(Research Article)

#### IJPSR (2020), Volume 11, Issue 6



INTERNATIONAL JOURNAL

Received on 23 July 2019; received in revised form, 22 November 2019; accepted, 29 February 2020; published 01 June 2020

# DEVELOPMENT OF ANALYTICAL METHOD AND VALIDATION OF NADOLOLIN PURE AND PHARMACEUTICAL FORMULATIONS USING UV-SPECTROPHOTOMETRY AND SPECTROFLUORIMETRY

V. Veeramanikandan<sup>1</sup>, R. Arun<sup>2</sup> and A. Antonsmith<sup>\*1</sup>

Department of Pharmacy<sup>1</sup>, Annamalai University, Annamalai Nagar - 608002, Tamil Nadu, India. Department of Pharmacy<sup>2</sup>, Karpagam University, Pollachi Main Road, Coimbatore - 641021, Tamil Nadu, India.

#### **Keywords:**

Nadolol, Spectrophotometry, Spectrofluorimetry, Determination Correspondence to Author:

### Dr. A. Antonsmith

Department of Pharmacy, Annamalai University, Annamalai Nagar - 608002, Tamil Nadu, India.

E-mail: auantonsmith@yahoo.co.in

**ABSTRACT:** The development of analytical methods is in need for the estimation of drugs in pure and different pharmaceutical formulations. A simple, sensitive, rapid, accurate, precise and economic spectrophotometric and spectro-fluorimetric method was developed and validated for Nadolol in pure and pharmaceutical formulations. The proposed method was validated according to the International Conference on Harmonization (ICH) guidelines. The wave length ( $\lambda_{max}$ ) used for the estimation of Nadolol is 267 nm by spectrophotometry, excitation ( $\lambda_{Ex}$ )-267 nm and emission ( $\lambda_{Em}$ )-300 nm by spectrofluorimetry. The linearity of the calibration curve was validated by the high values of the correlation coefficient of regression. The percentage of drugs recovered 100.37 ± 0.94% and 99.9 ± 0.59% for spectrophotometric and spectrofluorimetric methods respectively. LOD and LOQ values for Nadolol were found to be 3.531 µg/ml and 10.70 µg/ml by spectrophotometry and 0.45 µg/ml and 1.37 µg/ml by spectrofluorimetry. The developed methods are simple and suitable for the determination of Nadolol in pure and pharmaceutical preparations.

**INTRODUCTION:** Nadolol, chemically is (2R, 3S)-5{[(2R)-3-tert-butylaminol-2-hydroxy-propyl] oxy} - 1, 2, 3, 4-tetrahydro naphthalene-2,3-diol **Fig. 1** is a non-selective  $\beta$ -blocker which is official in BP<sup>-1</sup> and USP<sup>-2</sup> used in the treatment of high blood pressure and chest pain. It has a preference for beta-1 receptors, which are predominantly located in the heart, thereby inhibiting the effects of catecholamines and causing a decrease in heart rate and blood pressure, inhibition of beta-2 receptors, which are mainly located in the bronchial smooth muscles of the airways leads to airway constriction similar to that seen in Asthma.



Review of literature reveals that only a few methods like UV <sup>3-5</sup>, Colorimetry <sup>6-7</sup>, Fluorimetry <sup>8</sup>, HPLC <sup>5, 9-11</sup>, Biological fluids using HPLC <sup>12</sup>, UHPLC–MS <sup>13</sup>, and LC-MS <sup>14</sup>, were developed for the determination of Nadolol in pure and Pharmaceutical preparation.



FIG. 1: STRUCTURE OF NADOLOL

### **MATERIALS:**

**Instrument:** Absorption spectral measurements were carried out with a Systronics 2202 UV-Visible spectrophotometry, fluorescence spectra measurements were carried out with a Perkin Elmer LS 55 spectrofluorimetry, for sonication Branson 2510 sonicator was used.

**Chemicals:** Nadolol tablets (40 mg) were procured from Walmart Pharmacy 2051 Strachan Road S., Medicine Hat T1B OG4 from Canada. Hydrochloric acid was of AR grade from Nice Pharmaceuticals Pvt. Ltd., and in house produced distilled water was used. Nadolol working standard was obtained as a gift sample from a Pharma Industry, Industrial estate, Ahmedabad.

