E-ISSN: 0975-8232; P-ISSN: 2320-5148



# PHARMACEUTICAL SCIENCES



Received on 09 May 2019; received in revised form, 01 October 2019; accepted, 13 November 2019; published 01 March 2020

# PHYTOCHEMICAL SCREENING AND ANTIOXIDANT ACTIVITY OF LEAF EXTRACT PREPARED BY SOXHLET EXTRACTION OF SELECTED MEMBERS OF APOCYNACEAE

Annika Maria Paul \* and Fr. Jobi Xavier

Department of Life Sciences, CHRIST (Deemed to be University), Bengaluru - 560029, Karnataka, India.

# **Keywords:**

Spectrophotometer, Antioxidant property, Phytochemistry, Bovine serum albumin, Vinca

# Correspondence to Author: Annika Maria Paul

Research Scholar, Department of Life Sciences, CHRIST (Deemed to be University), Bengaluru - 560029, Karnataka, India.

E-mail: anikampaul0007@gmail.com

ABSTRACT: The present study aimed to understand, phytochemistry and antioxidant property of five members belonging to Apocynaceae, Catharanthus roseus (L) G. Don; Plumeria alba L.; Pentalinon luteum (L) B. F. Hanson & Wunderlin; Rauvolfia tetraphylla L and Nerium oleander L. The preliminary phytochemical analysis was carried out using the methanolic extract for protein, carbohydrate, alkaloids, phenol, tannin flavonoids, and saponin. They were quantified using spectrophotometer using different standards and methods like the Bradford method using BSA as standard for protein, phenol-sulphuric acid method using glucose as standard for carbohydrates, Folin-Ciocalteau method by Sidduraju and Becker for phenol using catechol as standard, tannins using Folin-Ciocalteu method using tannic acid as standard, Zhishen et al., modified method for flavonoids using quercetin as standard. Total alkaloids were estimated using the dry weight method. Antioxidant properties were studied using the ability of the Soxhlet extracts of the plant samples to scavenge the DPPH radicals using Blois method with modifications. The highest concentration for protein was found in Rauvolfia tetraphylla, and the lowest concentration was found in Nerium oleander. The highest concentration of carbohydrate and tannin was found in Catharanthus roseus and the lowest in Nerium oleander. The highest alkaloid, phenol, and flavonoid content were found among in Catharanthus roseus, Plumeria alba and the lowest in Nerium oleander. Nerium oleander exhibited the highest scavenging properties for two of the Soxhlet concentrations out of three (the highest antioxidant activity for 200 µg/mL,  $100 \mu g/mL$ ).

**INTRODUCTION:** Plants play an essential role in sustaining life on earth. It plays a diverse role like providing timber for construction and infrastructure; it provides valuable medicines to humanity, food for humans and animals, gum, resins, *etc*.



DOI:

10.13040/IJPSR.0975-8232.11(3).1290-95

This article can be accessed online on www.ijpsr.com

**DOI link:** http://dx.doi.org/10.13040/IJPSR.0975-8232.11(3).1290-95

So, the forest cover has immense aesthetic value, and the forests serve as the lungs for the supply of oxygen and also sustain millions of animals, insects, birds, and other creatures <sup>1</sup>.

This study aims to compare some selected members of the family Apocynaceae for their phytochemical compounds and antioxidant properties. Groups of secondary plant metabolites, antioxidant phenolics, and flavonoids are commonly found in various parts of the plants like fruits, leaves, and seeds <sup>2</sup> and they have been shown to provide a fruitful tool to compare the different plants in a family for their taxonomic evaluation.

Apocynaceae Five members of include Catharanthus roseus (L) G. Don (Authentication SMPU/RARIMD/BNG2018-19/1232) is a native of Madagascar and is commonly called as the Madagascar periwinkle. It is an annual shrub that grows up to a height of 1 m and is rich in Alkaloids like vincristine and vinblastine which medicinally relevant and used in the treatment of cancer. Plumeria alba L. (Authentication SMPU/RARIMD/BNG2018-19/1232) is a small tree or shrub which grows up to a height of 2-8m and is commonly called a Pagoda tree which is native to Central America.

