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## MICROENCAPSULATION: A STRATEGY TO SURPASS PHOTO INSTABILITY AND LOW PENETRABILITY OF SKIN LIGHTENING AGENTS

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### Keywords:

Kojic acid, Arbutin,  
Hyperpigmentation,  
Microemulsion, Formulation/stability,  
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**ABSTRACT: Objective:** Kojic acid and arbutin are potent agents for effective treatment of hyperpigmentation. However, low photostability, low permeability and high irritancy exhibited by them make topical delivery a challenge. The aim of this study was to develop novel photostable and highly penetrable formulation of kojic acid and arbutin to combat hyperpigmentation effectively at lower concentrations in comparison to marketed conventional formulations. **Methods:** Kojic acid and arbutin loaded microemulsion based crème' gel was formulated by the generation of water in oil system by spontaneous formation technique followed by incorporation into a gel. Thus manufactured by simple industry feasible technique and evaluated for effective reduction of skin irritancy, improvement in photostability, the penetrability of the hydrophilic drugs and anti-tyrosinase potential. **Results:** Comparative *in-vitro* and *ex-vivo* release study showed that microemulsion based crème' gel not only provided sustained effect up to 24 h but also retained approximately 10 folds higher drug concentration of drugs in the skin as against conventional gel formulation. This novel formulation was not associated with any irritation potential and had a higher tyrosinase inhibitory effect when compared against conventional gel. Furthermore, microemulsion based crème' gel when subjected to photostability studies was found to be stable over a period of 7-8 h of sun exposure, thus allowing day time use of the formulation. **Conclusion:** Microemulsion based crème' gel can be easily used in daylight, thus, surpassing the limitation of photodegradation of actives to some highly irritant products when exposed to sunlight.

**INTRODUCTION:** Hyperpigmentation is a common aesthetically unappealing skin disorder marked by uneven localized or diffused darkening of the skin. Despite the availability of various treatment options, clinical management of hyperpigmentation presents a challenge for dermatologists.

Kojic acid and arbutin are both potent tyrosinase inhibitors used to treat hyper-pigmentation<sup>1-4</sup>. Due to the virtue of low lipophilicity (logP -0.9 and -1.35) respectively of kojic acid<sup>5</sup> and arbutin<sup>6</sup>, it presents low penetrability.

The aim of this study was to develop water in oil microemulsion based crème' gel<sup>7-10</sup> to present the individual as well as combined advantages like sustained release profile, increase penetration of the hydrophilic agents, impart photostability and reduce skin irritancy<sup>3</sup>. The study also focused on including smart excipients in formulation thus allowing reduction of drug concentration without affecting the therapeutics and also lowering the

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side effects. Kojic acid and arbutin are highly water soluble and also susceptible to photodegradation<sup>11</sup>. Therefore, dermal penetration and photostability of these drugs can be increased with the help of microemulsion systems. Kojic acid and arbutin loaded water in oil microemulsion with and without frankincense oil were evaluated for its anti-tyrosinase activity, where microemulsion with frankincense oil<sup>12, 13</sup> exhibited higher inhibitory effect as against ME without frankincense oil. The results suggest that the microemulsion based crème' gel can be an efficient topical delivery system to treat hyperpigmentation.

### **MATERIALS AND METHODS:**

**Materials:** Kojic acid and Arbutin were obtained as gift samples from Otto Chemie Pvt. Ltd. and Croda Chemicals Pvt. Ltd. respectively. Tween 80, Span 80, Crodamol GTTC-LQ-(SG) and Frankincense oil were obtained from Evonik Industries, S. D. fine chemicals, Croda Chemicals, Indobiochem Pvt. Ltd. Carbopol 974, Tego carbomer, Hydroxypropyl cellulose, were supplied by Signet Chemical Corporation and Evonik Industries. All other reagents were of analytical grade and HPLC grades as per the requirement.

### **Methods:**

#### **Water in Oil Microemulsion Based Crème' Gel:**

#### **Solubility Studies and Emulsification Ability:**

The solubility of Kojic acid and arbutin in various surfactants and co-surfactants was determined by adding an excess amount of drug into 2 ml of each vehicle, followed by shaking (100 rpm) at 25 °C for 6 h. The samples were centrifuged at 10000 rpm for 10 min to remove excess drug and the concentration of drugs in the supernatant were measured by high-performance liquid chromatography (HPLC) after appropriate dilution with ethanol. The mixture of selected surfactant and oil were vortexed and gradually aqueous phase was added to obtain a transparent microemulsion. The microemulsion was then checked for its percent transmittance to identify its homogeneity and uniformity.

#### **Construction of Pseudoternary Phase Diagrams:**

Crodamol GTTC and Frankincense oil were used as the oil phase, Tween 80<sup>TM</sup> and Span 80<sup>TM</sup> as the surfactants and deionized water as the aqueous phase.

