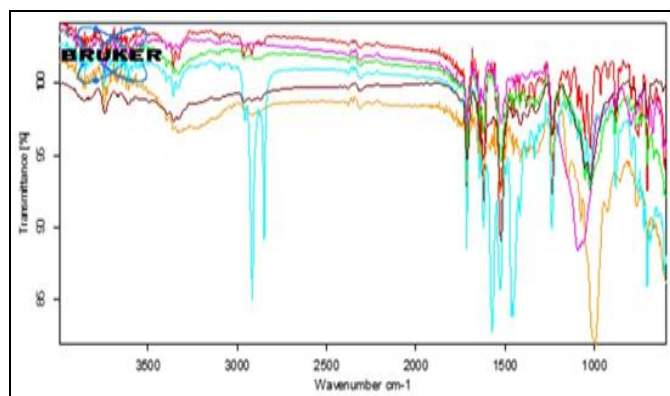
B. FT-IR SPECTRUM OF RITONAVIR + AVICEL PH 102



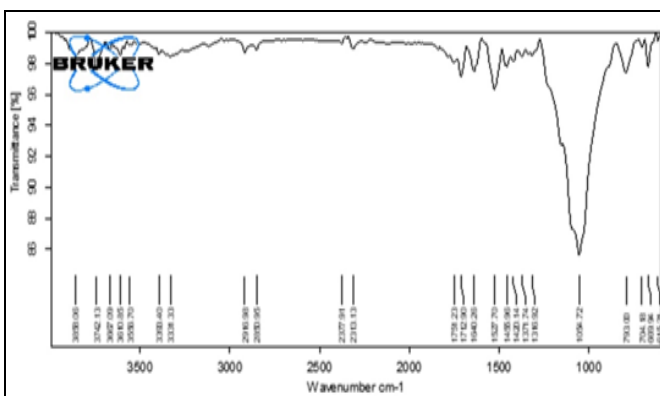
C. FT-IR SPECTRUM OF RITONAVIR + NANOMETER SIZED AMORPHOUS SILICON DIOXIDE



D. FT-IR SPECTRUM OF RITONAVIR + PHARMABURST 500 SPI

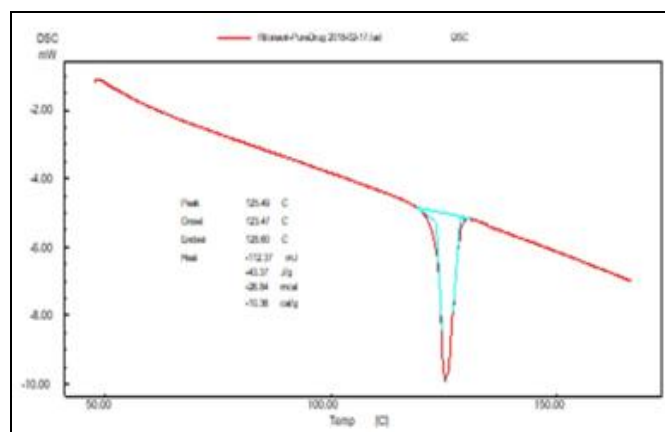


E. FT-IR SPECTRUM OF RITONAVIR WITH ALL EXCIPIENTS

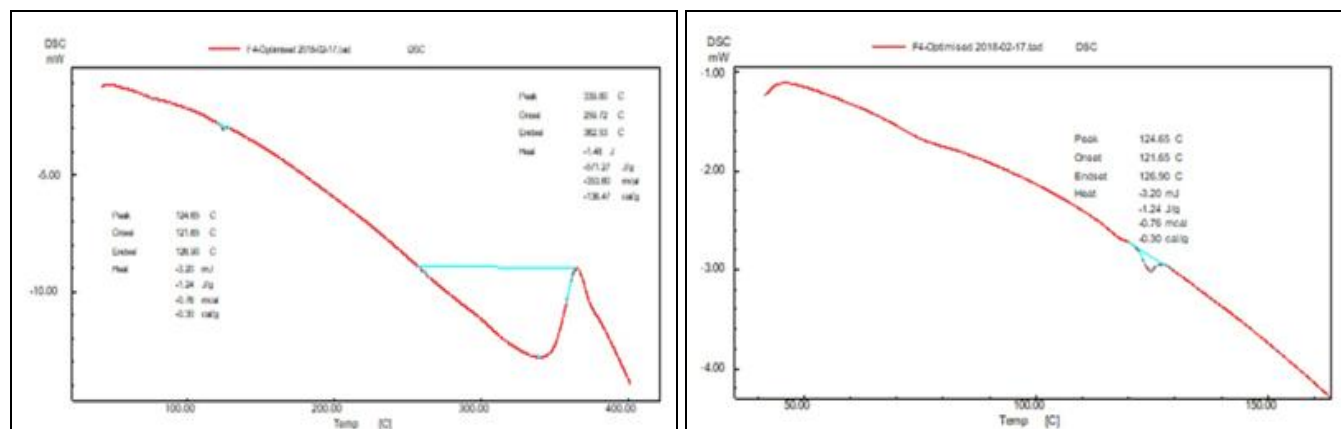


F. FT-IR SPECTRUM OF RITONAVIR OPTIMIZED FORMULATION

FIG. 7: FT-IR SPECTRUMS OF A) RITONAVIR PURE DRUG B) RITONAVIR + AVICEL PH 102 C) RITONAVIR + NANOMETERSIZED AMORPHOUS SILICON DIOXIDE D) RITONAVIR + PHARMABURST 500 SPI E) RITONAVIR + ALL EXCIPIENTS F) RITONAVIR OPTIMIZED FORMULATION



A. DSC THERMOGRAMS OF RITONAVIR PURE DRUG



B. DSC THERMOGRAMS OF RITONAVIR OPTIMIZED FORMULATION

FIG. 8: DSC THERMOGRAMS OF A) RITONAVIR PURE DRUG B) OPTIMIZED FORMULATION

CONCLUSION: Liquisolid technique is simple and efficient method to prepare a solid dosage form of high drug solubility with good pre and post characterization properties. Labrasol/labrafil (1:1) mixture is an optimum vehicle to prepare compatible ritonavir liquisolid tablets of high dissolution rate at the main absorption site.

The prepared labrasol/ labrafil (1:1) based liquisolid formula at 10% drug concentration showed rapid dissolution, and the complete drug release (100%) occurred only within 35 min. All these data led to the conclusion that solubility increase of BCS class II drugs at their main absorption site significantly increases their bioavailability.

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CONFLICT OF INTEREST: The authors report no conflict of interest.

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