# **METHODS:**

**Preparation of Stock Solution:** 25 mg of Nadolol was accurately weighed and transferred to a 25 ml volumetric flask. About 10 ml of water was added, vortexed for about 5 min. The volume was made up to 25 ml and mixed well with water to obtain a final concentration of 1 mg/ml.

Determination of Absorbance Maxima and Fluorescence Maxima: An appropriate aliquot portion of 2 ml of Nadolol from a standard stock solution of Nadolol was transferred to 100ml volumetric flask, mixed with water and the volume was made up to 100 ml with water to obtain the concentration 20  $\mu$ g/ml of Nadolol. Drug solutions were further diluted necessarily and scanned in spectrophotometry and spectrofluorimetry to determine the absorbance maxima and based on absorbance maxima, emission maxima were determined.

**Validation of the Proposed Method:** The proposed method was validated according to the International Conference on Harmonization<sup>15</sup>.

# Linearity and Range:

**Spectrophotometry:** An appropriate aliquot portion of 2, 2.5, 3, 3.5, 4 and 4.5 ml of Nadolol from standard stock solution of Nadolol were transferred to 100 ml volumetric flask, mixed with water and the volumes were made up to 100 ml with water to obtain concentrations 20, 25, 30, 35, 40 and 45  $\mu$ g/ml of Nadolol. The absorbance of all the resulting solutions was measured at 267 nm. The calibration curve was constructed by plotting drug concentration versus absorbance obtained.

**Spectrofluorimetry:** An appropriate aliquot portion of 1, 2, 3, 4 and 5 ml of Nadolol from standard stock solution of Nadolol were transferred to 100 ml volumetric flask, mixed with water and the volumes were made up to 100 ml with water to obtain the concentrations 1, 2, 3, 4 and 5  $\mu$ g/ml of

Nadolol. The fluorescence of the resulting solutions with water was measured at excitation ( $\lambda_{Ex}$ )-267 nm, emission ( $\lambda_{Em}$ )-300 nm.

**Precision:** 10 tablets were weighed accurately and crushed into a fine powder using glass mortar and pestle. An accurately weighed quantity of tablet powder equivalent to 25 mg of Nadolol transferred into to 25 ml volumetric flask. 10ml of water was added and sonicated for 5 min, the volume was made up to 25 ml with water, mixed well and filtered it. For spectrophotometric method, 1 ml of the solution of the filtrate was pipetted out and transferred into 25 ml of volumetric flask and the volume was made up to 50 ml with water so that the final solution concentration will be 20  $\mu$ g/ml. The absorbance of the resulting solution was measured at 267 nm.

For the spectrofluorimetric method, 1 ml of a solution of the filtrate was pipetted out and transferred into 100 ml of volumetric flask and the volume was made up to 100 ml with water. From the above solution, 2 ml was pipetted out and transferred into a 10 ml volumetric flask and made up the volume up to 10 ml with water so that the final solution concentration will be 2  $\mu$ g/ml. The fluorescence intensity of the resulting solution was measured at excitation ( $\lambda$ <sub>Ex</sub>)-267 nm, emission ( $\lambda$ <sub>Em</sub>)-300 nm using water as a solvent blank.

# Accuracy:

**Preparation of Stock Solution:** The first step is the preparation of the stock solution (500 mg of pure drug of Nadolol was dissolved in 25 ml of water).

# Spectrophotometry & Spectrofluorimetry:

Accuracy for 50%: 206 mg of tablet powder weighed accurately (equivalent to 40 mg of Nadolol) and transferred into three different 25 ml volumetric flask. 10ml of water was added and 1ml of the stock solution which contains 20 mg/ml of Nadolol. The solution was sonicated for 3 min, made up the volume up to 25 ml with water. The resulting solutions were filtered separately, and from the filtrate, 1 ml of the solution was pipette out and transferred to a 100 ml volumetric flask and made up the volume up to 100 ml with water. It was repeated for three times of three different weighings. Accuracy for 100%: 206 mg of tablet powder weighed accurately (equivalent to 40 mg of Nadolol) and transferred into a 25 ml volumetric flask. 10ml of water and 2 ml of the stock solution, which contain 40 mg/ml of Nadolol was added. The solution was sonicated for 3 min and made up the volume up to 25 ml with water. The resulting solutions were filtered and from the filtrate 1 ml was pipetted out transferred into 100 ml volumetric flask and made up the volume up to 100 ml with water. It was repeated for three times of different weighing.