Pseudo ternary phase diagrams were prepared with surfactant mix ratios from 1:1, 1:2,1:3, 2:1, 2:3 to 3:2 (w/w). By applying the aqueous phase titration method, distilled water was titrated drop-wise to the oil and S<sub>mix</sub> mixture and vortexed. After each addition, the mixture was observed for appearance. The endpoint of the titration was the point where the solution turns turbid or cloudy. The pseudo ternary phase diagram was constructed using CHEMIX School 3.6 Software. Clear and stable microemulsion zones were identified.

#### **Preparation of Kojic Acid and Arbutin Loaded Microemulsion (KA-ME):**

Due to the high water solubility of kojic acid and arbutin, the drugs were initially dissolved in the aqueous phase before adding of oily phase. The mixture of all constituents was further subjected to vortexing for proper mixing followed by sonication for removal of any air entrapment. The end product obtained was a transparent and uniform microemulsion formulation. The final concentration of kojic acid and arbutin in the microemulsion was 0.3% and 0.7% respectively. The blank microemulsion was obtained without the incorporation of the drugs.

#### **Preparation of Kojic Acid and Arbutin Loaded microemulsion Based Crème' Gel (KA-ME Crème' Gel):**

To improve its retention time microemulsion systems are incorporated in appropriate gelling agent to produce gels. Carbopols 974 P NF, 980 NF, 971 P NF, ETD 20 20, Ultrez 10 NF, HPMC at concentration 1% w/w and TEGO carbomer 140G at concentration 0.01% w/w were used as gelling agents. Since TEGO carbomer 140 G formed a very good gel with a pleasing appearance, thus was selected as a gelling agent in the microemulsion based crème' gel formulation.

#### **Characterization KA- ME and KA-ME Crème' Gel:**

Various microemulsion were selected after plotting the pseudo ternary phase diagram and were subjected to thermodynamic stability studies including Heating Cooling cycles, Centrifugation Cycles, and Freeze-Thaw cycles. Further, the microemulsion which did not show any separation in these studies were selected and analyzed for other parameters mentioned in the text below. Percent transmittance of the microemulsion was analysed using UV 1800 Shimadzu version 2.33 to determine transparency and uniformity of the

microemulsion. Globule size viscosity, polydispersity index, zeta potential, and particle size test: Globule size of the formulation was analyzed by DMWB1-223ASC Motic microscopy. The viscosity of each of the drug-loaded sample was tested using a DV-III Ultra programmable rheometer D220 (Brookfield Engineering Laboratories, Inc., Middleboro, MA, USA) while the particle size, polydispersity index, and zeta potential of each of the drug-loaded sample were estimated using a (Malvern Zetasizer, UK (NICOMP™ 380 ZLS; Malvern, Santa Barbara, CA, USA). The evidence of the presence of W/O microstructure in the microemulsion was confirmed with cooling thermograms using Differential scanning calorimetry DSC Q20 (TA Instrument, New Castle, DE – USA) under a dynamic nitrogen atmosphere with a flow rate of 50 mL min<sup>-1</sup> 14. The final selected microemulsion was incorporated in gel to generate an optimized kojic acid and arbutin loaded microemulsion based gel formulation which was further analyzed for the following parameters.

*In-vitro* diffusion and the *ex-vivo* study were carried out in phosphate buffer pH 5.5 placed in Franz diffusion cell (DBK instrument) using dialysis bag and human skin respectively. The release profiles were fitted in kinetic models to evaluate release kinetics. Assay content and

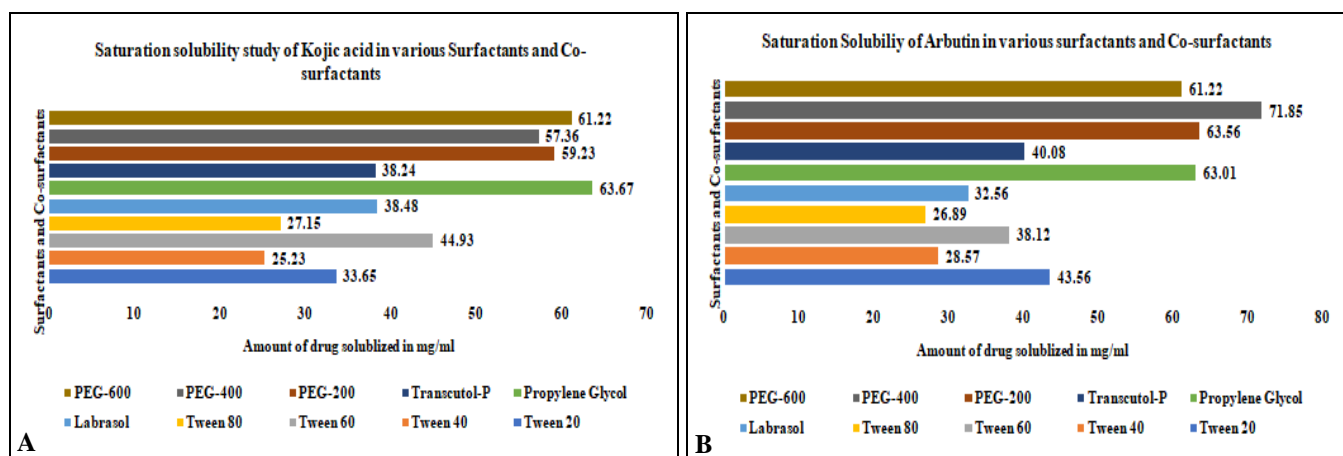
diffusion per hour release content were analyzed using the equation obtained by linear regression analysis from the calibration curve of kojic acid and arbutin by the HPLC method. *In-vitro* skin irritation study and tyrosinase activity study was performed to determine the cosmetic safety anti-hyperpigmented activity of the formulation.