These plant species are generally grown as an ornamental plant for its beautiful and fragrant flowers, but the extract obtained from the leaves and root are used in the treatment of syphilitic ulcers <sup>3</sup>. *Pentalinon luteum* (L) B. F. Hanson & Wunderlin (Authentication SMPU/RARIMD/BNG 2018-19/1232) is the wild allamanda which is a vine-like shrub which grows up to a height of 1.8 to 2.4 m, and is commonly called as the hammocks vipers' tail is a native plant Caribbean islands, Florida and USA <sup>4</sup>.

It is mainly used in the treatment of skin diseases like sores and cuts. Rauvolfia tetraphylla L. (Authentication SMPU/RARIMD/BNG2018-19/ 1232) is a small woody shrub that grows up to 0.6-1.2 m high and is commonly called as the devil pepper. It is a highly medicinal plant that is used for treating insomnia, reducing the level of blood sugar; the root extract is used in the treatment of diarrhea, dysentery, and cholera. Nerium oleander L. (Authentication SMPU/RARIMD/BNG2018-19/1232) is a very large evergreen shrub that grows up to 2-6 m and is highly poisonous. It is widespread in its distribution and found all over the world, including China, India, Japan, and the Mediterranean to Persia. It is highly poisonous but used for skin inflammations and chronic joint pains and cramps <sup>5</sup>.

Plants are rich in biologically active compounds like antioxidants and phytochemicals, and they play an essential role in the treatment of diseases <sup>2</sup>. Many diseases like cancer, inflammation, lesions, degenerative diseases are caused due to excess of Reactive Oxygen Species (ROS) in the cellular body. All aerobic organisms have an inbuilt

mechanism to neutralize this ROS but often insufficient as there is an increase in the rate of pollution in the environment and our changing food habits, which consist more of junk foods. Even though there are a large amount of synthetic antioxidants that are used in the food processing technologies, most of them have side effects that increase the demand for naturally occurring antioxidants in plants <sup>6</sup>. Secondary metabolites like phenol and flavonoids play an important in plant growth as well as reproduction. These compounds also have antioxidant properties as well as anticancer properties <sup>7</sup>.

# MATERIALS AND METHODOLOGY:

Plant Sample: The plant samples were collected from Christ (Deemed to be University) Main Campus, Bangalore, Karnataka, and was identified and authenticated from Regional Ayurveda Research Institute for Metabolic Disorders by Dr. V Rama Rao (Authentication SMPU/RARIMD/BNG2018-19/1232). The leaves were taken separately, washed, and shade dried for one week. They were ground into a fine powder using a mechanical blender and sieved and stored in an airtight container.

Instrumentation: Spectrophotometer (SHIMADZU, UV-1800. Measurement Wavelength Range: 190-1100 nm, Photometric range: 4-4 Abs. Light source 20W Halogen Lamp, Deuterium Lamp, Detector: Silicon photodiode, Gas chromatography-Mass Spectrometer (GC-MS)-SHIMADZU, GCMS-QP2010SE. Temp Range- 50-350 °C, Mass Analyser: Quadrupole Mass, Range: 1.5 to 1000 m/column: SH-Rxi<sup>TM</sup>-5Sil MS-Low-polarity phase.

Preparation of Plant Extract: 2 g of the sample was extracted in a Soxhlet apparatus using 150ml of methanol as the solvent, and the Soxhlet was run overnight until the sample loaded became colorless. After the Soxhlet extraction is over, extracts were put inside rotor evaporator until it gave the solidified residues. The sample extracts were collected, weighed and stored in a freezer.

#### **Preliminary Phytochemical Analysis:**

**Protein:** One mL of sample was taken to which two drops of 0.2% of ninhydrin solution was added and was kept in a boiling water bath till a purple color was formed <sup>1</sup>.