Accuracy for 150%: 206 mg of tablet powder weighed accurately (equivalent to 40 mg of Nadolol) and transferred into a 25 ml volumetric flask. 10 ml of water and 3 ml of the stock solution, which contain 20 mg/ml of Nadolol was added. The above solutions were sonicated for 3 min and made up the volume up to 25 ml with water. The resulting solutions were filtered and from the filtrate 1 ml was pipetted out transferred to 100 ml volumetric flask and make up the volume up to 100 ml with water. It is repeated for three times of different weighing.

For Spectrophotometry the absorbance of all the resulting solutions was measured at 267 nm and for Spectrofluorimetry the intensity of fluorescence of all the resulting solutions was measured at excitation ( $\lambda_{Ex}$ )-267 nm and emission ( $\lambda_{Em}$ )-300 nm using water as solvent blank.

**Limit of Detection:** The detection limit of an individual analytical procedure is the lowest amount of analyte in a sample which can be detected but not necessarily quantitated as an exact value. Detection limit (DL) may be expressed as:

$$DL = 3.3 \sigma / S$$

Where,  $\sigma$  = the standard deviation of the response, S = the slope of the calibration curve.

The slope 'S' is estimated from the calibration curve of the analyte. The estimation of  $\sigma$  was carried out using a calibration curve.

**Limit of Quantitation:** The quantitation limit of an individual analytical procedure is the lowest analyte in a sample, which can be quantitatively determined with suitable precision and accuracy. The quantitation limit is a parameter of quantitative assays for low levels of compounds in sample matrices and is used particularly for the determination of impurities or degraded products.

$$QL = 10 \sigma / S$$

**Ruggedness:** Ruggedness is a measure of reproducibility of test results under normal, expected operational conditions from laboratory to laboratory and from analyst to analyst. Ruggedness is determined by the analysis of aliquots by different analysts. 20  $\mu$ g/ml and 2  $\mu$ g/ml solutions were prepared and analyzed using spectrophotometer and spectrofluorometer respectively.

**Robustness:** The robustness of an analytical procedure is a measure of is capacity to remain unaffected by small, but deliberate variations in method parameters and provides an indication of its reliability during normal usage. It was carried out by changing the wavelength by one nm at 266 nm and 269 nm for spectrophotometry and emission wavelength at 299 nm and 301 nm respectively.

# **RESULTS AND DISCUSSION:**

**Determination of Absorption Maximum:** Absorbance maxima were determined in the spectro-photometer by taking 20  $\mu$ g/ml Nadolol drug which is dissolved in water and scanned from 200-400 nm using UV-Visible Spectrophotometer. The absorption spectra presented in **Fig. 2.** It was found that the absorption maximum of 267 nm was identified in the spectra.



FIG. 2: UV-SPECTRUM OF NADOLOL WITH WATER

**Determination of Emission Maxima:** Absorbance maxima were determined in spectrofluorimetry by taking 3  $\mu$ g/ml Nadolol drug which is dissolved in water and excitation maxima were fixed at 267 nm. Since the wavelength shows the highest absorbance

the emission spectra were scanned from 270-500 nm. The emission spectrum is presented in **Fig. 3**. It was found that an emission maximum 300nm was identified in the spectra.



FIG. 3: FLUORIMETRY SPECTRUM OF NADOLOL USING WATER

Linearity and Range: In spectrophotometry Calibration standards for Nadolol covering a range of 20-45  $\mu$ g/ml were prepared in serial dilutions that were made with water. The absorbance of all resulting concentrations was measured at 267 nm. The graph between the concentration and absorbance was plotted. The regression equation was found to be y = 0.004x - 0.005. The correlation coefficient (R<sup>2</sup>) of the standard curve was found to be 0.985. The obtained data are presented in Table 1 and the calibration graph is presented in Fig. 4 respectively.

 TABLE 1: LINEARITY AND RANGE OF NADOLOL

 USING SPECTROPHOTOMETER

S. no.	Concentration (µg/ml)	Absorbance
1	20	0.089
2	25	0.112
3	30	0.135
4	35	0.157
5	40	0.181
6	45	0.207



FIG. 4: CALIBRATION GRAPH OF NADOLOL USING SPECTROPHOTOMETER

In spectrofluorimetry calibration standards for Nadolol covering a range of 1 to 5  $\mu$ g/ml were prepared in serial dilutions that were made with water. The intensity of fluorescence of all the resulting concentrations was measured at excitation ( $\lambda_{Ex}$ )-267 nm and emission ( $\lambda_{Em}$ )-300 nm.