**Formulation Stability Test:** The optimized microemulsion based crème' gel was subjected to stability as per ICH guidelines (ICH Q1 R<sub>2</sub>). The formulations were stored at different temperatures and humidity conditions 25 °C / 60 ± 5% RH and accelerated temperature (40 °C / 75 ± 5% RH) for a period of three months. The optimized formulations stored in amber color spray bottles were evaluated for parameters like appearance, pH, content, drug release and viscosity over a period of 3 months

## RESULTS AND DISCUSSION:

### Kojic Acid and Arbutin Loaded Microemulsion: Solubility Studies and Emulsification Ability:

The solubility of kojic acid and arbutin in the various surfactants and co-surfactants / cosolvents is shown in **Fig. 1**. To avoid leaching of the drug from the interface, which can further lead to the fast release of a drug, Tween 80 showing the least solubility for both the drugs was chosen as the co-surfactant along with Span 80 as the surfactant.



**FIG. 1: SOLUBILITIES OF KOJIC ACID (A) AND ARBUTIN (B) IN SURFACTANTS AND CO-SURFACTANTS**

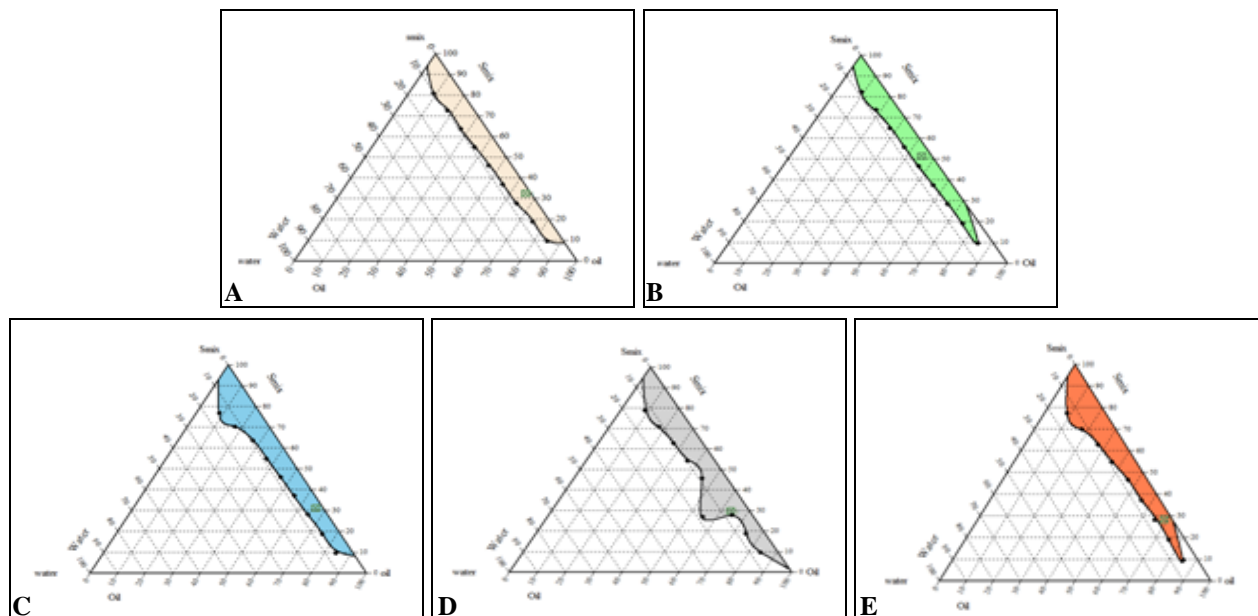
**Pseudo ternary Phase Diagrams and Preparation of Kojic Acid and Arbutin Loaded Micro-emulsion (KA-ME):** To obtain the appropriate components of oil, surfactant and co-surfactant and their concentration range for microemulsion, Pseudo ternary phase diagrams were plotted for 1:1, 1:2, 1:3, 2:1, 2:3 and 3:2 ratio

of S<sub>mix</sub> to oil. From these phase diagrams, the largest microemulsion region was observed in the 3:2 ratio of S<sub>mix</sub>. As shown in **Fig. 2E**, the area of w/o microemulsion is maximum at an S/coS ratio of 3:2. The exact composition according to water, surfactant, co-surfactant, and oil phases are shown in the table below.