# **Carbohydrate:**

**Molisch's Test:** 2 ml of sample was taken in a test tube to which 2 ml of Molisch's reagent was added and shaken. After which drop by drop, Concentrated H<sub>2</sub>SO<sub>4</sub> was added through the sides until there was the appearance of a violet ring confirm the presence of carbohydrate <sup>1</sup>.

**Alkaloid:** 1 ml of the extract was taken to which few drops of HCl were added to make it more acidic to which add a few drops of Wagner's reagent. The presence of orange-brown precipitate confirmed the presence of alkaloids <sup>8</sup>.

**Saponins:** 0.5 g of the sample was taken in a boiling test tube to which 3 ml of water was taken and continuously shaken for 2-3 min. Positive results showed foam production for which is persistent <sup>8</sup>.

**Flavonoids:** 1 ml of plant extract was treated with sodium hydroxide solution. The positive result showed the presence of an intense yellow color and became colorless after the addition of dilute acid <sup>8</sup>.

**Phenol:** 2 ml of plant extract were treated with FeCl<sub>3</sub> positive result shows the presence of deep blue color indicates the presence of phenol <sup>8</sup>.

# **Quantitative Estimation of Phytochemicals:**

**Protein Estimation Using Bradford:** Modified Bradford's method was used.0.5 g of leaf sample was ground using motor and pestle, and it was centrifuged at 10000 rpm for 10 min 4 °C. 0.5 ml of the supernatant was taken and made up to 5 mL, and 5 mL of Bradford's reagent was added to this and incubated for 10 min until the formation of a blue color. The absorbance was measured at 595 nm by plotting against a standard graph made up of BSA as standard. The estimation was carried out in triplicates, and the Protein content was expressed in milligram/gram of dried sample of the leaves <sup>9</sup>.

Carbohydrate Estimation: Carbohydrate was estimated using a slightly modified sulphuric acid phenol method. 100 mg of the plant sample was taken in a boiling tube, and it was hydrolyzed in 5 mL of 2.5 N HCl for 3 h and is allowed to cool at room temperature. After which it is neutralized with solid sodium carbonate until the brisk effervescence is ceased. Then this was made up to 50 mL using distilled water and centrifuge. 1 mL of

the supernatant was taken to which 1 mL of phenol was added and 5 ml of sulphuric acid. 10 min it was shaken continuously and kept for incubation for 20 min. The absorbance was measured at 490 nm using glucose as standard. The estimation was carried out in triplicates, and the carbohydrate content was expressed in milligram/gram of dried sample of the leaves.

**Total Alkaloid Estimation:** 1g of powdered leaf sample was taken and 40 mL of 20% acetic acid and was allowed to stand for 4 h. The mixture after filtration was reduced to  $1/4^{th}$  of its volume by placing it in a water bath. To this ammonium hydroxide was added drop-wise inside a fume hood. The solution was made to stand for 2-3 h until all the filtrate settles. The precipitate was filtered, dried, and weighed. The percentage of total alkaloid content was calculated as  $^2$ .

Percentage of total alkaloids (%) = Weight of residue  $\times$  100 / Weight of the sample taken

Estimation of Total Phenol: Phenol in the Soxhlet sample was estimated using a modified Folin-Ciocaltaeu assay. 200 μl of the sample (10mg/ml) was taken in a test tube and made up to 1 ml with distilled water to which 1 ml of Folin Ciocalteu's reagent was added and incubated for 6 min in room temperature and 1.25 mL of 7% NaCO<sub>3</sub> and incubate in the dark for 90 min. The absorbance was measured at 638 nm by plotting against a standard graph made up of catechol as standard. The estimation was carried out in triplicates, and the Phenol content was expressed in mg/g of dried sample of the leaves <sup>2</sup>.

