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FORMULATION, OPTIMIZATION AND EVALUATION OF GLIMEPIRIDE TRANSFERO-SOMAL GEL FOR ANTIDIABETIC ACTIVITY IN RATS

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Keywords:

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ABSTRACT: Transferosomes are a special liposome type consisting of phosphatidylcholine and an edge activator. Glimepiride is a second-generation sulphonylurea exerting its effects in patients with type II Diabetes. Glimepiride is hydrophobic in nature possess less solubility and permeability. The study aims to formulate, optimize and evaluate Glimepiride Transferosomal gel for antidiabetic activity in Rats. Glimepiride loaded Transferosomes were prepared using thin film hydration method and were optimized using factorial design (3²) using Designexpert® software (Version 7.0.0, Stat-Ease Inc., Minneapolis, USA); different formulations (F1-F9) were prepared. Two different independent variables were used, which include: Amount of Soya lecithin (X1), Amount of tween 80 (X2), and the responses are Entrapment Efficiency (Y1), vesicle size (Y2). The optimized formulation of Glimepiride transferosomes was incorporated into a Carbopol 934 gel base and 1.5 % of Glimepiride transferosomal gel was prepared, which was evaluated for drug content, pH, spreadability, viscosity and *in-vitro* drug release. The prepared Glimepiride transferosomes had a maximum EE% was found to be (73.25 %), particle size was found to be (323 nm), the Spreadability value was 12.88 cm. The prepared transferosomal gel was found to be 86.02%, which represents good content uniformity. The viscosity was found to 1656 ± 1.25 cps. The percentage drug release for Glimepiride transferosomal gel was found to be 89.05±0.25. R² values for the optimized formulation were highest for the Higuchi model. This indicated that the drug release from all the formulations followed diffusion controlled release mechanism. Stability studies showed that Glimepiride Transferosomal gel is more stable at 4°C when compared to room temperature.

INTRODUCTION: The skin is the vast and maximum effortlessly reachable organ of the body; it serves as a prospective path of drug direction for systemic effects. However, skin is divided into 4 layers in which the upper part of the skin, the stratum corneum, represents the most resistible blockade for the drug penetration throughout the skin, which controls the transdermal bioavailability of drugs.

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Therefore, distinctive transporters are needed to fight the natural skin barrier to administer drug molecules with discrete physicochemical effects to the systemic circulation $^{1, 2}$. Drug and vaccine transcutaneous administration is a feasible surrogate to the oral and parenteral routes of delivery.

It is viable to avoid "first-pass" deactivation with the aid of using hepatocyte, lower the chance of abdominal digestive irritation, offer steady absorption of medicine above lengthy intervals of period and decrease the density of drugging, which boom adherence ³. Because of its big exterior portion and effects that ease drug administration, the transcutaneous path has gained awareness over the ancient period. However, the skin is the most fruitful blockage to drug passage, which is critical to plan the drug administration means in the most effective way possible, it involves selecting a mechanism to pass the dosage into the skin surface or Percutaneous absorption is the immersion of particles along the skin and circulation throughout the body except lungs ⁴. For use on the skin, different types of drug delivery systems have evolved. Cutaneous pharmacotherapy mainly aims to administer drugs to the skin to induce local effects at the application area⁵.

Skin is thought to be a highly effective shield that keeps our internal in and external out. Transitioning the skin barrier property, which includes penetration of medications, synthesized materials, and biologically active substances, is the most important prerequisite and goal for transdermal delivery. As a result, numerous approaches have been tried to improve the permeation rate of different agents. One of the approaches is to implement novel transporters effective in control delivery, drug release at a preestablished pace, and selected release. This could result in increased effectiveness, protection, and patient compliance. Microparticles, nanoparticles, liposomes, elastic liposomes, niosomes, ethosomes, and other novel release systems or transporters for efficient cutanoeus and transcutanoues administration. The best delivery methods are those that are stable, nontoxic, non-immunogenic, and cost-effective. In addition, those methods must be taken up pharmaceutically, steady, biologically suited and forbearing friendly ⁶⁻⁹.

Transferosomes: A transferosome is a device that allows medications to be transferred from one person to another. Administered to the desired location via skin penetration ¹⁰⁻¹² a vesicle that is ultradeformable and has a It has an aqueous core surrounded by a complicated lipid bilayer favoured format shown in Fig. 1. The interdependency of local composition as well as the self-regulating and selfoptimizing characteristics of transferosomes are due to their interdependency of local composition. The bilayer's form they can pass through quickly. Different transportation hurdles and then function as a cost-effective mode of transportation medication carrier for therapeutic compounds' long-term release. The approaches resulted in two

new vesicular carriers ¹³: transferosomes and ethosomes, which are flexible elastic lipid-based vesicles ¹⁴. Each transferosome's inner layer is made up of an aqueous compartment surrounded by a lipid bilayer with specific features due to the insertion of "edge activators" to the vesicular membrane. Surfactants, including sodium cholate, sodium deoxycholate, Span 80, and Tween 80 are common edge activators ¹⁵.



FIG. 1: TRANSFEROSOMES STRUCTURE

Diabetes Mellitus: Diabetes mellitus is a collection of metabolic illnesses marked by chronic hyperglycemia caused by insulin production, insulin action, or both. Uncontrolled diabetes may lead to stupor, coma and if not treated death, due to ketoacidosis or rare from nonketotic hyperosmolar syndrome. Type 2 diabetes is an impairment in the way the body regulates and uses sugar (glucose) as a fuel. This long-term (chronic) condition results in too much glucose circulating in the bloodstream. Eventually, high blood glucose levels can lead to disorders of the circulatory, nervous and immune systems¹⁶.