**TABLE 1: THE CONTENTS OF OPTIMIZED MICROEMULSION FORMULATION**

Formulation	% Water	% Surfactant (S)	% Co-surfactant (CS)	% Oil (O)
W/O ME	5	16	24	55

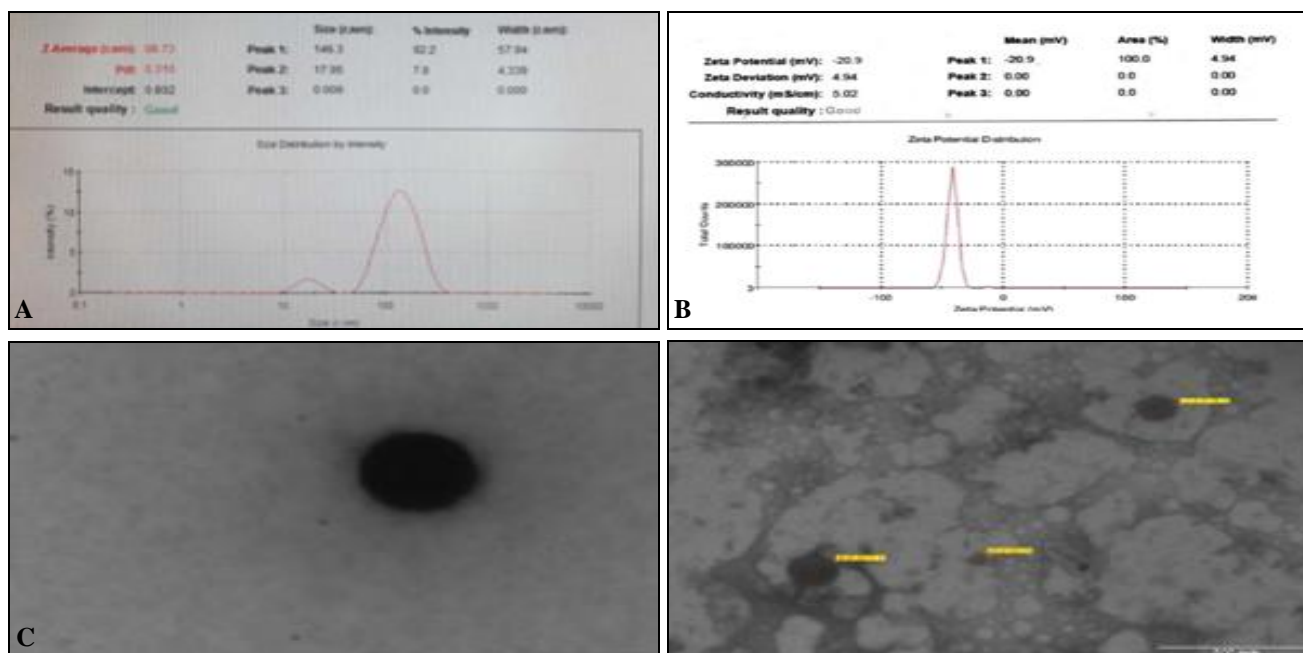
Notes: S: Span 80, CS: Tween 80, O: Crodamol GTTC-LQ-(SG) and Frankincense oil (5% of the total oil content)



**FIG. 2: PSEUDOTERNARY PHASE DIAGRAMS OF MICROEMULSION SYSTEM. (A) 1:1 S/COS RATIO, (B) 1:2 S/COS RATIO, (C) 1:3 S/COS RATIO, (D) 3:2 S/COS RATIO, (E) 2:3 S/COS RATIO**

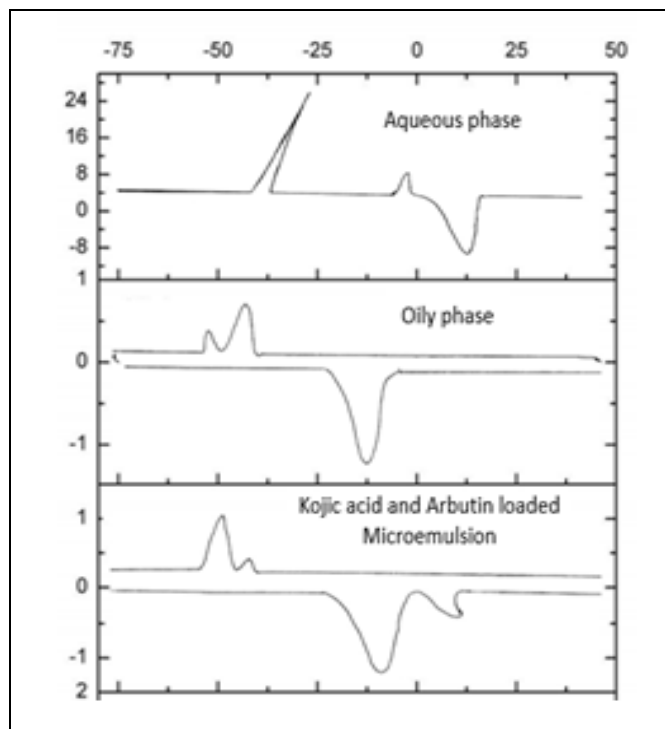
**Characterization KA-ME and KA-ME Based Crème’ Gel:** The optimized microemulsion had a viscosity of  $100.8 \pm 0.82$  cps and passed all the thermodynamic stability tests with percent transmittance of  $98.81 \pm 0.22\%$  indicating transparency and clarity. The mean globule size of the water droplets was 98.73 nm with a polydispersity index of 0.31 presenting the monodispersity in the

globule size distribution. Zeta potential of the microemulsion was found to be -20.9. This suggests that aggregation is not expected to take place due to the slightly negative charge of the droplets. Morphology, particle size, polydispersity, and zeta potential are depicted in **Fig. 3A, 3B,** and **3C.**