The data is presented in **Table 2** and the graph between the concentration and absorbance was plotted in **Fig. 5**. The regression equation was found to be y = 130.9x + 6.5. The correlation coefficient ( $\mathbb{R}^2$ ) of the standard curve was found to be 0.99.

 TABLE 2: LINEARITY AND RANGE OF NADOLOL

 USING SPECTROPHOTOMETER

S. no.	Concentration	Intensity of
	(µg/ml)	Fluorescence
1	1	151
2	2	237
3	3	418
4	4	532
5	5	658



FIG. 5: CALIBRATION GRAPH OF NADOLOL USING SPECTROFLUOROMETER

**Precision:** The method was carried out as described. The results were presented in **Tables 3** and **4** for spectrophotometry and spectrofluorimetry respectively.

It was found that the amount of drugs found to be present in the sample was found to be  $100.25 \pm 0.20$  and  $100.65 \pm 0.035$  for spectrophotometry **Table 3** and spectrofluorimetry **Table 4**, respectively.

The values obtained in the repeatability (precision) shows that there is no significant difference in the precision value. Hence, that developed method can be used to analyte the Nadolol in tablet formulation. There is no evidence of interference of excitation with Nadolol.

S. no.	Weight of the tablet powder(mg)	Absorbance	Drug content present (mg)	Percentage found (%)
1	206.05	0.103	40.40	101
2	206.04	0.102	40.04	100.1
3	206.01	0.101	39.71	99.27
4	206.03	0.102	40.11	100.24
5	206.02	0.103	40.50	101.25
6	206.04	0.101	39.72	99.3
	Μ	ean		100.25
	S	.D		0.207
	R	SD		0.002

#### TABLE 3: PRECISION STUDY OF NADOLOL USING SPECTROFLUOROMETER

The mean precision vale was found to be  $100.19 \pm 0.828\%$ . The value was obtained from 99.3 to 101.25% by the spectrophotometric method.

S. no.	Weight of the tablet	Intensity of	Drug content	Percentage
	powder (mg)	fluorescence	present (mg)	found (%)
1	206.03	231.02	40.33	100.82
2	206.01	230.19	40.15	100.37
3	206.02	230.91	40.31	100.77
4	206.04	229.97	40.15	100.37
5	206.03	232.04	40.50	101.25
6	206.05	230.14	40.15	100.37
		Mean		100.65
		S.D		0.0357
		RSD		0.003

# TABLE 4: PRECISION STUDY OF NADOLOL USING SPECTROFLUOROMETER

The mean precision vale was found to be 100.65  $\pm$  0.0357%.

The value was obtained from 100.37 to 100.82% by the spectrofluorimetric method.

S.	Percentage	Sample	Drug in the	Pure drug	Total drug	Absorbance	Amount	Amount	Percentage
no.	level	weight (mg)	tablet	added	content		found	Recovered	recovery
			powder (mg)	(mg)	(mg)		(mg)		(%)
1	50	206.05	40	20	60	0.231	60.14	20.14	100.7
2	50	206.06	40	20	60	0.232	60.39	20.39	101.95
3	50	206.04	40	20	60	0.230	59.98	19.88	99.4
4	100	206.01	40	40	80	0.308	80.2	40.2	100.5
5	100	206.03	40	40	80	0.306	79.67	39.67	99.17
6	100	206.04	40	40	80	0.308	80.18	40.19	100.47
7	150	206.02	40	60	100	0.385	100.25	60.25	100.6
8	150	206.05	40	60	100	0.383	99.71	60.71	99.27
9	150	206.01	40	60	100	0.386	100.51	60.51	101.27
				Mean					100.37
				SD					0.940
				RSD					0.009

Accuracy: From the data drug-excipients interactions and drug-solvent interactions have not been found. Hence, there is no interference of any component with the drug has been proved. The

percentage of recovery was found to be  $100.37 \pm 0.940\%$ . The value was obtained from 99.27 to 101.95% by the spectrofluorimetric method. The resulting data is presented in **Table 5**.