MATERIALS AND METHODS:

Materials: Glimepiride and Soya lecithin were obtained as gift samples from Apex Laboratories limited, span 80, tween 80, sodium deoxycholate, Carbopol were purchased from S d Fine chemical Ltd, Mumbai, India. Chloroform, methanol, and triethanolamine were purchased from Otto Chemicals, Mumbai, India.

Methods:

Preparation of Glimepiride-Loaded Transferosomes using Thin Flim Hydration Method: The Glimepiride loaded transferosomes were prepared by rotary thin film hydration method ¹⁷. An accurate amount of Soya lecithin, surfactants and Glimepiride were dissolved in a chloroform and methanol mixture (in the ratio of 2:1 v/v) in a round bottom flask as seen in **Table 1**. The organic solvent mixture was removed by using a rotary film evaporator under reduced pressure at 60 °C \pm 2 °C and 60 rpm for 15 min to get a homogeneous lipid film. The flask was kept under vacuum to remove residual solvent. The thin lipid film was hydrated with Phosphate buffer pH 6.8 above the lipid transition temperature at 60 rpm for 60 min to obtain large multilamellar vesicles (MLVs). The resulting MLVs were kept overnight at 4 °C to allow the complete hydration of the vesicles. The MLVs were then subjected to probe sonication at 4 °C for 30 min using an ultrasonic sonicator (Mangaldeep tech solutions) to get small unilamellar vesicles (SUVs) and stored at 4°C for further investigation.

Formulation Design:

TABLE 1	: FORMUL	ATION DESIGN	VOF GLIM	EPIRIDE TRAI	NSFEROSOMES

Quantities in W/W % (100mg)									
Ingredients	F1	F2	F3	F4	F5	F6	F7	F8	F9
Glimepiride (mg)	4	4	4	4	4	4	4	4	4
Soya lecithin (W/V)	20	40	60	20	40	60	20	40	60
Chloroform: methanol (V/V)	2:1	2:1	2:1	2:1	2:1	2:1	2:1	2:1	2:1
Sodium deoxycholate (w/w)	5	15	25	-	-	-	-	-	-
Span 80 (V/V)	-	-	-	5	15	25	-	-	-
Tween 80 (W/W)	-	-	-	-	-	-	5	15	25
Sodium benzoate (w/w)	1	1	1	1	1	1	1	1	1
Distilled water (v/v)	Q.S								

Design of Experiment: Design Expert software was employed to execute the statistical evaluation of the experimental design. The most popular response surface method was the central composite design. A 3-level design was employed in this study, requiring 9 experiments. Optimization of formulation has a great extent of influence on the composition and development method of transferosome. The 3^2 design for optimization of the transferosome was employed to study the effect of independent variables selected as soy lecithin (X1) and tween 80 (X2) on dependent variables, *i.e.*, entrapment efficiency (EE%) and size of a vesicle (nm) and respectively. The coded levels translated to the experimental units, experimental runs, and their factor combinations considered in the present study are summarized in **Table 2A and Table 2B.** The significant model was analyzed using ANOVA.

TABLE 2A: INDEPENDENT VARIABLES

Coded values Level	Independent variables	
	X ₁ , Soy lecithin	X ₂ , Tween 80
-1	30	5
0	60	15
+1	90	25

TABLE2B:FORMULATIONOFGLIMEPRIDETRANSFEROSOMES

	Factor 1	Factor 2
Run	A:Soya lecithin	B:Tween 80
1	-1	-1
2	-1	1
3	1	0
4	1	-1
5	-1	0
6	1	1
7	0	1
8	0	0
9	0	-1

Characterization of Glimepiride Loaded Transferosomes:

Vesicle	mor	pholo	gy:	The	mor	phological
characterist	ics	of	the	prepa	ared	vesicular

formulations were examined. A drop of the vesicular formulation was homogeneously put to a clean glass slide and allowed to air dry. The sample was gold coated with a Sputter coater (JEOL, Japan) and examined under SEM with a 20kV accelerating voltage.

Entrapment Efficiency (EE): The amount of Glimepiride entrapped inside the vesicular systems was determined by ultracentrifugation technique ¹⁸. An aliquot of the vesicular suspension was placed in the centrifuge tubes and centrifuged at 40,000 rpm for 3 hr at 4 °C. The free Glimepiride present in the supernatant was isolated from the pellet, appropriately diluted, and quantified for

Glimepiride content. To assess the total Glimepiride present in the prepared vesicular formulations, the vesicular dispersion was lysed with chloroform and subjected to sonication. The samples were analyzed under UV visible spectrophotometer.

The percentage entrapment efficiency (% EE) was determined as follows ¹⁹:

$$EE \% = (T-S) / T \ge 100$$

Where,

T is the total amount of Glimepiride; S is the amount of Glimepiride present in supernatant only; T-S is the amount of Glimepiride present inside the vesicles.

Zeta Potential (ZP): The ZP of the vesicular formulations was also measured by Zetasizer nano ZS using the M3PALS technique. Before analysis, samples were appropriately diluted with milli-Q water, and measurements were done out at 25 °C by determining the electrophoretic mobility.