**FIG. 3: PARTICLE SIZE AND POLYDISPERSITY (A), ZETA POTENTIAL (B), TEM IMAGE(C)**

Analysis of DSC cooling curves of an aqueous phase, oil phase, and Drug loaded microemulsion are shown in **Fig 4**. Thermogram showed a sharp exothermic peak at  $-26.05\text{ }^{\circ}\text{C}$  corresponding to the freezing process and an endothermic peak at  $8.32\text{ }^{\circ}\text{C}$  corresponding to the melting process. The thermogram of the oil phase showed two exothermic peaks at  $-46.23\text{ }^{\circ}\text{C}$  and  $-53.15\text{ }^{\circ}\text{C}$  and one endothermic peak at  $-11.56\text{ }^{\circ}\text{C}$ . In kojic acid and arbutin loaded microemulsion exothermic peak was observed at  $-43.51\text{ }^{\circ}\text{C}$  and  $-51.85\text{ }^{\circ}\text{C}$  and endothermic peak at  $-11.33\text{ }^{\circ}\text{C}$  and  $-1.96\text{ }^{\circ}\text{C}$ . Based on the thermal analysis, it was seen that the microemulsion thermogram did not show the presence of a water peak, while only the exothermic and endothermic peaks of oil were observed. The absence of a water peak in the microemulsion system suggested that the aqueous phase has bound water molecules strongly interacting with the surfactant layer, and this phenomenon altered the freezing point to a very low temperature which is not detected. This result suggested that the aqueous phase corresponding to the internal phase and oil was the external phase of the system, characterizing a w/o microstructure type.



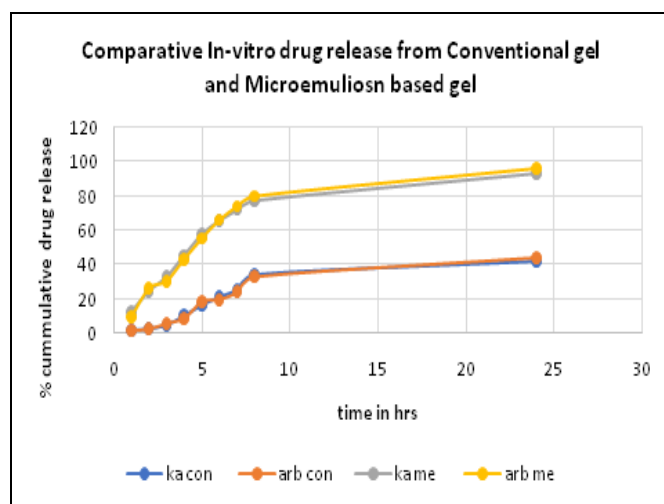
**FIG. 4: DSC COOLING/FREEZING THERMOGRAMS OF AQUEOUS PHASE, OILY PHASE AND DRUG LOADED MICROEMULSION**

**Comparative *in-vitro* and *ex-vivo* Release Study:**  
A comparative *in-vitro* and *ex-vivo* release study

was conducted for conventional gel, microemulsion based crème' gels are shown in **Fig. 5A** and **5B**. The *in-vitro* kojic acid /arbutin release from microemulsion based crème gel and conventional gel were  $77.35 \pm 2.33 / 80.15 \pm 1.03$  and  $34.14 \pm 0.056\% / 32.89 \pm 0.67$  respectively. The *ex-vivo* release of kojic acid /arbutin from microemulsion based crème gel and conventional gel were  $75.125 \pm 0.60 / 74.03 \pm 1.88$  and  $29.04 \pm 0.438 / 26.23 \pm 3.88$  respectively. From the above results, it is seen that conventional gel shows much lower and sustained release as compared to kojic acid and arbutin loaded microemulsion based crème' gel.

Further, the skin retention study revealed the actual situation. The amount of drugs retained at the end of 24 h from the formulations on to the human skin (used in the *ex-vivo* study) was  $2.13 \pm 0.23$  and  $29.45 \pm 1.56$  in the case of conventional gel and micro-emulsion based crème' gel respectively. Thus, it can be concluded that the microemulsion based crème' gel provided sustained release as well as better retention of a drug as against conventional gel which showed false sustained effect as it had high chances of getting wiped off soon after application without any retention of the gel on the skin surface.

The sustained release of drugs from the microemulsion maybe by the virtue of the resistance provided by the oily layer, which affects the release of the water-soluble drugs from the internal phase into the release medium. The kinetic study reports suggest that microemulsion based crème' gel follows the Higuchi square root law, indicating sustained effect.