Estimation of Total Flavonoids: 0.5mL of Soxhlet sample was taken in a test tube. 1.5 mL of ethanol was added and mixed thoroughly. To this 0.1 mL of 10%, aluminum nitrate solution was added and to which 0.1 mL of 1 M potassium acetate was added, and 2 mL of water was added. Blank was prepared similarly.

Standard was prepared using a different concentration of quercetin, and the absorbance was measured at 415 nm using a spectrophotometer. The estimation was carried out in triplicates, and the flavonoid content was expressed in milligram/gram of dried sample of the leaves <sup>2,11</sup>.

Estimation of Tannin: 0.5 mL of the soxhlet extract is taken in a test tube and make it up to 1 mL using distilled water. To this add 0.5 mL of Folin Ciocalteu's reagent and add 2.5 mL of Sodium Carbonate (20 %) mix thoroughly and incubate for 40 min. Read the absorbance at 725 nm using the spectrophotometer. Tannic acid prepared at different concentrations from the stalk solution was used as the standard. The estimation was carried out in triplicates, and the tannin content was expressed in milligram/gram of dried sample of the leaves <sup>2,12</sup>.

#### **Estimation of Antioxidant Activity:**

*In-vitro* Antioxidant Activity using DPPH: Modified Blois method was used to estimate the antioxidant activity. 0.2 mmol/L solution of DPPH in methanol was prepared, and 500 μL of this solution was added to different concentrations of the extract (50-150 μg/mL). This solution was vigorously shaken and allowed to stand in the dark for 30 min. The blank was prepared with all the reagents except the sample. Absorbance was measured at 517 nm using a spectrophotometer <sup>10</sup>,

E-ISSN: 0975-8232; P-ISSN: 2320-5148

# **RESULTS:**

TABLE 1: PRELIMINARY QUALITATIVE PHYTOCHEMICAL ANALYSIS OF METHANOLIC SOXHLET LEAF EXTRACT OF VARIOUS MEMBERS OF APOCYNACEAE

S. no.	Chemical compound	Plumeria	Vinca	Rauvolfia	Pentalinon	Nerium
1	Protein	+++	++	+++	+	+
2	Carbohydrate	+++	+++	++	++	+
3	Alkaloid	+++	+++	+++	+	+
4	Phenol	+++	+++	+++	++	+
5	Flavonoid	+++	+++	+++	++	+
6	Tannin	++	+++	+++	++	++
7	Saponin	-	+++	-	-	-

TABLE 2: QUANTITATIVE ANALYSIS FOR PROTEIN, CARBOHYDRATE AND ALKALOID AMONG FIVE DIFFERENT MEMBERS OF APOCYNACEAE EXPRESSED (mg/g) OF LEAF SAMPLE AND  $\pm$  STANDARD DEVIATION

Sample	Protein Concentration	Carbohydrate Conc. (mg/g) of	Total Percentage of Alkaloid in 1 g of
	(mg/g) of leaf	leaf	leaf
Plumeria	$0.67 \pm 0.025$	$1.56 \pm 0.021$	$0.30\% \pm 0.04$
Rauvolfia	$0.72 \pm 0.025$	$1.33 \pm 0.015$	$0.29\% \pm 0.04$
Vinca	$0.35 \pm 0.025$	$2.41 \pm 0.076$	$0.33\% \pm 0.05$
Pentalinon	$0.27 \pm 0.305$	$0.66 \pm 0.097$	$0.16\% \pm 0.015$
Nerium	$0.12 \pm 0.025$	$0.23 \pm 0.092$	$0.10\% \pm 0.025$

TABLE 3: QUANTITATIVE ANALYSIS FOR PHENOL, FLAVONOID, AND TANNIN AMONG FIVE DIFFERENT MEMBERS OF APOCYNACEAE EXPRESSED (mg/g) OF LEAF SAMPLE AND ± STANDARD DEVIATION