Determination of In-vitro Diffusion of Glimepiride Transferosomes: A diffusion study of Transferosomes formulations was carried out using a Franz diffusion cell through a dialysis membrane. The dialysis membrane was soaked in distilled water for 24 hours. Franz diffusion cells contain two compartments upper donor and the lower receptor compartment. The receptor compartment was filled with 6.8 pH, and donor compartment contained transferosome suspension on the dialysis membrane with an exposure area of 2cm² to receptor medium, and whole assembly was kept on a magnetic stirrer at 600rpm for 10 hours, and samples were withdrawn at specified time interval of 1 hr and replaced with equal volume of buffer. Samples were appropriately diluted with buffer and analyzed using UV spectrophotometer at 249nm. Steady-state Flux (Jss) was calculated from the slope of the linear part of the cumulative amount of drug permeated per unit area ($\mu g/cm^2$) against a time (h) plot.

Permeability coefficient (Kp) =Jss / Co

(Co = initial CIP concentration) 20

Preparation of Transferosomal Gel: To make the vesicular dispersions suitable for skin application,

the optimized Glimepiride -TFS formulations were incorporated into carbopol gel ²¹. Briefly, carbopol 934 was dispersed in water with continuous stirring at 500 rpm to prepare 2% w/w dispersion is formed. The dispersion was mechanically stirred and then neutralized with 0.5% v/v triethanolamine solution. The dispersion was allowed to stand overnight for complete hydration and swelling. Finally, the optimized Glimepiride-TFS were added to carbopol dispersion and mixed with gentle stirring.

Characterization of Glimepiride Loaded Transferosomal Gel:

Homogeneity: Patient compliance needs to determine the homogeneity of semisolid dosage forms applied topically on the skin. This was done by pressing a small quantity of gels (GLIM Transferosomal gel) between the thumb and the index finger. The consistency was determined as homogeneous or not.

Spreadability: The Spreadability of gel formulations was determined by measuring the spreading diameter of 1g of gel between two horizontal plates $(20 \text{ cm} \times 20 \text{ cm})^{22}$.

$$S = M \ge L / T$$

Where, S is the spreadability in g/s, M is the mass in grams L is the length of gel spread & T is the time in seconds.

pH Measurement: One gram of gel was dispersed in 20 mL of distilled water, and a digital pH meter (Systronics Digital - 335) was used to determine the pH value. The measurement was performed three times, and the mean \pm SD was calculated ²².

Viscosity Measurement: The gel viscosity was measured by a Brookfield viscometer (Brookfield DVE) using spindle number S64 rotated at a speed of 12 rpm for a 10-s run time at 37°C.

Drug Content Determination: The GLIM content was measured by placing 1 g of gel onto a clean volumetric flask (100 mL) and completing the volume with 6.8 pH buffer. This was then stirred for 2 h. The solution was filtered, and samples were analyzed spectrophotometrically at 249 nm 22 .

In-vitro **Diffusion Studies:** A diffusion study of GLIM-loaded Transferosomal gel was carried out using a Franz diffusion cell through a dialysis membrane. The dialysis membrane was soaked in distilled water for 24 hours.

The receptor compartment was filled with 6.8 pH. Donor compartment contained Glimepiride Transferosomal gel on a dialysis membrane with an exposure area of 2cm^2 to receptor medium, and the whole assembly was kept on a magnetic stirrer at 600rpm for a period of 10 hours and samples were withdrawn at a specified time interval of 1 hr and replaced with an equal volume of buffer.

Samples were appropriately diluted with buffer and analyzed using UV spectrophotometer at 249nm. Steady-state Flux (Jss) was calculated from the slope of the linear part of the cumulative amount of drug permeated per unit area (μ g/cm²) against a time (h) plot.

Permeability coefficient (Kp) = Jss / Co

(Co = initial CIP concentration.)

Release Kinetic Profile for Glimepirde Transferosomal Gel: The drug release kinetics studies were estimated to determine the type of release mechanism followed. Release kinetic study of optimized formulation was studied for different kinetic equations (zero order, first order and Higuchi & Peppas).

- Zero-order Kinetic Model: Cumulative %drug released versus time.
- First-order Kinetic Model: Log cumulative percent drug remaining versus time.
- Higuchi's Model: Cumulative percent drug released versus square root of time.
- Korsmeyer-Peppas Model: Log % CDR versus Log time.

Zero-order Kinetics: The following equation would predict Zero-order release:

$$\mathbf{A}_{\mathrm{t}} = \mathbf{A}_{\mathrm{0}} - \mathbf{K}_{\mathrm{0}}\mathbf{t}$$

Where, A_t = Drug release at time't', A_0 = Initial drug concentration. K_0 = Zero-order rate constant (hr⁻¹).

When the data is plotted as cumulative percent drug release versus time, if the plot is linear, the data obeys zero-order kinetics and its slope equals zero-order release constant K_0 .

First-Order Kinetics: The following equation could predict first-order kinetics:

$$Log C = log C_0 - K_t / 2.303$$

Where, C= amount of drug remained at time't'. C_0 = Initial amount of drug. K= First order rate constant (hr⁻¹).

The data plotted as log cumulative percent drug remaining versus time yields a straight line, indicating that the release follows first-order kinetics. The constant ' K_t ' can be obtained by multiplying 2.303 with the slope value.