**FIG. 5A: COMPARATIVE *IN-VITRO* RELEASE STUDIES**

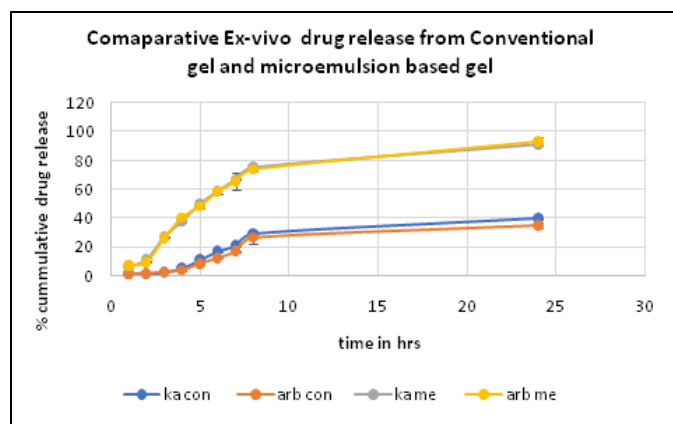


FIG. 5B: COMPARATIVE EX-VIVO RELEASE STUDIES

**In-vitro Activity Test:**

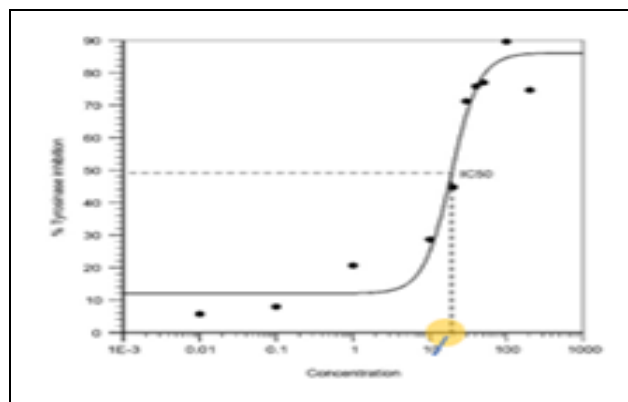
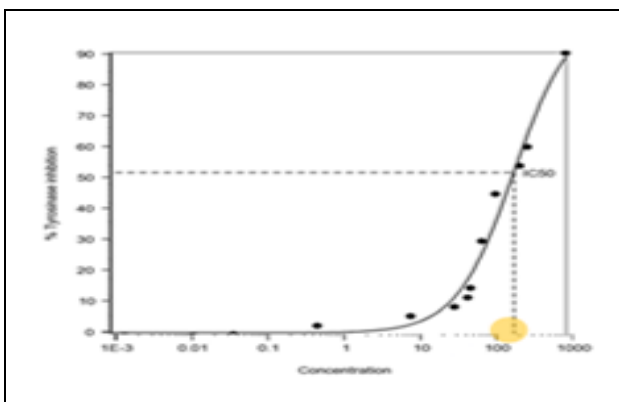
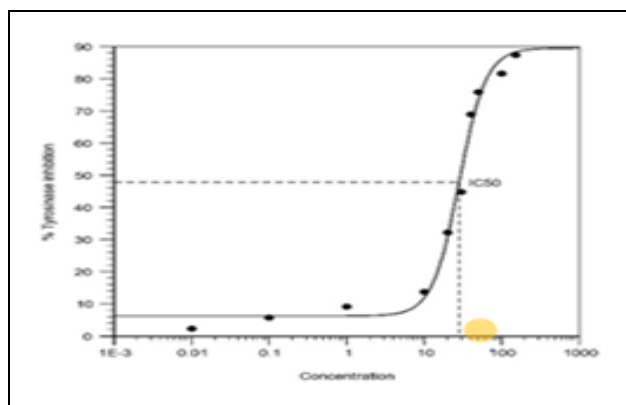
**In-vitro Tyrosinase Inhibition for Determining Synergism / Potentiation:** Tyrosinase inhibition study was performed individually for kojic acid and arbutin and also a combination of kojic acid and arbutin in the ratio of 3:7. The following  $IC_{50}$  curves depicted in Fig. 6A was obtained.  $IC_{50}$  values for kojic acid, arbutin and 3:7 mixtures of kojic acid and arbutin were 19.302  $\mu$ M, 164  $\mu$ M, and 28.33  $\mu$ M respectively, suggesting that kojic acid can potentiate the effect of arbutin, thus indicating possible synergism between the two

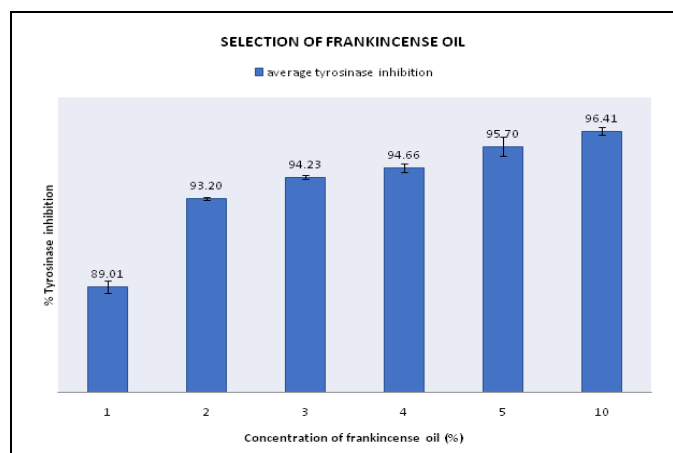
moieties. Comparative anti-tyrosinase activity exhibited by various samples is mentioned in Table 2.