Sample	Concentration of Phenol	The concentration of Flavonoid	The concentration of Tannin (mg/g)	
~ <b>p</b>	(mg/g) of leaf	(mg/g) of leaf	of leaf	
Plumeria	$1.3 \pm 0.015$	$0.45 \pm 0.02$	$1.75 \pm 0.07$	
Rauvolfia	$1.26 \pm 0.20$	$0.42 \pm 0.04$	$3.43 \pm 0.35$	
Vinca	$1.36 \pm 0.015$	$0.44 \pm 0.02$	$4.36 \pm 0.42$	
Pentalinon	$0.82 \pm 0.087$	$0.32 \pm 0.03$	$1.73 \pm 0.55$	
Nerium	$0.54 \pm 0.02$	$0.074 \pm 0.01$	$1.46 \pm 0.30$	

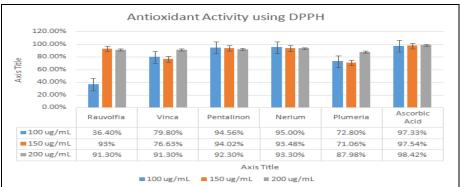


FIG. 1: THE DIFFERENT CONCENTRATION OF ANTIOXIDANT ACTIVITY IN THREE DIFFERENT CONCENTRATIONS OF THE LEAF SAMPLE OF AMONG VARIOUS MEMBERS OF APOCYNACEAE

**DISCUSSION:** Phytochemical analysis of leaves of five members of Apocynaceae was carried out successfully with methanol and was tested for protein, carbohydrate, alkaloid, phenol, flavonoid, tannin, and saponin and the results were found to be positive for protein, carbohydrate, alkaloid, phenol, flavonoid and tannin. These results agreed with the earlier studies reported on the plants which proved that these phytochemical compounds were present in the selected plants in the present work <sup>14, 15, 16, 17</sup>. The phytochemical compounds that showed positive results were quantified and the results were compared among the five members belonging to the family Apocynaceae for protein, carbohydrate, alkaloid, phenol, flavonoid, and tannin.

The results compared with the earlier studies carried out by the scientists and found a similar response as we found in the present work <sup>18</sup>. Variations in the qualitative range may have occurred due to the change the climacteric factors and location and the amount of sunlight *etc*. that is received by the plant. The highest concentration for protein was found in *Rauvolfia tetraphylla*, and the lowest concentration was found in *N. oleander*. The highest concentration of carbohydrate and tannin was found in *Catharanthus roseus* and the lowest in *Nerium oleander*. The highest alkaloid, phenol, and flavonoid content were found among in *Catharanthus roseus*, *Plumeria alba* and the lowest in *Nerium oleander*.

Antioxidant activity was estimated based on the scavenging property of free radicals using DPPH. Nerium oleander exhibited the highest scavenging properties for two of the Soxhlet concentrations out of three (the highest antioxidant activity for 200  $\mu$ g/mL, 100  $\mu$ g/mL). With reference to the previous works done it was found that Nerium oleander has high antioxidant properties <sup>6</sup>. Although, most of the members showed a very high antioxidant property; hence, we can conclude that the members of the family Apocynaceae have very high antioxidant scavenging properties.

**CONCLUSION:** The phytochemical analysis carried out concluded that the selected five members of Apocynaceae are rich in various phytochemicals and antioxidant scavenging properties. Hence, these plants which were often used for ornamental purposes can be exploited in metabolic

engineering for the production of secondary metabolites.

E-ISSN: 0975-8232; P-ISSN: 2320-5148

**ACKNOWLEDGEMENT:** We are thankful to the Vice-chancellor and Management of CHRIST (Deemed to be University), Bangalore for their encouragement to carry out this study.

**CONFLICTS OF INTEREST:** The authors have no conflict of interest.

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#### How to cite this article:

Paul AM and Xavier FJ: Phytochemical screening and antioxidant activity of leaf extract prepared by Soxhlet extraction of selected members of Apocynaceae. Int J Pharm Sci & Res 2020; 11(3): 1290-95. doi: 10.13040/IJPSR.0975-8232.11(3).1290-95.

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