Higuchi's Model: Drug release from the matrix devices by diffusion has been described by following Higuchi's classical diffusion equation:

$$Q=\left[D\epsilon / \tau \left(2A - \epsilon C_{S}\right) C_{S} t\right]^{\frac{1}{2}}$$

Where, Q= amount of drug release at time't'. D = Diffusion coefficient of the drug in the matrix. A = Total amount of drug in a unit volume of the matrix. CS = Solubility of the drug in matrix. ε = Porosity of the matrix. τ = Tortuosity. t = Time (hrs at which q amount of drug is released). Above equation can be simplified as if we assume that 'D', 'C_S' and 'A' are constant. Then equation becomes:

$$\mathbf{Q} = \mathbf{K}\mathbf{t}_{1/2}$$

When the data is spited according to the equation *i.e.*, cumulative drug release versus square root of time yields a straight line, indicating that the drug was released by diffusion mechanism. The slope is equal to 'K' (Higuchi's 1963)²³.

Korsmeyer-Peppas Model: Korsmeyer *et al.* (1983) derived a simple relationship which described drug release from a polymeric system equation

$$Mt / M\infty = Ktn$$

Where, Mt / $M\infty$ is a fraction of drug released at time t, k is the release rate constant and n is the release exponent ²⁴.

In-vivo Evaluation of Glimepiride Transferosomal Gel:

Animals: Albino rats of male 9-to-11-week age, weighing 180–250g were procured from Mahaveera enterprises, Hyderabad.

Animals were housed in standard laboratory conditions at 25°c with 12 hr light-dark cycle with free access to chow and water *ad libitum*. The research protocol was approved by Institutional Animal Ethical Committee (DSOP/IAEC/DEC2020/02).

Induction of Diabetes: All the rats were fasted overnight before the administration of streptozotocin. Diabetes was induced in rats by intraperitoneal injection of streptozotocin at 60 Milligram/Kilogram body weight, as shown in **Fig. 2 & 3**. Blood was taken from tale vein to examine blood glucose with a glucose monitoring device. The blood glucose was examined. The animals with more than 200 mg/dl of blood glucose levels were considered diabetic rats and were used for this study.



FIG. 2: ANIMAL HOLDER



FIG. 3: INDUCTION OF DIABETES USING STREPTOZOTOCIN

Excision Wound Model: Albino rats 180-250 gm were taken for studies; the rats were anesthetized prior to and during the infliction of the experimental wounds. The surgical interventions were carried out under sterile conditions using Anaesthetic Ether. A wound of 500 sq. mm on dorsal thoracic region was made.

- Rats were divided into three groups.
- Group I: (control group): No application of any medication.
- Group II: (standard group): Animals of this group were applied to topically marketed product (Megaheal gel).

• Group III: (test group): Animals of this group were applied topically 1.5 % Glimepiride Transfersomal gel.

The gel was applied by using a cotton bud. After the application of gel, the wounds were covered with transparent film dressing and then wrapped using a bandage to prevent the film dressing from slipping out of position.

The animals were observed for wound closure on 1^{st} , 4, 8, 12, 14 and 16th days using a transparency sheet and a permanent marker and for a period of epithelialization.

Stability Studies: The optimized GLIM-TFS gel was subjected to stability studies in lacquered aluminium collapsible tubes stored at three different temperatures *i.e.*, 4 ± 2 °C, 25 ± 2 °C and 40 ± 2 °C for 3 months period, and evaluated for appearance, color, pH, viscosity and drug content as described earlier ²¹.

RESULTS AND DISCUSSION: Physicochemical properties of glimepiride **Table 3 & 4**.

Color:

TABLE 3: TABLE SHOWING THE DESCRIPTION OFGLIMEPIRIDE

Test	Description
Colour	White to yellowish white

The melting point was found to be 196.0°C-208.0°C by the open capillary method.

DISCUSSION: The melting point of Glimepiride was found to be 196.0°C, which complied with standards, thus indicating the purity of obtained drug sample.

Solubility:

TABLE 4: TABLE SHOWING THE SOLUBILITY OFGLIMEPIRIDE

Solvents	Solubility (µg/ml)
Methanol	35.5 ± 0.05
Water	$52.5\pm\ 0.07$
0.1N HCl	40.5 ± 0.03
Phosphate buffer 6.8	45.5 ± 0.06

Preformulation Study:

Determination of \lambdamax of Glimepiride: A solution of Glimipiride concentration of 10 µg/ml was scanned in the wavelength 200-400 nm range. It was observed that the API showed considerable absorbance at a wavelength of 249 nm **Fig. 4 & 5**, **Table 5**. The absorption spectrum was found to be

sharp and maximum at wavelength of 249 nm, therefore, it was selected as the wavelength for detection in phosphate buffer 6.8.



FIG. 4: AMAX CURVE OF GLIMEPIRIDE

TABLE5:ABSORBANCEOFDIFFERENTCONCENTRATIONS

Concentration (µg/ml)	Absorbance
0	0.00 ± 0.00
2.5	0.145 ± 0.076
5	0.295 ± 0.012
7.5	0.467 ± 0.081
10	0.589 ± 0.105
12.5	0.751 ± 0.057
15	0.888 ± 0.089



FIG. 5: CALIBRATION CURVE OF GLIMEPIRIDE

FTIR: The identity of Glimepiride obtained was confirmed by FTIR. The FTIR spectrum Glimepiride shows a characteristic sharp peak at 3367.95 cm⁻¹ **Fig. 6** specifying the presence of N-H- (amine) group. The strong peak at 1669.68 cm⁻¹ was assigned as the S=O (sulphur) character.