Microemulsion based crème' gels showed better activity as against the conventional gel. The tyrosinase inhibitory effect of ME1 and ME2 was  $97.54 \pm 1.25\%$  and  $99.9 \pm 1.03\%$  respectively proving the additive anti-tyrosinase effect of frankincense oil. The tyrosinases inhibitory of frankincense oil at different concentrations were analyzed, and thus 5% w/w was the final chosen concentration as represented in Fig. 6B. Thus, suggesting that the use of frankincense oil in the microemulsion can enhance the ability to treat hyperpigmentation.

TABLE 2: ANTI-TYROSINASE ACTIVITY EXHIBITED BY VARIOUS SAMPLES

Sample name	% Inhibition
Kojic acid and arbutin loaded conventional gel	$94.99 \pm 0.89$
Marketed formulation	$99.22 \pm 0.15$
Kojic acid and arbutin loaded microemulsion (without frankincense oil) based gel (ME1)	$97.54 \pm 1.25$
Kojic acid and arbutin loaded microemulsion (with frankincense oil) based gel (ME2)	$99.9 \pm 1.03$

KOJIC ACID  $IC_{50}$ : 19.302  $\mu$ MARBUTIN  $IC_{50}$ : 164  $\mu$ MKOJIC ACID AND ARBUTIN IN RATIO (3:7)  $IC_{50}$ : 28.33  $\mu$ MFIG. 6A:  $IC_{50}$  CURVES OF KOJIC ACID, ARBUTIN AND DRUG BLEND



**FIG. 6B: TYROSINASE INHIBITORY EFFECT OF VARIOUS CONCENTRATION OF FRANKINCENSE OIL**

***In-vitro* Skin and Ocular Irritation Test by HET-CAM [ICCVAM-Recommended Test Method Protocol:**

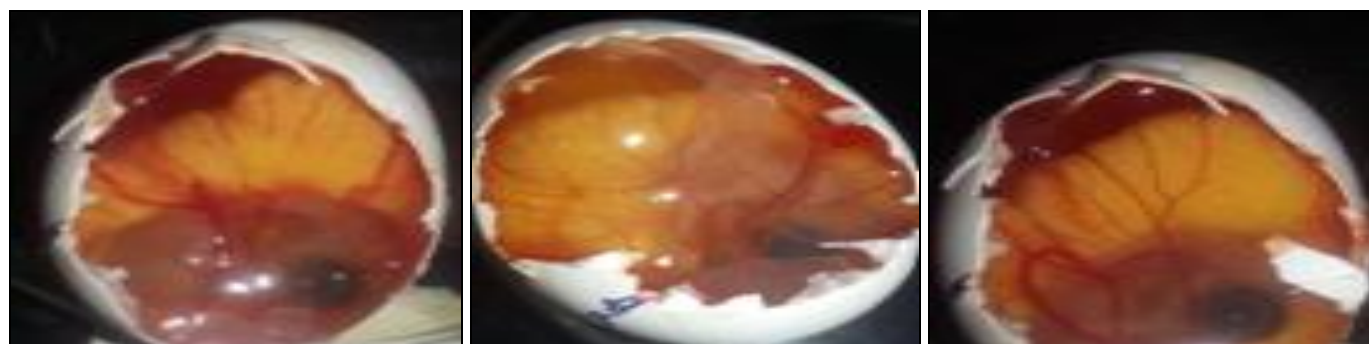
**Hen’s Egg Test - Chorioallantoic Membrane (HET-CAM) Test Method]:** As per the standard irritation scoring (IS) provided by HET CAM protocol, individual solutions, as well as blend solution of kojic acid and arbutin, presented mild irritation.

Irritancy scoring of kojic acid and arbutin loaded microemulsion based crème gel was 0, indicating that it does not have irritation potential. The images of CAM surfaces are depicted in the figure below.