#### TABLE 6: ACCURACY STUDY OF NADOLOL USING SPECTROFLUORIMETER

S.	Percentage	Sample	Drug in the	Pure drug	Total drug	Intensity of	Amount	Amount	Percentage
no.	level	weight (mg)	tablet	added	content	fluorescence	found	Recovered	recovery
			powder (mg)	( <b>mg</b> )	(mg)		(mg)		(%)
1	50	206.01	40	20	60	301.04	59 .99	19.99	99.97
2	50	206.07	40	20	60	303.12	60.37	19.37	100.92
3	50	206.04	40	20	60	302.13	60.21	19.21	100.52
4	100	206.05	40	40	80	402.54	80.11	40.11	100.27

Veeramanikandan et al., IJPSR, 2020; Vol. 11(6): 2962-2968.

E-ISSN: 0975-8232; P-ISSN: 2320-5148

5	100	206.03	40	40	80	400.56	79.83	39.83	99.57
6	100	206.05	40	40	80	401.23	79.95	39.95	99.87
7	150	206.01	40	60	100	499.26	99.51	59.51	99.18
8	150	206.06	40	60	100	499.13	99.46	59.46	99.1
9	150	206.02	40	60	100	501.02	99.86	59.86	99.76
				Mean					99.90
				SD					0.596
				RSD					0.005

The percentage of recovery was found to be 99.90  $\pm$  0.596%. The value was obtained from 99.1 to 100.92% by the spectrophotometric method. The resulting data is presented in **Table 6**.

**LOD and LOQ:** The LOD was found to be 3.53  $\mu$ g/ml and the LOQ concentration was found to be 10.7  $\mu$ g/ml with water by spectrophotometric method.

The LOD was found to be 0.45  $\mu$ g/ml and the LOQ concentration was found to be 1.37 $\mu$ g/ml with Water by the spectrofluorimetric method.

**Ruggedness:** Ruggedness data is presented in **Table 7** for spectrophotometric and spectro-fluorometric methods.

TABLE /: RUGGEDNESS OF NADOLOL					
S. no.	Analyst -1 Analyst-2				
	Absorbance				
1	0.089	0.085			
2	0.087	0.087			
3	0.088	0.086			
Intensity of fluorescence					
1	237	238			
2	236	237			
3	237	236			

 TABLE 7: RUGGEDNESS OF NADOLOL

**Robustness:** There is no significant difference in absorbance and fluorescence observed when the minor changes like the one-nanometer difference in spectrophotometer and spectrofluorometer.

TABLE 8: VALIDATION PROFILE OF NADOLOLWITH WATER

Parameters	Spectrophotometry	Spectrofluorimetry
Linearity range	20-45	1-5
(µg/ml)		
Precision (%)	$100.25 \pm 0.20\%$ .	$100.65 \pm 0.035\%$
Accuracy (%)	$100.37 \pm 0.94$ %.	$99.90 \pm 0.59\%$
50%	$100.68 \pm 1.27$	$100.47 \pm 0.46$
100%	100.04±0.75	99.90±0.35
150%	$100.38 \pm 1.01$	99.34±0.50
LOD (µg/ml)	3.53	0.45
LOQ (µg/ml)	10.7	1.37

Validation Profile: Performing replicate analysis of the standard solutions was used to assess the

accuracy and precision and reproducibility of the proposed methods. The selected concentration within the calibration range was prepared in water and analyzed with the relevant calibration curves to determine the intra-day and inter-day variability. The validation profile of Nadolol with water is presented in **Table 8** by spectrophotometric and spectrofluorimetric methods.

**CONCLUSION:** Spectrophotometric and spectrofluorimetric method for quantifying Nadolol in pure and formulation has been developed and validated. The developed method is precise, accurate and linear over the concentration range from 20-45 µg/ml and 1-5 µg/ml for spectrophotometric and spectrofluorimetric method respectively. The precision was found to be 100.25  $\pm 0.2\%$  and 100.65  $\pm 0.03\%$  for spectrophotometry and spectrofluorimetry respectively.

The percentage of drugs recovered  $100.37 \pm 0.94\%$ and  $99.9 \pm 0.59\%$  for spectrophotometric and spectrofluorimetric methods respectively. The LOD and LOQ were found to be  $3.53 \ \mu\text{g/ml}$  and  $10.70 \ \mu\text{g/ml}$  for spectrophotometry and LOD and LOQ were found to be  $0.45 \ \mu\text{g/ml}$  and  $1.371 \ \mu\text{g/ml}$  for spectrofluorimetry with water. Among the two developed methods, the spectrofluorimetric method is highly sensitive than the spectrophotometric method. These methods are simple and suitable for the determination of Nadolol in pure and pharmaceutical preparations.