FIG. 6: FTIR SPECTRA OF PURE GLIMEPIRIDE



FIG. 7: FTIR SPECTRA OF DRUG+SOYA LECITHIN



FIG. 8: FTIR SPECTRA OF GLIMEPIRIDE TRANSFEROSOMES

Inference: No chemical interaction and disappearance of peaks between drug and excipients was observed in **Fig. 7, 8.**

Differential Scanning Calorimetric Studies: DSC was used to determine the melting point of Glimepiride. The DSC thermogram of Glimepiride shows a sharp endotherm at 196.98 °C confirming the melting point (**Fig. 9**.



TRANSFEROSOMES

Characterization of Glimepiride Loaded Transferosomes:

Vesicle Shape and Morphology: The preliminary characterization of Glimepiride-TFS (prior to sonication) was done by using an optical

microscope. The optical microscopic images of glimepirde loaded vesicles are shown in **Fig. 10 & 11**, which confirm the formation of vesicles having a spherical shape and different lamellarity **Table 6**.



FIG. 10: OPTICAL PHOTOMICROGRAPH OF GLIMEPIRIDE LOADED TRANSFEROSOMES FORMULATION

TABLE 6:	MEAN	VESICLE	SIZE	OF	GLIMEPIRIDE
TRANSFEI	ROSOM	ES			

Formulation code	Size of Vesicle (nm)
F1	353
F2	392
F3	435
F4	448
F5	340
F6	412
F7	362
F8	385
F9	323

The values are expressed as mean, \pm SD (n=3)



FIG. 11: GRAPHICAL REPRESENTATION OF MEAN VESICLE SIZE

Entrapment Efficiency (EE):

TABLE 7: % ENTRAPMENT EFFICIENCY OFGLIMEPIRIDE TRANSFEROSOMES

Formulation code	Entrapment Efficiency (%)
F1	55.43
F2	61.25
F3	69.43
F4	67.56
F5	57.15
F6	62.22
F7	68.27
F8	64.35
F9	73.25

The values are expressed as mean, \pm SD (n=3)

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FIG. 12: GRAPHICAL REPRESENTATION OF % ENTRAPMENT EFFICIENCY OPTIMIZATION OF FORMULATION

Inference: The following **Fig. 12 and Table 7** show the Entrapment Efficiency of Vesicles formulation F9 has the highest value.

The central composite design (CCD) was used to find suitable variables. Total 9 experimental runs were executed, and the recorded results are represented in **Table 8**.

TABLE 8: COMPOSITION AND CHARACTERISTICS OF FORMULATIONS

111212 01 00	Easter 1 V	Easton 2 V	Degnonge 1 V	Dognongo 2 V
	Factor 1, Λ_1	Factor $2, \Lambda_2$	Response 1, 1_1	Response 2, 1 ₂
Run	A:Soy lecithin	B:Tween 80	Entrapment Efficiency (%)	Size of Vesicle (nm)
1	-1	-1	55.43	353
2	-1	1	61.25	392
3	1	0	69.43	435
4	1	-1	67.56	448
5	-1	0	57.15	340
6	1	1	62.22	412
7	0	1	68.27	362
8	0	0	64.35	385
9	0	-1	73.25	323

TABLE 9: ANOVA FOR QUADRATIC MODELRESPONSE 1: ENTRAPMENT EFFICIENCY:

Source	Sum of Squares	Df	Mean Square	F-value	p-value	
Model	276.17	5	55.23	2759.82	< 0.0001	significant
A-Soy lecithin	220.95	1	220.95	11039.73	< 0.0001	
B-Tween 80	51.39	1	51.39	2567.83	< 0.0001	
AB	0.0042	1	0.0042	0.2111	0.6772	
A ²	1.75	1	1.75	87.36	0.0026	
B ²	2.08	1	2.08	103.97	0.0020	
Residual	0.0600	3	0.0200			
Cor Total	276.23	8				

The Model F-value of 2759.82 implies the model is significant. P-values less than 0.0500 in Table 9 indicate model terms are significant. In this case A, B, A², B² are significant model terms.

Fit Statistics: The Predicted R^2 of 0.9976 is in reasonable agreement with the Adjusted R^2 of 0.9994; *i.e.* the difference is less than 0.2. Adeq

Precision measures the signal-to-noise ratio. A ratio greater than 4 is desirable. Here ratio of 155.744 indicates an adequate signal.

Std. Dev.	0.1415	R ²	0.9998
Mean	64.32	Adjusted R ²	0.9994
C.V. %	0.2199	Predicted R ²	0.9976
		Adeq Precision	155.7439

RESPONSE 2: SIZE	OF VESICLE					
Source	Sum of Squares	Df	Mean Square	F-value	p-value	
Model	14629.00	5	2925.80	585.16	0.0001	significant
A-Soy lecithin	12973.50	1	12973.50	2594.70	< 0.0001	
B-Tween 80	1536.00	1	1536.00	307.20	0.0004	
AB	9.00	1	9.00	1.80	0.2722	
A ²	60.50	1	60.50	12.10	0.0401	
B ²	50.00	1	50.00	10.00	0.0508	
Residual	15.00	3	5.00			
Cor Total	14644.00	8				

TABLE 10: ANOVA FOR QUADRATIC MODEL RESPONSE 2: SIZE OF VESICLE

The Model F-value of 585.16 implies the model is significant. P-values less than 0.0500 Table 10 indicate model terms are significant. In this case A, B, A² are significant model terms.

Fit Statistics: The Predicted R^2 of 0.9904 is in reasonable agreement with the Adjusted R^2 of 0.9973; *i.e.*, the difference is less than 0.2. Adeq Precision measures the signal-to-noise ratio. A ratio greater than 4 is desirable. Here the ratio of 68.465 indicates an adequate signal. Response analysis through polynomial equations.