**NEGATIVE CONTROL: 0.9% NaCl (NO EFFECT/0)**

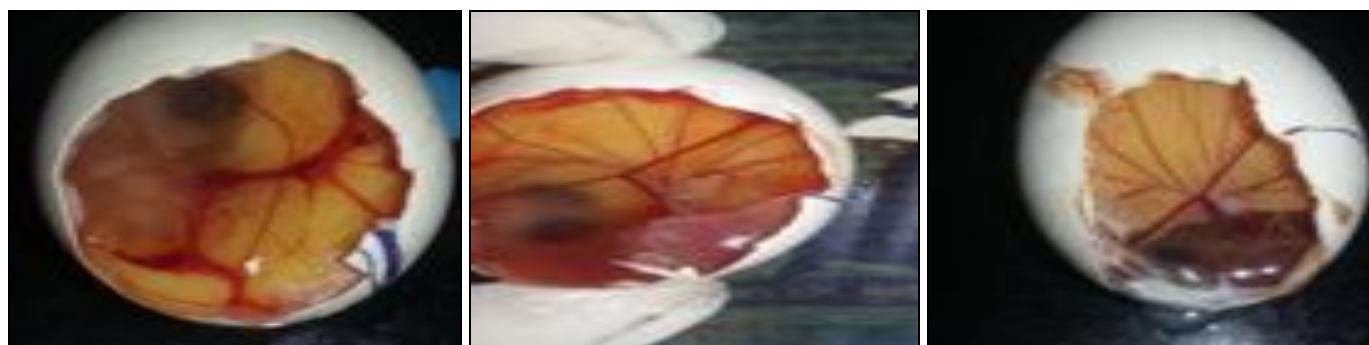
**POSITIVE CONTROL: 1% NaOH (NO EFFECT/0)**



**MARKETED FORMULATIONS (HAEMORRHAGE/7)**

**0.3 % KOJIC ACID SOLUTION (HAEMORRHAGE/5)**

**0.7 % ARBUTIN SOLUTION (LYSIS/3)**



**1% KOJIC ACID AND ARBUTIN SOLUTION (HAEMORRHAGE/5)**

**1% KOJIC ACID AND ARBUTIN LOADED MICROEMULSION ONLY (NO EFFECT/0)**


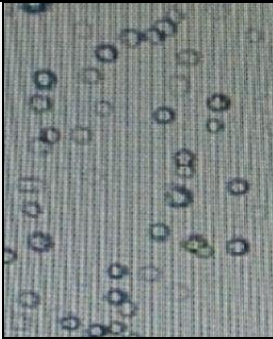
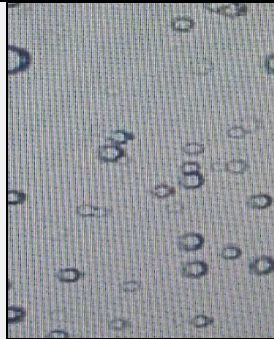
**1% KOJIC ACID AND ARBUTIN LOADED MICROEMULSION BASED GEL (NO EFFECT/0)**

**FIG. 7: CAM IMAGES OF SAMPLES**

**Formulation Stability Study:** The novel system was evaluated for its appearance, microscopical properties, content, release, and pH for a period of 3 months. The results observed are mentioned below in **Table 3**. Stability studies conducted as per ICH guidelines suggest that the optimized micro-emulsion based crème' gel was stable over a period of 3 months. Further, the microemulsion

based gel was exposed to sunlight for a period of 7 h, and the photostability of the formulation was checked. Photostability studies reports revealed that the formulation was found to be stable ever after 7-8 h of exposure to UVA/UVB irradiation (280-400 nm), thus, suggesting that the optimized micro-emulsion based crème gel can be used by patients even when exposed to sunlight.

**TABLE 3: STABILITY DATA OF MICROEMULSION BASED CRÈME' GEL AT 25 °C ± 2 °C/ 40% ± 5% RH AND PHOTO EXPOSURE PROFILE**

Parameters	1 week sample	3 month sample	Sample exposed to UVA/UVB irradiation (280–400 nm)
Motic microscopic images			
Average Particle diameter	0.18 µm	0.16 µm	0.14 µm
% content	97.11 ± 1.8 96.21 ± 1.33	95.78 ± 1.64 94.15 ± 0.12	96.26 ± 1.03 94.56 ± 0.75
Appearance	Smooth, shiny homogenous gel	Smooth, shiny homogenous gel	Smooth, shiny homogenous gel
pH	5.5 ± 1.35	5.6 ± 0.56	5.5 ± 1.28
Drug release by <i>in-vitro</i> diffusion study within 8 h	72.33 ± 1.56 / 82.44 ± 2.01	76.46 ± 1.22 / 84.12 ± 1.22	75.98 ± 1.22 / 81.31 ± 1.22

**CONCLUSION:** The research work proves to be a promising strategy to facilitate better penetration of the water-soluble skin lightening agents. The penetrability of the formulation allowed the agents to reach the lower layers of the skin from where the hyperpigmentation originates. Thus, the concentration of the kojic acid and arbutin in highly penetrable novel encapsulated formulation can be reduced to 0.3% and 0.7 %, respectively as against the superficially treatable low penetrable conventional/ marketed formulation containing kojic acid and arbutin at concentrations 2% and 5% w/w respectively. In this study, microemulsion based crème gel was successfully fabricated by aqueous titration method. Incorporation of kojic acid and arbutin in microemulsion provided a dramatic improvement in its photostability and penetrability as compared to conventional gel. *In-vitro* and *ex-vivo* drug diffusion studies revealed that microemulsion based crème' gel showed sustained release and sufficient skin retention thus,

delivering drugs at an effective concentration as against conventional gel. *In-vitro* skin irritation by HET CAM test indicated that the microemulsion based crème' proved the safety of the formulation.

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