**ACKNOWLEDGEMENT:** The authors are thankful to the Department of Pharmacy, Annamalai University, to carry out the study.

**CONFLICT OF INTEREST:** The authors have declared no conflicts of interest.

### **REFERENCES:**

1. British Pharmacopoeia, The Department of Health and Social Services and Public Safety, British Pharmacopoeial Commission 2008; 2: 1504.

- United States Pharmacopeia and National Formulary (USP 30-NF 25). Rockville, MD: United States Pharmacopeia Convention 2007; 2: 2694.
- 3. Olajire A, Olakunle SI and Olaniyi AA: A new spectrophotometric method for the determination of Nadolol. Journal of the Iranian Chemical Society 2006; 3(3): 277-84.
- 4. Vijayalakshmi R, Ramya NS and Dhanaraju M: Method development for quantification of oxidation complexes of Nadolol and Resveratrol by visible spectrophotometry. International Journal of Pharmacy and Pharmaceutical Sciences 2014; 7(1): 304-07.
- Camila T, Pedro LG, Fabio PG, Erika RMKH and Maria IRMS: Quantitative determination of Nadolol in tablets by High-Performance Liquid chromatography and UVderivative spectrophotometry. Analytical Letters 2008; 41(3): 424-36.
- 6. Ivashkiv E: Colorimetric determination of Nadolol in tablets. J of Pharmaceutical Scis 1978; 67(7): 1024-25.
- Amin AS, Ragab GH and Saleh H: Colorimetric determination of beta blockers in pharmaceutical formulations. J of Pharmaceutical Analysis 2002; 30(4): 1347-53.
- Eugene I: Fluorimetric determination of Nadolol in human serum and urine. Journal of Pharmaceutical Sciences 1977; 66(8): 1168-72.
- 9. Perlman S, Szyper M and Kirschbaum JJ: Highperformance liquid chromatographic analysis of nadolol and bendroflumethiazide combination tablet formulations. Journal of Pharmaceutical Sciences 1984; 73(4): 259-61.

- Patel BR, Joel JK and Raymond BP: High-pressure liquid chromatography of nadolol and other β-adrenergic blocking drugs. Journal of Pharmaceutical Sciences 1981; 70(3): 336-38.
- 11. Chandana and Manepalli: Method development and validation for simultaneous estimation of Nadolol and Bendroflumethiazide in pharmaceutical dosage form by using RP-HPLC method. Journal of Pharmaceutical Sciences 2012; 4(1): 216-27.
- 12. Srinivas NR, Shyu WC, Shah VR, Campbell DA and Barbhaiya RH: High-performance liquid chromatographic assay for the quantitation of nadolol in human plasma using fluorescence detection. Biomedical Chromatography 1995; 9(2): 75-79.
- Magdalena O, Katarzyna D, Waldemar B, Barbara Ć, Maciej S, Grzegorz S, Juliusz and Witold D: Radio degradation of nadolol in the solid state and identification of its radiolysis products by UHPLC–MS method. Chemical Papers 2018; 72(2): 349-57.
- 14. Alaa A, Wasfi SI, Salama S, Al-Nassib M, Allawy MN and Al-Katheeri: Determination of some  $\beta$ -blockers and  $\beta$ 2-agonists in plasma and urine using Liquid Chromatography–tandem Mass Spectrometry and Solid Phase Extraction. Journal of Chromatography Sciences 2017; 55(8): 846-56.
- 15. International Conference on Harmonization of Technical Requirements for Registration of Pharmaceuticals for Human Use. Validation of Analytical Procedures: Text and Methodology, Q2 (R1), 2005.

#### How to cite this article:

Veeramanikandan V, Arun R and Antonsmith A: Development of analytical method and validation of nadololin pure and pharmaceutical formulations using uv-spectrophotometry and spectrofluorimetry. Int J Pharm Sci & Res 2020; 11(6): 2962-68. doi: 10.13040/IJPSR. 0975-8232.11(6).2962-68.

All © 2013 are reserved by the International Journal of Pharmaceutical Sciences and Research. This Journal licensed under a Creative Commons Attribution-NonCommercial-ShareAlike 3.0 Unported License.

This article can be downloaded to Android OS based mobile. Scan QR Code using Code/Bar Scanner from your mobile. (Scanners are available on Google Playstore)