Std. Dev.	2.24	R ²	0.9990
Mean	383.33	Adjusted R ²	0.9973
C.V. %	0.5833	Predicted R ²	0.9904
		Adeq Precision	68.4653

Effect of Variables on Entrapment Efficiency: Data was analyzed to fit full second-order quadratic or cubic polynomial equation(s) with added interaction terms to correlate the various studied responses with the examined variables. As depicted by 2D contour plot Fig. 14A and 3D response surface plot Fig. 13A, the percent entrapment efficiency of drug is positively correlated with X_1 , soy lecithin concentration and X_2 , tween 80 concentrations. The final mathematical model in terms of coded factors as determined by the Design-Expert software is shown below in Eq. (1) for entrapment efficiency.

Response 1: Entrapment Efficiency, Y1

$$\begin{split} Y_1 = +64.27 &+6.07 X_1 + 2.93 X_2 - 0.0325 X_1 X_2 - 0.9350 \\ X_1^2 + 1.02 X_2^2 & \dots \ \text{Eq. (1)} \end{split}$$

Effect of Variables on size of Vesicle: As depicted in 2D and 3D plots (Fig. 14B and 13B), it is indicated that at lower levels of soy lecithin, the levels of tween 80 concentration showed a negative influence on particle size. Whereas, increasing the levels of soy lecithin, at a constant tween 80 concentration, an increase in particle size was observed. Thus, the lowest level of soy lecithin and the highest level of tween 80 concentrations resulted in minimum particle size. The final mathematical model in terms of coded factors as determined by the Design Expert software is shown below in Eq. (2) for the size of vesicle.

Response 2: Size of a vesicle, Y₂

 $Y_2 = +383.00 + 46.50 X_1 - 16.00 X_2 - 1.50 X_1 X_2 + 5.50 X_1^2 - 5.00 X_2^2 \dots Eq. (2)$



FIG. 13: 3D RESPONSE SURFACE PLOTS FOR EVALUATING THE INFLUENCE OF SOY LECITHIN (X1) AND TWEEN 80 (X2) ON ENTRAPMENT EFFICIENCY (Y1) AND SIZE OF VESICLE (Y2)



FIG. 14: 2D CONTOUR PLOTS FOR EVALUATING THE INFLUENCE OF SOY LECITHIN (X1) AND TWEEN 80 (X2) ON ENTRAPMENT EFFICIENCY (Y1) AND SIZE OF VESICLE (Y2)

Selection of Optimize **Formulation:** The optimized Glimepiride loaded transferosomal formulation was chosen from the nine trial formulations prepared according to the factorial design based on the criteria of lowest vesicle size and maximum entrapment efficiency. After a thorough evaluation, it was found that glimepiride-TFS 9 (soya lecithin: 60 mg and Tween 80: 25 mg) fulfilled the requisites of an optimum formulation. The optimized Glimepiride loaded transferosomes GLIM-TFS 9 (OPT-GLIM-TFS) showed 323 nm vesicle size and 73.25 ± 0.32 % entrapment efficiency. The OPT-GLIM- TFS formulation was used for further studies.

Surface Morphology (Vesicle size) of Optimize Formulation: The surface morphology and threedimensional configurations of the optimized vesicular formulations were further observed by Fig. 15. The SEM photographs of SEM transferosomes confirmed their three dimensional (3D) structure and shown the presence of the welldefined spherical or near-spherical shaped vesicles with a smooth surface.



FIG. 15: SEM IMAGES OF GLIMEPIRIDE TRANSFEROSOMES

Zeta Potential of Optimize Formulation: The observed ZP value of optimized formulation was - 23.0 mV for GLIM-TFS Fig. 16.



FIG. 16: ZETA POTENTIAL

Determination of *In-vitro* **Diffusion of Glimepiride Transferosomes:** Diffusion studies of all formulations **in Table 11** were carried out using a dialysis membrane for 10 hours, and samples E-ISSN: 0975-8232; P-ISSN: 2320-5148

were analyzed using double beam UV spectrophotometer and were shown in Graphical representation in **Fig. 17**.

	TABLE 11:	CUMULA	TIVE %	DRUG	RELEASE
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	F1	F2	F3	F4	F5	F6	F7	F8	F9
Time	%CDR								
in hours									
0	0	0	0	0	0	0	0	0	0
1	22.84 ± 0.37	17.85 ± 0.43	24.33±0.56	24.00 ± 0.91	28.90 ± 0.35	22.31±0.76	18.13 ± 0.83	18.4 ± 0.54	22.74±0.33
2	34.28 ± 0.43	25.63±0.36	30.10 ± 0.54	26.22 ± 0.81	30.40 ± 0.67	23.12±0.45	35.92±0.76	29.49 ± 0.54	30.21±0.84
3	46.16±0.22	29.66±0.32	33.97±0.53	38.33 ± 0.65	33.20±0.74	35.65 ± 0.43	44±0.83	36.49 ± 0.32	37.38 ± 0.66
4	49.34±0.91	34.69 ± 0.47	38.83 ± 0.45	47.81±0.76	39.21±0.37	40.00 ± 0.68	45.73±0.65	48.53±0.83	49.24±0.61
5	54.64 ± 0.22	40.38±0.12	42.81±0.17	53.34 ± 0.43	40.24 ± 0.66	43.217±0.47	48.06±0.63	50.24 ± 0.76	52.020.32
6	56.25 ± 0.33	45.22 ± 0.78	48.54 ± 0.34	57.31±0.86	43.81±0.57	46.71±0.74	54.23 ± 0.35	52.54 ± 0.56	55.22 ± 0.38
7	59.66±0.37	54.88 ± 0.15	51.22±0.16	62.31±0.47	42.93±0.54	63.43±0.23	57.57±0.65	58.54±0.19	61.37±0.34
8	60.25 ± 0.58	57.65 ± 0.43	64.82 ± 0.86	65.32 ± 0.35	54.42 ± 0.65	66.42 ± 0.76	59.60±0.23	66.24 ± 0.10	69.57±0.18
9	63.54 ± 0.48	62.19 ± 0.48	66.22 ± 0.25	72.40±0.23	60.31±0.44	74.91±0.56	68.11±0.49	70.56 ± 0.54	73.44±0.27
10	74.66 ± 0.45	69.45±0.55	72.15±0.33	75.9 ± 0.27	61.24±0.78	76.64±0.56	70.60±0.51	73.66±0.11	86.05±0.12

The values are expressed as mean, \pm SD (n=3).



FIG. 17: GRAPHICAL REPRESENTATION OF CUMULATIVE % DRUG RELEASE

Formulation of Glimepiride Transfersomal Gel: Glimepiride Transferosomal gel was prepared using 1.5 % Carbopol 934 as a gelling agent. The concentration of Glimepiride in the prepared Transferosomal gel was 1.5% w/w.

Evaluations of Glimepiride Transferosomal Gel: Glimepiride Transferosomal gel is smooth with a

In-vitro Drug Release:

homogenous appearance. The Spreadability value is 12.88 ± 0.50 cm, which indicates that they can be spared easily on the skin surface with little stress. The pH value is 6.4 ± 0.34 , which is considered within the normal pH range for topical preparations. The drug content of the glimepiride Transferosomal gel was found to be $86.02\pm0.21\%$, representing good content uniformity. The viscosity of GLIM Transferosomal gel is found to be 1656 ± 1.25 cps **Table 12**.

TABLE12:EVALUATIONOFGLIMEPIRIDETRANSFEROSOMAL GEL

Evaluation	Results
Homogeneity	Good
Spreadability (Gm.cm/sec)	12.88±0.50
pH measurements	6.4±0.34
Drug content (% w/w)	86.02±0.21
Viscosity (cps)	1656±1.25

ТА	RLF	E 13	• %	CDR	ISS	AND	KP	OF	1.5%	OF	GLIM	EPIR	IDE	TRA	NSFE	ROSON	ЛАТ.	GEL.
1 1	DLI	213	. /0	UDIN,	100	AND	111	OF	1.0 /0	UT.	GLIM		ID L'	INA		lobor	IAL	GLL



FIG. 18: IN-VITRO DRUG RELEASE OF 1.5 % GLIMEPIRIDE TRANSFEROSOMAL GEL

Release Kinetic Profile for Glim Transferosomal Gel: The drug release kinetics studies were estimated to determine the type of release mechanism followed shown in **Table 14 and Fig.** **19, 20, 21, 22.** Release kinetic study of GLIM Transferosomal gel of optimized formulation was studied for different kinetic equations (zero order, first order, Higuchi and Peppas equation)

	Zero	First	Higuchi	Peppas
	% CDR Vs T	Log % Remain Vs T	%CDR Vs √T	Log C Vs Log T
Slope	7.091727273	-0.12191267	25.13428286	1.204854203
Intercept	12.92681818	2.160525354	-2.95309629	0.835777221
Correlation	0.974705137	-0.74576377	0.993980618	0.769989248
R 2	0.939950104	0.983363613	0.987997469	0.592883442



DISCUSSION: Glimepiride transferosomal gel formulation shows Higuchi kinetics that indicate the order of drug release of drug i.e, square root of time (Higuchi) model indicates that drug release is by diffusion. R^2 values for the optimized formulation were found to be highest for the zero-order and Higuchi models. This indicated that the drug release from all the formulations followed diffusion controlled release mechanism.

In-vivo Anti-Diabetic Studies:

Serum Glucose Level (mg/dl) Estimation in Rats: The blood glucose concentration were determined for each group by taking blood from the tail vein as per the following schedule; (0th, 1st, 3rd, 7th, 12th days), after the a fore mentioned treatment was given. The blood glucose concentration was determined using the glucometer and the comparisons between the various groups were made.

Pharmacological Studies: Wound Healing Activity:

Excision Wound Model: Male Albino rats 180-250 gm were taken for studies; the rats were anesthetized before and during infliction of the experimental wounds. The surgical interventions were carried out under sterile conditions using Anaesthetic Ether. A wound of 500 sq. mm on dorsal thoracic region was made. Animals were closely observed in Table 17 for any infection, and those who showed signs of infection were separated and excluded from the study and replaced. The animals were observed for wound closure at 0, 4th, 6th, 8th, 12th and 16thday shown in Fig. 23, 24, 25, and for a period of epithelialization 25-28.

Measurement of Wound Area: The progressive changes in the wound area were monitored by a 2^{nd} , 4^{th} , 6^{th} , 8^{th} , 12^{th} , and 16^{th} day.

The wound size was also measured using a daily scale, and the wound area was calculated. Wound contraction was calculated as a percentage of the reduction in wound area. (Nayak BS *et. al* 2009).

 $\begin{array}{l} \mbox{Percentage of wound contraction} = (\mbox{Initial wound area} - \mbox{Specific day wound area}) \ / \ \mbox{Initial wound area} \times 100 \end{array}$

Statistical Analysis: Data analysis was performed using the statistical package for social version 17.0 (SPSS) computer software. Descriptive statistics were adopted to display data in the mean of \pm SEM ANOVA was used to compare the mean value obtained between the different groups' Table 15, Fig. 23.

TABLE 15: SERUM	GLUCOSE LEVEL	(MG/DL) IN RATS
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Group	Serum glucose level (mg/dl)						
	0 day	1 st day	3 rd day	7 th day	10 th day	12 th day	
Control	88.76±1.21	91.06±1.01	90.46±1.11	92.76±1.21	93.16±1.41	92.11±1.21	
STZ	167.6±2.11	264.6±3.11	298.11±4.51	312.1±5.61	324.1±5.21	300.1±5.71	
Transferosomes	169.3±3.41	259.76±4.21	289.1±5.21*	283.3±6.21**	171.5±3.21**	151.1±5.21***	

Differences were considered significant whenever the P value are reported as mean \pm SEM. ***p<0.001, **p<0.01 and *p<0.05.



Effect of Glimepiride Transferosomal Gel in Diabetic Wound:

	TABLE 16: % WOUND CON	FRACTION WITH GLIME	PIRIDE TRANSFEROSOMAL GEL
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Tre	atment	% Wound Contraction with Transferosomal gel						
	1 st day	4th day	8th day	12th day	14th day	16th day	18th day	20 th day
Control	2.279±	$2.146 \pm$	$1.928 \pm$	$1.581\pm$	$1.211 \pm$	$1.006 \pm$	$0.819\pm$	$0.510\pm$
	0.839 00.00	1.185	1.675	1.499	1.700	0.5158	0.11	0.214
		8.85	19.83	29.67	43.46	62.81	71.66	82.61
Transfer	2.259 ± 1.43	$1.927 \pm$	$1.114 \pm$	$0.708 \pm$	0.397±	-	-	-
Osomal	00.00	1.507	1.054	1.354	0.6750			
Gel		24.69	42.61	88.61	97.22			
Megahel	$2.249 \pm$	1.991±	$1.547 \pm$	$1.043 \pm$	$0.720\pm$	0.231±	-	-
-	1.198	1.516	1.384	0.8695	0.3782	0.8883		
	00.00	23.26	59.65	81.38	87.64	97.69		

Differences were considered significant whenever the P value are reported as mean \pm SEM. ***p<0.001, **p<0.01 and *p<0.05.



FIG. 24: EFFECT OF GLIMEPIRIDE TRANSFEROSOMAL GEL ON %WOUND CONTRACTION

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FIG. 25: EFFECT OF GLIMEPIRIDE TRANSFEROSOMAL GEL ON %WOUND CONTRACTION

TABLE 17: WOUND HEALING OF RATS

Days		Groups	
	Group I	Group II	Group III
0 Day		a de la dela	
4 th Day			
8 th Day			
12 th Day			
16 th Day		and the second sec	

Stability Studies of 1.5 % Glimepiride Transferosomal Gel:

	TABLE 18: STABILITY OF 1.5 % GLIMEPIRIDE TRANSFEROSOMAL	GEL AT 4±2°C AND ROOM TEMPERATURE
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Duration	4±2°C	Room Temperature				
	DC	%CDR	DC	%CDR		
Initial	85.5±1.25	81.6±0.42	86.5±1.41	78.01±0.42		
30 days	86.89±1.01	80.45±1.5	84.5±0.54	80.72±0.19		
60 days	86.02±0.21	82.32±0.21	81.1 ± 0.84	77.57±0.64		
90 days	84.53±1.22	$81.44{\pm}1.41$	79.9±0.16	76.21±1.51		

The values are expressed as mean, \pm SD (n=3).

Stability studies showed that Glimepiride transferosomal gel is more stable at 4°C compared to other temperatures, as seen in **Table 18**.

CONCLUSION: In the present study, Glimepiride entrapped transferosomes were successfully prepared and optimized using 3^2 factorial designs. The optimized formulations of GLIM-TFS showed a unilamellar vesicular structure with lower vesicle size (323nm) and greater entrapment efficiency (73.25%). The FTIR studies revealed no chemical interaction between drugs and excipients. The invitro drug release studies of transferosomes were 86.75% of the optimized formulation F9. The optimized formulation of **GLIM-TFS** was successfully incorporated into carbopol gel and characterized for pH, viscosity, drug content, spreadability.

The % drug release of transferosomal gel was found to be 89.05±0.25, and the kinetic study reveals that drug release follows the Higuchi model, as the R^2 value is highest, which depicts that the release of the drug is by diffusion. It can be concluded that transferosomes can serve as a potential therapeutic strategy for type II diabetes treatment by delivering the Glimepiride into the deepest layers of skin. Further investigation of GLIM-TFS gels in the treatment of type II Diabetes by using suitable animal models was conducted to confirm the efficiency of these formulations. The applicability of these developed vesicular gels for the treatment of diabetes control is being explored. Finally, Stability studies showed that Glimepiride Transferosomal gel is more stable at 4° C compared to room temperature.

Scope:

- Stability studies can be further extended as per ICH Guidelines.
- IVIVC correlation has to be done.
- Further study in Inflammatory cytokines, serum insulin.
- Study on immune histochemistry of pancreas and liver.

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